

1 **Footprints and conditions of multistep alkali enrichment in basaltic**
2 **melts at Piton de la Fournaise (La Réunion Island, Indian Ocean)**

3 **G. BOUDOIRE**^{1,2,3*}, A. DI MURO^{4,5}, L. MICHON^{4,6}, N. METRICH⁴

4 ¹Université Clermont Auvergne, CNRS, IRD, OPGC, Laboratoire Magmas et Volcans, F-
5 63000 Clermont-Ferrand, France

6 ²GZG, Abt. Geochemie, Universität Göttingen, Goldschmidtstr. 1, 37077 Göttingen, Germany

7 ³Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Palermo, Italy

8 ⁴Université de Paris, Institut de physique du globe de Paris, CNRS, F-75005 Paris, France

9 ⁵Observatoire volcanologique du Piton de la Fournaise, Institut de physique du globe de Paris,
10 F-97418 La Plaine des Cafres, France

11 ⁶Université de La Réunion, Laboratoire GéoSciences Réunion, F-97744 Saint Denis, France

12 * Corresponding author. Present address : Université Clermont Auvergne, Laboratoire Magmas
13 et Volcans, 6 avenue Blaise Pascal, 63178 Aubière, France. Telephone: +33 (0)7 85221911. E-
14 mail: guillaume.boudoire@uca.fr

15 **ABSTRACT**

16 Deciphering magma evolution below ocean basaltic volcanoes is all the more challenging
17 because magma mixing is a common process tending to modify the pristine geochemical
18 diversity during magma ascent. On the western flank of the Piton de la Fournaise volcano,
19 transitional basalts have compositions that testify to origins down to the upper mantle and
20 display a widespread geochemical diversity ranging from a tholeiitic affinity to an alkaline one.

21 There, we show that evolved melt inclusions and matrix glasses (MgO < 9 wt%) record
22 an alkali enrichment coupled with a Ca/Al ratio decrease, which tracks the effect of
23 clinopyroxene crystallisation at the depth of the mantle-crust underplating layer. At this depth
24 and shallower, reverse zoning of olivine crystals, clinopyroxene dissolution, and hybrid melt
25 compositions testify of extensive mixing processes leading to a homogenization of the pristine
26 geochemical footprint of melts upon ascent. Enrichment in incompatible trace elements in some
27 evolved melt inclusions suggests that magma ponding at the depth of the mantle-crust
28 underplating layer favours also assimilation of melts originating from low degrees of partial
29 melting of cumulates (wehrlites, dunites).

30 Conversely the most primitive melt inclusions documented so far on La Réunion (MgO
31 up to 11.2 wt%) better preserve a pristine geochemical variability related to partial melting of
32 a slightly heterogeneous mantle source. We suggest that these slightly distinct source
33 components may mirror the compositions of recent melts from the two closely located Piton de
34 la Fournaise and Piton des Neiges volcanoes.

35

36 **INTRODUCTION**

37 Ocean basaltic islands are among the largest volcanoes on Earth. Their magmas commonly span
38 large geochemical spectra (Dasgupta et al. 2009) predominantly mirroring pre-eruptive magma
39 evolution in either deep (mantle) or shallow (crustal) reservoirs. Actually, magma ascent often
40 proceeds by progressive steps of transfer between multiple magma ponding zones (Hildner et
41 al. 2011; Klügel et al. 2015) that leads to the emission of variably degassed and apparently
42 homogeneous evolved melts (Jennings et al. 2017). On the contrary, fast magma ascent in a few
43 days/weeks from the deep part of the plumbing system up to the surface is less common,
44 meanwhile favours the emission of gas-rich primitive melts at the surface, as at Kilauea
45 (Hawaii; Helz et al. 2014; Poland et al. 2014; Rae et al. 2016).

46 At Piton de la Fournaise (PdF), historical eruptive activity, which mainly takes place
47 inside the Enclos Fouqué caldera (central area), is characterized by weak intereruptive
48 outgassing and by the emission of homogenous evolved and degassed transitional basalts. These
49 “steady state basalts” have a chemical composition intermediate between tholeiitic and alkaline
50 basalt and ascend from crustal reservoirs (Upton and Wadsworth, 1966; Albarède et al. 1997;
51 Boivin and Bachèlery, 2009; Di Muro et al. 2014, 2015, 2016; Gurioli et al. 2018). Conversely,
52 the NW rift zone (peripheral area: NWRZ), linking the Enclos Fouqué caldera with the
53 quiescent neighbouring Piton des Neiges (PdN) volcano, has been recently found to be the locus
54 of (i) a large emission of lavas with compositional and petrological variability (Valer 2016;
55 Boudoire et al. 2019), (ii) significant soil CO₂ gas emissions (Liuzzo et al. 2015; Boudoire et
56 al. 2017), and (iii) deep seismicity down to mantle level (Michon et al. 2015; Boudoire et al.
57 2017). The scoria cones within this rift zone emit transitional basalts with either a tholeiitic or
58 an alkaline affinity, defined on the basis of the alkalinity index ($AI = [K_2O + Na_2O - 0.37 \times$
59 $(SiO_2 - 39)]$; Rhodes, 1996; Bureau et al. 1998a; Lénat et al. 2012; Walther et al. 2015; Valer,
60 2016; Boudoire et al. 2019). As a whole, the geochemical variability of basalts emitted within
61 the NWRZ has been mainly attributed to deep fractionation of clinopyroxene (Albarède et al.
62 1997; Brugier 2016; Boudoire et al. 2019), minor source effects (Valer et al. 2017; Boudoire et

63 al. 2019), and contamination by small amounts of low-degree partial melts derived from
64 cumulates (Boudoire et al. 2019).

65 However, the geochemical footprint of each process and the conditions favouring such
66 melt variability remain still poorly constrained. Indeed, previous studies were mainly based on
67 bulk rocks and experimental petrology that makes it challenging to decipher melt evolution
68 during magma ascent. For instance, if the role of deep fractionation of clinopyroxene in the
69 evolution of PdF melts has been extensively inferred (Albarède et al. 1997; Brugier 2016;
70 Boudoire et al. 2019), barometric and physico-chemical conditions controlling clinopyroxene
71 stability need further constraints. This is even more true considering that: (i) no clinopyroxene
72 phenocrysts were described in basalts with an alkaline affinity, and (ii) the rare magnesian melt
73 inclusions previously analysed do not present an alkali enrichment (Bureau et al. 1998a, 1999;
74 Brugier 2016; Boudoire et al. 2019). Likewise, slight source effects were inferred by slight
75 variations of some canonical incompatible trace element ratios (Nb/U; Valer et al. 2017) and
76 Sr-Nd isotopes (Boudoire et al. 2019), but their origin and their influence on the full
77 geochemical melt variability are not well established.

78 Here, we aim at identifying the geochemical characteristics of Piton de la Fournaise
79 melts, determining the origin of their geochemical variability, and more broadly, improving our
80 understanding of magma evolution beneath ocean basaltic islands. To address these questions,
81 we combined detailed geochemical study of melt inclusions and crystal zoning from a set of
82 transitional basalts emitted along the NWRZ, whose bulk rocks have an alkaline affinity and
83 show limited crystal recycling. Our results are compared with those previously obtained from
84 transitional basalts emitted in the same peripheral area and recording a tholeiitic affinity
85 (Bureau et al. 1998b; Valer et al. 2017), and “steady state basalts” emitted in the central area
86 since 1998 (Bureau et al. 1999; Vigouroux et al. 2009; Villemant et al. 2009; Di Muro et al.
87 2014). Our study fully complements previous works focused on the geochemical variability of
88 lavas emitted at Piton de la Fournaise and aims at deciphering both the early steps of deep
89 magma evolution and the footprints of magmatic processes controlling melt evolution in the
90 plumbing system.

91

92 **GEOLOGICAL SETTINGS**

93 La Réunion Island (Indian Ocean) is the youngest intraplate basaltic island related to the hotspot
94 that formed the Deccan flood province in India, 65 Myr ago (Mahoney et al. 2002). The 7 km
95 high volcanic edifice is built on the Indian oceanic crust with a Moho at 10.2-13 km below sea

96 level (Gallard et al. 1999; Fontaine et al. 2015). Two volcanic edifices form the island (Fig. 1).
97 Piton des Neiges (3.1 km a.s.l.) forms the north-western part of the island and is the oldest and
98 largest edifice. Its subaerial activity started at least 2.2 Myr ago, and it has now been dormant
99 for 29 kyr (Quidelleur et al. 2010; Salvany et al. 2012). Piton de la Fournaise (2.6 km a.s.l.)
100 has been built on the south-eastern flank of Piton des Neiges starting at least 0.53 Myr ago. It
101 is one of the most active basaltic shield volcanoes on Earth (Roult et al. 2012). Both volcanoes
102 emit transitional magmas produced by partial melting of a common mantle source (Fisk et al.
103 1988; Hanyu et al. 2001; Valer et al. 2017). However, some authors have reported minor
104 variations of incompatible trace and alkali elements (Fisk et al. 1988), and of Pb-Hf-Nd-Sr
105 isotopes, which were attributed to either (i) the impingement of two small-scale blobs of plume
106 material at the base of the lithosphere of La Réunion, one leading to the Piton des Neiges
107 activity and the other to the Piton de la Fournaise activity (Bosch et al. 2008), or (ii) a two-step
108 mixing involving three mantle source components (Nauret et al. 2019).

109 At Piton de la Fournaise, nearly 95 % of the eruptions since the 18th century took place
110 inside the central Enclos Fouqué caldera, while less than 5 % of the eruptions occurred along
111 the rift zones radially cross-cutting the volcanic edifice (Fig. 1; Villeneuve and Bachèlery 2006;
112 Michon et al. 2015). Central eruptions inside the Enclos Fouqué caldera and peripheral eruption
113 along the NE and SE rift zones (NERZ and SERZ) are closely linked to the central shallow
114 plumbing system and to the seawards displacement of the steep eastern flank (Michon et al.
115 2015; Peltier et al. 2015). This activity classically leads to the emission of strongly degassed
116 transitional basalts displaying a very homogeneous evolved composition ($\text{MgO} = 6.5 \pm 0.8$ wt%
117 and $\text{K}_2\text{O} + \text{Na}_2\text{O} = 3.5 \pm 0.2$ wt% on average in glasses; Villemant et al. 2009; Di Muro et al.
118 2014, 2015, 2016). Conversely, the NW rift zone, on which we focus here, connects the western
119 flank of Piton de la Fournaise with the older Piton des Neiges edifice. It has been proposed that
120 the deepest part of the Piton de la Fournaise plumbing system is offset beneath the western
121 flank of the volcano, and that magma bodies settled beneath the NWRZ may feed the Piton de
122 la Fournaise reservoirs and their eruptions (Michon et al. 2015; Liuzzo et al. 2015; Boudoire et
123 al. 2017, 2019). The transitional basalts emitted within the NWRZ are generally more primitive
124 and alkaline than the central ones ($\text{MgO} = 7.3 \pm 0.8$ wt% and $\text{K}_2\text{O} + \text{Na}_2\text{O} = 4.0 \pm 0.3$ wt% on
125 average in glasses; Brugier 2016; Boudoire et al. 2019), show a bi-modal distribution with
126 tholeiitic and alkaline affinity (Brugier 2016; Valer et al. 2017), and are coeval with the central
127 activity (Boudoire et al. 2019).

128

129 **SAMPLING STRATEGY**

130 The few studies of basalts emitted along the NWRZ (peripheral area) of Piton de la Fournaise
131 have been mostly based on (1) transitional lavas carrying an abundant amount of olivine crystals
132 having a polymodal distribution (in size and composition) testifying to crystal recycling and
133 accumulation (Bureau et al. 1998a, b; Welsch et al. 2013; Valer et al. 2017; Boudoire et al.
134 2019), and (2) aphyric transitional basalts with a tholeiitic affinity (Bureau et al. 1998a, b). In
135 this study, we excluded lavas and pyroclasts carrying abundant and potentially recycled crystals
136 in order to focus only on crystal-poor fast-quenched olivine basalts. Selected basalts were
137 emitted by NWRZ vents extensively studied for the geochemistry of bulk rocks (Fig. 1; Walther
138 et al. 2015; Valer 2016; Boudoire et al. 2019).

139 In order to cover the whole known geochemical diversity of basalts emitted from NWRZ
140 vents, we selected basalts on the basis of their bulk rock Mg-number ($Mg\# = [Mg^{2+} / (Mg^{2+} +$
141 $Fe^{2+})]$) by assuming the proportion of Fe_2O_3 is 18% in Piton de la Fournaise melts, on average;
142 Pichavant et al. 2016) and alkalinity index (AI; see equation above from Rhodes (1996)). These
143 geochemical markers allow us to discriminate between primitive and more evolved basalts, and
144 transitional basalts with either an alkaline (AI>0) or a tholeiitic affinity (AI<0). We have
145 selected fast-quenched unaltered pyroclasts (lapilli, scoriae, bombs) to minimize the effects of
146 post-trapping processes that may have affected melt inclusions composition. Such an approach
147 allows us to discriminate (Fig.1; Table 1):

148 (1) primitive basalts ($Mg\# > 0.65$) with either an alkaline (AI = 0.4 ± 0.1 ; Trouis Blancs:
149 TB) or a tholeiitic affinity (AI = -0.3; Piton de Caille: PdC). Trouis Blancs (>9 kyrs) glassy
150 lapilli layer is here investigated also for crystal composition and zoning, fluid inclusions and
151 melt inclusions composition, and barometry. Piton de Caille (<5 kyrs) basalts have been already
152 analysed for crystal composition, fluid inclusions and melt inclusions composition, and
153 barometry (Bureau et al. 1998b). We have completed here the PdC dataset with a study of
154 crystal zoning.

155 (2) More evolved transitional basalts ($Mg\# = 0.60-0.62$; Below Petit Cratère: BPC;
156 Plaine des Sables fall: PdS; Piton Haüy: HA, Petit Cratère: PC) (Fig. 1). We studied the melt
157 inclusions in all these products. Below Petit Cratère (<5 kyrs) glassy lapilli were also analysed
158 for crystal composition and zoning, fluid inclusions composition, and barometry. A study of
159 crystal zoning was also performed in basalts from Petit Cratère (470 yrs BP). Plaine des Sables
160 (380 yrs BP) and Piton Haüy (<5 kyrs) glassy lapilli were previously analysed for crystal
161 compositions and fluid inclusions (Boudoire et al. 2019), and we have completed the dataset
162 with a study of melt inclusions composition.

163 Our analyses from basalts emitted from peripheral vents in the NWRZ were compared
164 with published data on basalts from the 1998-2010 period emitted in the central area between
165 two long periods of rest (1992-1998 and 2010-2014; Bureau et al. 1999; Vigouroux et al. 2009;
166 Villemant et al. 2009; Di Muro et al. 2014). A particular attention was paid to basalts emitted
167 at the Hudson central vent (H) during the 1998 central eruption, the only one characterized by
168 a positive AI during this period and associated with deep seismicity and recording high
169 pressures of crystallization down to the mantle depth (Bureau et al. 1999; Di Muro et al. 2016;
170 Boudoire et al. 2019).

171 Lava and melt inclusions compositions are then correlated with the available bulk rock
172 composition of intrusive rocks like dunites (D) and wehrlites (W) found as enclaves in tephra
173 of La Plaine des Sables (Upton et al. 2000; Peters et al. 2016), and of gabbroic blocks (G) found
174 scattered in the ashes of the Bellecombe explosive event (Upton et al. 2000), whose main
175 characteristics are reported in Table 1.

176

177 **ANALYTICAL METHODS**

178 **Mineralogy and crystal zoning**

179 Zoning profiles in clinopyroxenes were acquired with a JEOL JXA-8900RL electron
180 microprobe using 15 kV potential, 15 nA defocused beam coupled with a 5-15 s counting time
181 per element (University of Göttingen, Germany). Relative errors are 0.5-1.5 % (SiO₂, MgO,
182 FeO, CaO, Al₂O₃), 1-4 % (TiO₂), 5-15 % (Na₂O, Cr₂O₃, NiO), 5-35 % (K₂O, MnO). Electron
183 microprobe analyses on olivine were performed using a CAMECA SXFive at Camparis (Paris,
184 France) with 15 kV, 20 nA focused beam and a counting time varying from 10 to 50 s depending
185 on the element (Appendix 1). Relative analytical error measured is 0.8-4 % for concentrations
186 >15 wt%, 6-18 % for concentrations in the 1-15 wt% range, and 15-100% for concentrations in
187 the 1-0.2 wt% range. Absolute error on forsterite content (Fo) in olivine is <0.8 %.

188

189 **Fluid inclusions and shrinkage bubbles**

190 Fluid inclusions and shrinkage bubbles in melt inclusions were analysed at room temperature
191 at IPGP (Paris, France) using a Horiba T64000 Raman spectrometer manufactured by Jobin-
192 Yvon, offering an excellent spectral resolution (<1 cm⁻¹) for peak identification (counting time:
193 2x180s). Fluid inclusions (Appendix 2) were excited using an argon ion laser at 488 nm

194 wavelength with an average power of 1.13W. The stabilized laser power on the sample was not
195 measured. Based on former acquisitions, we consider that it was in the range 10-100 mW and
196 most probably lower than 100 mW because no surrounding glass was burnt during the analyses.
197 Laser heating of fluid inclusions during analysis (>60 s per fluid inclusion) should lead to a
198 constant and stable temperature increase in fluid inclusions in the range 4-100 °C above room
199 temperature by considering the homogeneity of host crystal radius (around 1 mm), thickness
200 (<400 μm), and mineralogy (olivine only in our case) (Hagiwara et al. 2021). In our study, the
201 temperature of the fluid inclusions is estimated in the range 33-49 °C (during analyses) based
202 on the relation linking the temperature to the intensity of CO₂ hot bands and of the Fermi dyad
203 bands (Hagiwara et al. 2021; Appendix 2). Such a range of temperature allowed to exceed the
204 critical temperature of CO₂ (31 °C) during the analysis. Consistently, no moving CO₂ vapor
205 bubble was detected in fluid inclusions immediately after the analysis arguing for the presence
206 of a homogenous CO₂ phase during the analysis. As reported by Boudoire et al. (2018) during
207 the Raman investigation of CO₂-rich fluid inclusions in other samples from Piton de la
208 Fournaise, no other gaseous species were detected in both fluid inclusions and shrinkage
209 bubbles in melt inclusions over an extended spectral window of 200 – 4000 cm⁻¹.

210 Pressure estimates were obtained using the CO₂ density deduced from the Fermi dyad
211 split (Δ ; van den Kerkhof and Olsen, 1990; Frezzotti et al. 2012) between the 1388 cm⁻¹ and
212 1285 cm⁻¹ bands, using the equation of Wang et al. (2011) calibrated for Jobin-Yvon Horiba
213 spectrometer at 1800 grooves/mm gratings with a 532 nm wavelength. This instrumental
214 configuration is similar to that used in our study with the exception of the laser wavelength.
215 However, laser wavelength plays only a minor role in the equations (maximum difference of
216 0.01 g.cm⁻³ for CO₂ density calculations with 514 and 532 nm wavelengths), whereas gratings
217 and spectral resolution have a larger influence (up to 0.1 g.cm⁻³ between a 600 and a 1800
218 grooves/mm gratings) (Lamadrid et al. 2017). Raman spectra have been acquired on a single
219 window collection according to the recommendations of Lamadrid et al. (2017). The total
220 uncertainty (ϵ (TOT)) on CO₂ density related to the use of the CO₂ densimeter of Wang et al.
221 (2011) and to the analysis of fluid inclusions is:

$$\epsilon \text{ (TOT)} = \sqrt{\epsilon(\text{analytical})^2 + \epsilon(\text{temperature})^2 + \epsilon(\text{processing})^2}$$

224 Where ϵ (analytical) is the uncertainty related to the use of the CO₂ densimeter of Wang et al.
225 (2011), ϵ (temperature) is the uncertainty related to the effect on the Fermi dyad of the heating
226 of fluid inclusions by the laser (Hagiwara et al. 2021), and ϵ (processing) is the uncertainty

227 related to spectral resolution by peak fitting ($<0.014 \text{ g.cm}^{-3}$; Kobayashi et al. 2012). By
228 comparing the CO_2 densimeters of Wang et al. (2011) with those obtained with an instrumental
229 configuration close to that used in our study (Song et al. 2009; Fall et al. 2011; Lamadrid et al.
230 2017) the maximum difference between CO_2 densities ($\varepsilon(\text{analytical})$) is 0.019 g.cm^{-3} for Δ in
231 the range $103.7 - 104.45 \text{ cm}^{-1}$ (covering the range of fluid inclusions) and 0.031 g.cm^{-3} for Δ in
232 the range $102.9 - 103.6 \text{ cm}^{-1}$ (covering the range of shrinkage bubbles). This analytical
233 uncertainty overlaps also the standard deviation (0.011 g.cm^{-3}) on the equation of Wang et al.
234 (2011). The temperature-dependency of the Fermi dyad bands between 33 and $49 \text{ }^\circ\text{C}$ leads to a
235 maximum $\varepsilon(\text{temperature})$ of 0.018 g.cm^{-3} for Δ in the range $103.7 - 104.45 \text{ cm}^{-1}$ (covering the
236 range of fluid inclusions) and 0.012 g.cm^{-3} for Δ in the range $102.9 - 103.6 \text{ cm}^{-1}$ (covering the
237 range of shrinkage bubbles). Finally, $\varepsilon(\text{TOT})$ in this study is equal to 0.029 g.cm^{-3} for fluid
238 inclusions and 0.036 g.cm^{-3} for shrinkage bubble.

239 Rare sulphate and (hydrated) carbonates were sometimes detected in a few fluid
240 inclusions. A similar observation was made by Boudoire et al. (2018, 2019) who highlight that
241 the initial presence of a $\text{CO}_2\text{-H}_2\text{O}$ mixture has to be taken into account by considering $10 \text{ mol}\%$
242 of H_2O in the initial exsolved phase that leads to a density increase not exceeding 0.03 g.cm^{-3}
243 (equation from Hansteen and Klügel (2008)). Corresponding pressures were estimated with the
244 ISOC code provided by Bakker (2003) based on the equation of state (EOS) of Duan et al.
245 (1992, 1996) for: (i) $\text{CO}_2\text{-H}_2\text{O}$ mixtures ($90 \text{ mol}\%$ of CO_2 , $10 \text{ mol}\%$ of H_2O and $0 \text{ mol}\%$ for
246 the other species), (ii) fluid inclusions olivine(forsterite)-host, and (iii) a temperature of 1250
247 $^\circ\text{C}$ (intermediate temperature between that of melt inclusions and matrix glasses; Table 3). Any
248 uncertainty on melt temperature results only in minor errors on pressure ($<50 \text{ MPa}$ for a shift
249 of 100°C).

250

251 **Melt inclusions and glasses**

252 **Major, trace, and volatile elements**

253 Major elements in melt inclusions were measured with the CAMECA SX5 electron microprobe
254 at Camparis (Paris, France) using 15 kV potential, 10 nA defocused beam coupled with a
255 counting time varying from 10 to 30 s depending on the element (Appendix 3). Analytical error
256 measured is $<1.7 \%$ for concentrations higher than $15 \text{ wt}\%$, $3\text{-}10 \%$ for concentrations in the 2-
257 $15 \text{ wt}\%$ range, and up than 13% for concentrations $<2 \text{ wt}\%$. Measurements were reiterated
258 when possible (between 2 and 7 measurements) in order to minimize the uncertainty (Appendix

259 3). Concentrations of volatiles in glasses (S, Cl, and P) were determined with a 30 nA defocused
260 beam and a longer counting time (200 s) on peak (Appendix 3). Their accuracy was checked
261 using international reference glasses (VG2: S=1458 ppm and Cl=310 ppm, and ALV98R23 :
262 S=1110 ppm). Our measurements are 1423 ± 11 ppm of S and 281 ± 12 ppm of Cl for VG2, and
263 1076 ± 47 ppm of S for ALV98R23. Detection limits are 46 ppm for S and 47 ppm for Cl.

264 Trace elements in melt inclusions, glassy matrices, and minerals were determined by
265 laser ablation ((LA)-ICP-MS) using a 193 ArF excimer laser system (Resonetics M50) with a
266 7500 cs Agilent ICP-MS (LMV Clermont-Ferrand, France) with helium as carrier gas for
267 analyses (Appendix 3). Laser repetition of 2-3 Hz and a 26-44 μm -diameter spot were used.
268 The background was measured for 30s before ablation and acquisition. Calibration was made
269 on a NIST 612 glass standard using CaO as the internal reference element. Reproducibility and
270 accuracy are better than 10% for most elements with respects to the BCR2-G standard (Jenner
271 and O'Neill 2012).

272

273 **Dissolved H₂O**

274 Dissolved H₂O in glasses was measured by Raman spectroscopy with a 514.5 nm wavelength
275 argon ion laser using a Labram HR800 spectrometer (ENS Lyon, France) equipped with a
276 Peltier-cooled CCD detector (Appendix 3). Analyses were performed by focusing a 1 μm -wide
277 laser beam (average power on the sample of 7.4 mW) through an Olympus microscope in
278 confocal setting with the highest magnification (x100). Spectra were collected in the 200-1500
279 cm^{-1} (counting time of 2x30 s) and 2800-4000 cm^{-1} (counting time of 2x50 s) domains
280 corresponding to the aluminosilicate network + CO₂ domains and the OH+H₂O domain,
281 respectively. Selected analytical conditions minimize scatter from the embedding medium, and
282 prevent sample heating and oxidation. Calibration was performed on a set of basaltic reference
283 glasses (H₂O_T Etna range: 0.4-3.04 wt%; Mercier et al. 2009, 2010; Di Muro et al. 2014). Those
284 glasses were regularly analysed during a same session in order to check and correct any
285 potential instrumental drift. Water estimation was obtained by external calibration on peak
286 intensities (H₂O_T band at 3580 cm^{-1}), cross-checked with estimations from the internal
287 calibration (silicate network band at 980 cm^{-1}) in order to correct potential errors related to
288 defocusing (see Mercier et al. (2009) for a review of the procedure). By reiterating analysis of
289 the basaltic reference glasses, the accuracy of the analysis was better than 0.05 wt% H₂O, on
290 average. In this study, reiteration of the measurements in a same melt inclusion led also to a

291 standard deviation of 0.05 wt% H₂O on the average of the measurements that we consider as
292 representative of the analytical uncertainty in this study.

293

294 **Dissolved CO₂**

295 Due to its low solubility in melts, quantifying the initial CO₂ in melt inclusions is challenging
296 (Venugopal et al. 2020). Recent studies have highlighted the role of the shrinkage bubble in the
297 underestimation of the CO₂ initially dissolved in the melt (Wallace et al. 2015; Aster et al. 2016;
298 Tuohy et al. 2016; Venugopal et al. 2020) and promoted the combination of several analytical
299 methods to recalculate it (Moore et al. 2015). In this study, we make an attempt to estimate the
300 initial content of CO₂ dissolved in melt inclusion by adding the CO₂ exsolved in the shrinkage
301 bubble ([CO₂]_{sh}) with that dissolved in the glass ([CO₂]_{glass}).

302 In the absence of carbonates in shrinkage bubbles (no peaks in the 1040-1100 cm⁻¹
303 Raman spectra; Frezzotti et al. 2012), the CO₂ content in shrinkage bubbles ([CO₂]_{sh}) was
304 calculated using the formula of Hartley et al. (2014):

$$305 \quad [\text{CO}_2]_{\text{sh}} \text{ (ppm)} = 10^6 \times \frac{d_{\text{CO}_2} \times R_{\text{sh}}}{d_{\text{glass}} \times R_{\text{glass}}}$$

306 Where d_{CO_2} is the CO₂ density (g.cm⁻³) of the shrinkage bubble estimated by the use of the
307 Fermi dyad (see the methodology for fluid inclusions above) with an uncertainty of 0.036 g.cm⁻³
308 and d_{glass} is the glass wet density (g.cm⁻³) calculated from the equation of Lange and
309 Carmichael (1987) based on the composition of melt inclusion (ranging from 2.63 to 2.72 g.cm⁻³
310 in this study). $R_{\text{sh}}/R_{\text{glass}}$ is the ratio between the volume of the shrinkage bubble and the glass,
311 respectively. The petrographical estimation of this ratio assuming an ellipsoid and a sphere as
312 representative of the melt inclusion and of the shrinkage bubble, respectively, may lead to a
313 significant uncertainty due to the potentially irregular shape of melt inclusion ($\pm 10\%$; Hartley
314 et al. 2014; Hanyu et al. 2020). To overcome this difficulty, we used the computational method
315 of Riker (2005) that predicts the pre-eruption shrinkage bubble volume fraction as a function
316 of the difference between trapping temperature and pre-eruption temperature. Both
317 temperatures were estimated from the composition of the glass in the melt inclusion and of the
318 embayment by using the thermometer of Helz and Thornber (1987), recalibrated by Putirka
319 (2008). Predicted fraction of the shrinkage bubbles is lower than 1.9 % in this study.
320 Petrographical control performed on two melt inclusions has revealed that the absolute
321 difference between measured and estimated volume fraction of the shrinkage bubble is lower
322 than 0.2 % of the total volume in this study. This value is considered as representative of the
323 uncertainty on volume fraction in this study. Only melt inclusion with a single shrinkage bubble

324 and an apparent bubble volume fraction lower than 5% were investigated to avoid the analysis
325 of pre-eruptive vapor bubbles trapped during crystal growth (Hanyu et al. 2020).

326 The residual CO₂ dissolved in the glass ([CO₂]_{glass}) was estimated by the use of the
327 thermodynamically-based model for the solubility of H₂O-CO₂ in silicate melt of Duan (2014)
328 that is valid over a wide range of temperature, pressures and melt compositions. Estimations of
329 the residual dissolved CO₂ content with this model was performed by considering (i) the
330 dissolved H₂O content determined by Raman spectroscopy, and (ii) the pressure estimation
331 from the CO₂ density of the shrinkage bubble (see details above). The variability on CO₂ content
332 by taking into account the uncertainties on H₂O measurements and pressure estimation does not
333 exceed 200 ppm of CO₂. Duan (2014) reports an additional uncertainty of 9.7% on the
334 estimation of the CO₂ estimation in his model.

335 Results (with all the uncertainties described above) and details are provided in Table 3,
336 together with some FTIR measurements performed in glassy melt inclusions.

337

338 **Correction from post-entrapment processes**

339 Melt inclusions were corrected for post-entrapment crystallisation (PEC) on the basis of
340 olivine-melt equilibrium ($K_D=0.306$; Fisk et al. 1988) and of clinopyroxene-melt equilibrium
341 ($K_D=0.27$; Putirka 2008). These equilibrium values are in accordance with: (i) recent and
342 various experimental results obtained on Piton de la Fournaise melts (Brugier 2016), and (ii)
343 conclusions from the modelling of predicted K_D as a function of the pressure, melt composition
344 and water content through the AlphaMELTS code (Boudoire 2017). At Piton de la Fournaise,
345 Fe²⁺/Fe³⁺ in spinel ranges from 1.31 to 2.19 (1.83 in average). The correlation between forsterite
346 content in olivine and the Fe²⁺/Fe³⁺ of hosted spinel (Bureau et al. 1998b; Boudoire 2017) was
347 used to estimate the Fe²⁺/Fe³⁺ in basaltic melts (Maurel and Maurel 1982).

348 PEC values (see Appendix 3 for a full description) range from 0.1 to 5.8 % in melt
349 inclusions from evolved peripheral basalts with an alkaline affinity (BPC, CHI, HA) to 6.3-12.4
350 % in those from primitive peripheral basalts with an alkaline affinity (TB). A few primary melt
351 inclusions trapped in clinopyroxene crystals from Trouis Blancs lavas and to a minor extent
352 some primary melt inclusion in olivine crystals from La Plaine des Sables fall display a negative
353 PEC percentage (down to -21%). These negative PEC percentage may reflect post-trapping
354 dissolution of the walls of the host crystals (see Discussion). Melt inclusions from evolved
355 peripheral basalts with a tholeiitic affinity (PC) undergo weak PEC in olivine (2.6-7.2 %) but
356 record large PEC in clinopyroxenes (16.0-22.1 %). In order to select PEC threshold able to limit

357 the influence of PEC on recalculated melt inclusions composition, we filtered melt inclusions
358 based on the Mahalanobis distance method through a homemade Python code. This method
359 allows recognizing outliers in multivariate datasets (De Maesschalk et al. 2000). A greater
360 dispersion in major elements (CaO, SiO₂, MgO, FeO_T) contents (i.e. a higher frequency of
361 outliers for each major element) is evidenced for PEC threshold >7.5 %, on average. For
362 instance, Boudoire (2017) reports a greater variability of the FeO_T content in melt inclusions
363 for PEC values >7.5 % ($2\sigma = 3.4$ wt%) than for PEC values <7.5 wt% ($2\sigma = 2.2$ wt%).
364 Consequently, we retained only melt inclusions corrected from a PEC ranging from 0 to 7.5 %
365 (31 melt inclusions for 51 analyses performed) in order to minimize any bias related to the post-
366 trapping processes.

367 Following the recommendations of Boudoire (2017) for the investigation of melt
368 inclusions at Piton de la Fournaise, Fe-loss correction (Danyushevski et al. 2000; Gaetani and
369 Watson 2000; Davis et al. 2003) was not applied in order to not introduce a bias due to the
370 natural variability of FeO_T content in melt inclusion potentially linked to various extents of
371 early clinopyroxene crystallisation. Fe-loss effect was only inferred for some melt inclusions in
372 crystals from lavas erupted in the central area that are not the scope of this study. On the
373 contrary, melt inclusions in crystals from tephra erupted in the peripheral area (like those
374 investigated in this study) do not show evidence of Fe-loss effect as testified by: (i) the absence
375 of systematic S-loss related to the formation of sulphur globules (Danyushevski et al. 2000),
376 (ii) the absence of MgO-enrichment, and (iii) the position of the melt inclusions close to the
377 bulk rock-matrix glass line (see Boudoire (2017) for a complete discussion). As potential Fe-
378 loss is expected to significantly affect FeO_T and MgO content only (Tuohy et al. 2016;
379 Boudoire 2017), this effect is even more limited in our study by excluding melt inclusions
380 whose PEC is higher than 7.5% that have the largest variability in composition.

381

382 **RESULTS**

383 **Crystal zoning**

384 **Olivine**

385 The composition of olivine phenocrysts in the primitive basalts with an alkaline affinity (TB)
386 from NWRZ peripheral vents ranges from Fo_{80.1} to Fo_{87.7} (Fig. 2; Appendix 4). A main mode is
387 identified at Fo_{86.5-87}, close to the equilibrium with the bulk rock (Fig. 2) and similar to that

388 reported in primitive products with a tholeiitic affinity (PdC) from the same area (Fo₈₅₋₈₇ ;
389 Bureau et al. 1998b).

390 In more evolved basalts (BPC) from the NWRZ, olivine composition is more iron-rich
391 and ranges from Fo_{81.8} to Fo_{85.9}, with a main mode at Fo_{84.5-85} at the equilibrium with the hosting
392 melt (Fig. 2; Appendix 4). This mode is consistent with that at the equilibrium with bulk rocks
393 in other evolved basalts with an alkaline affinity (PdS, HA, PC; Fo₈₃₋₈₅) in the same area (Fig.
394 2; Boudoire et al. 2019).

395

396 In TB and PdC primitive basalts from the NWRZ area, the most magnesian olivine
397 phenocrysts (i.e. Fo_{>85}) are unzoned (10 profiles; Fig. 3a, b). Only a few rare crystals with iron-
398 rich cores were identified (Fo_{<84}). These inherited crystals show a progressive reverse zoning
399 towards Fo_{85-85.5} rims (2 profiles; Fig. 3c).

400 In more evolved basalts from the NWRZ, the same pattern is observed: olivine crystals
401 whose core composition is at the equilibrium with the hosting melt (Fo₈₃₋₈₅ in BPC) are unzoned
402 (3 profiles). Conversely, olivine crystals with iron-rich cores (Fo₈₁₋₈₃) are reversely zoned (4
403 profiles; Fig. 3d). For these inherited crystals, forsterite content is constant on the first 100-200
404 μm of the core and then progressively increases towards the rim (Fo₈₄₋₈₅).

405 In order to avoid the analysis of melt inclusions in inherited crystals retrieved during
406 magma ascent, only unzoned homogeneously sized (>0.5 mm) olivine phenocrysts at the
407 equilibrium with the hosting melt or even more magnesian were analysed for melt inclusions
408 (Fig. 2).

409

410 **Clinopyroxene**

411 In the NWRZ area, clinopyroxene phenocrysts are abundant in primitive basalts with
412 alkaline affinity (TB; Appendix 5). Their average composition (En₅₀Wo₄₀FS₁₀) falls in the range
413 of augitic composition previously reported in other peripheral clinopyroxene-bearing lavas
414 (Appendix 5; Boudoire et al. 2019). Using a Kd = 0.27 for Fe-Mg partition between melt and
415 clinopyroxene (Putirka 2008; Brugier 2016), we found that these clinopyroxenes are slightly
416 more evolved (Mg# = 84) than expected from the composition of their host bulk rock (Mg# =
417 87).

418

419 In the more evolved basalts from NWRZ vents (PC; Appendix 5), augite phenocrysts have
420 an average composition (En₄₉Wo₃₉FS₁₂) similar to that analysed in primitive basalts with an

421 alkaline affinity (TB). In these more evolved basalts, augite phenocrysts ($Mg\# = 0.85 \pm 0.02$) are
422 in equilibrium with the hosting bulk rock ($Mg\# = 0.85$).

423 In primitive basalts from peripheral vents along the NWRZ (TB), the compositions of
424 clinopyroxene cores and rims are indistinguishable. Conversely, clinopyroxenes phenocrysts in
425 more evolved basalts from NWRZ vents (PC) present both sector-zoning in cores (crystal
426 growth effect; Kouchi et al. 1983; Mollo et al. 2010; Hammer et al. 2016), and oscillatory
427 zoning in rims (Fig. 4; Fig. 5a, b). High resolution (2-5 μm) elemental profiles from crystal core
428 to rim (section cut across the c-axis; Fig. 5) reveal a progressive Mg-depletion anticorrelated
429 with Cr-Ti-enrichment in cores, followed by various steps of Mg-Cr-enrichment and Ti-
430 depletion toward the rims. The Mg-Cr-rich bands are often preceded by resorption surface (dark
431 arrows on Fig. 5). Fe contents and Ti/Al ratios are almost flat in most parts of the profiles (core
432 and mantles), and increase close to the rim, independently of the oscillatory zoning (Fig. 5a).

433

434 **Fluid Inclusions**

435 Measurements (46) of CO_2 fluid inclusions were performed in olivine phenocrysts from TB and
436 BPC basalts to complete the dataset available on basalts from peripheral vents along the NWRZ
437 (Table 2; Bureau et al. 1998b; Boudoire et al. 2018, 2019). Even rare, isolated and primary (Fig.
438 6a) CO_2 fluid inclusions are predominant in olivine phenocrysts of our samples. Secondary CO_2
439 fluid inclusions are only ubiquitous in rare olivine crystals not at the equilibrium with the
440 hosting melt (Fig. 6b).

441 The density of the mixture CO_2+H_2O (90+10%) varies from 0.06 to 0.79 $g.cm^{-3}$ without
442 systematic difference between primary and secondary fluid inclusions. The corresponding
443 pressures, which correspond to minimum trapping pressures (Hansteen and Klügel 2008), range
444 between 3.3 and 4.4 kbar for fluid inclusions in olivine crystals from primitive basalt with an
445 alkaline affinity (TB; Table 2). Interestingly, they overlap with the pressure domain reported
446 from primitive basalts with tholeiitic affinity (3.1-5.7 kbar for PdC; Bureau et al. 1998b). The
447 $P_{CO_2+H_2O}$, determined between 0.4 and 4.6 kbar for the more evolved BPC basalts, are in full
448 agreement with those measured in other basalts with either a tholeiitic or an alkaline affinity
449 emitted in the peripheral area along the NWRZ (Boudoire et al. 2019).

450

451 **Melt Inclusions**

452 Melt inclusions (MIs) described in this study have been analysed far from cracks and do not
453 present evidence of post-trapping crystallisation as oxides and daughter minerals. Their longest

454 axis varies from $>200\ \mu\text{m}$ to $<50\ \mu\text{m}$) independently of the hosting olivine composition. They
455 present a well-rounded shape, are glassy, and often host a single shrinkage bubble (Fig. 6c).
456 Because up to 80-90 % CO_2 can diffuse into this bubble, the inferred initial CO_2 in melts can
457 be highly underestimated (Moore et al. 2015; Aster et al. 2016). Hence, we tried to retrieve the
458 CO_2 amount initially dissolved in the melt by measuring the P_{CO_2} into the bubble and the
459 dissolved H_2O content (Table 3). The few melt inclusions hosting several bubbles and
460 potentially trapping a deep exsolved fluid phase were not taken into consideration for the
461 recalculation (Fig. 6c). Trail of fluid inclusions tracking volatile leakage from melt inclusions
462 by decrepitation are sometimes present but rarely optically resolvable (Fig. 6d).

463

464 Major elements

465 The most magnesian melt inclusions ($\text{MgO} = 10.8\text{-}11.2\ \text{wt}\%$; $\text{CaO}/\text{Al}_2\text{O}_3 = 0.82\text{-}0.83$)
466 described at Piton de la Fournaise, are hosted in olivine $\text{Fo}_{>85}$ of TB basalts (Fig. 7a; Table 4).
467 Most of them (the inclusion $\text{Fo}_{87.1}$ on Fig. 7b) have a positive alkalinity index (AI up to 0.7),
468 which is correlated with high contents in alkali elements ($\text{K}_2\text{O} + \text{Na}_2\text{O} = 3.4\text{-}3.7\ \text{wt}\%$), in P_2O_5
469 ($0.30\text{-}0.34\ \text{wt}\%$), in TiO_2 ($2.5\text{-}2.8\ \text{wt}\%$), and a low content in SiO_2 ($<48\ \text{wt}\%$; Fig. 7c). A few
470 melt inclusions (the inclusion $\text{Fo}_{87.7}$ in Fig. 7b) have a tholeiitic affinity (AI down to -1.4),
471 which is correlated with lower content in alkalis ($\text{K}_2\text{O} + \text{Na}_2\text{O} = 3.0\text{-}3.1\ \text{wt}\%$), in P_2O_5 (0.18-
472 $0.28\ \text{wt}\%$), in TiO_2 ($2.4\text{-}2.5\ \text{wt}\%$), and higher content in SiO_2 ($49\text{-}51\ \text{wt}\%$; Fig. 7c). The
473 composition of this second group is fully comparable with that previously described in other
474 high magnesian olivine crystals from lavas having a tholeiitic affinity (PdC: Bureau et al.
475 1998b; H: Bureau et al. 1999).

476 Melt inclusions in $\text{Fo}_{<85}$ in BPC, PdS, HA basalts display lower MgO contents ($7.5\text{-}10.4$
477 $\text{wt}\%$) and $\text{CaO}/\text{Al}_2\text{O}_3$ ratio ($0.71\text{-}0.87$). Their contents in alkali elements ($\text{K}_2\text{O} + \text{Na}_2\text{O} = 3.0\text{-}$
478 $3.9\ \text{wt}\%$), in P_2O_5 ($0.26\text{-}0.40\ \text{wt}\%$), in TiO_2 ($2.3\text{-}3.0\ \text{wt}\%$), and in SiO_2 ($47\text{-}49\ \text{wt}\%$) span the
479 full range of compositions between the two poles observed in the most magnesian melt
480 inclusions. Interestingly, similar variability is observed for melt inclusions previously analysed
481 in olivine crystals from basalts emitted in the central area (Fig. 7).

482 In our dataset, the most evolved melts correspond to (1) melt inclusions in $\text{Fo}_{<82}$ olivine
483 crystals and $\text{En}_{47}\text{Wo}_{39}\text{Fs}_{14}$ clinopyroxene ($\text{MgO} = 5.7\text{-}6.5\ \text{wt}\%$; $\text{CaO}/\text{Al}_2\text{O}_3 = 0.66\text{-}0.96$) from
484 PC basalts, and to (2) most of matrix glasses in all samples ($\text{MgO} = 4.4\text{-}7.7\ \text{wt}\%$; $\text{CaO}/\text{Al}_2\text{O}_3$
485 $= 0.60\text{-}0.79$).

486

487 **Trace elements**

488 The composition of high magnesian melt inclusions (TB) covers a relatively large range of
489 compatible (Ni = 57-180 ppm; Co = 34-50 ppm; Fig. 8a, b, c), highly incompatible (Th = 1.4-
490 2.5 ppm; La = 13-21 ppm; U = 0.39-0.60 ppm; Fig. 8d, f) and alkali (Ba = 94-202 ppm; Rb =
491 12-26 ppm; Fig. 8e) trace elements. Melt inclusions with a well-marked alkaline affinity (
492 inclusion Fo_{87.1}, hereafter called “Pole 1”) are characterized by high contents in alkali (Ba, Rb)
493 and highly incompatible (Th, La, U, Pb, Nb) trace elements (Appendix 3). Surprisingly,
494 contents in highly compatible trace elements (Ni, Co, Sc) do not decrease (and in some cases
495 increase) with the increasing contents in highly incompatible trace elements like Th (Fig. 8a, b,
496 c).

497 The composition of more evolved melt inclusions (BPC, PdS, HA) covers an even larger
498 range of trace elements contents that overlaps previous published data (Fig. 8; Di Muro et al.
499 2014; Valer 2016; Valer et al. 2017). Among more evolved melt inclusions, we address here
500 the peculiar composition of two melt inclusions labeled Fo_{84.3} and Fo_{84.4} in Fig. 8. Together
501 with the inclusion Fo_{87.1} (Pole 1) from TB, they mostly bracket the trace element geochemical
502 diversity described in this study. The first one (inclusion Fo_{84.3}, hereafter called “Pole 2”)
503 displays a similar content in compatible trace elements (Ni = 146 ppm; Co = 45 ppm) than Pole
504 1 (inclusion Fo_{87.1}) but is more depleted in both alkali (Ba = 94 ppm; Rb = 12 ppm) and
505 incompatible (Th = 1.6 ppm; La = 15 ppm; U = 0.44 ppm) trace elements. The second one
506 (inclusion Fo_{84.4}, hereafter called “Pole 3”) displays a similar content in highly incompatible
507 trace elements (Th = 2.6 ppm; La = 22 ppm; U = 0.72 ppm) than Pole 1 but, is more enriched
508 in compatible trace elements (Ni = 353 ppm; Co = 59 ppm) and depleted in alkali trace elements
509 (Ba = 140 ppm; Rb = 18 ppm).

510 As a matter of fact, the trace element composition of most melt inclusions lies between
511 that of these three peculiar melt inclusions. For instance, ratios of variably incompatible trace
512 elements such as La/Yb and Th/Yb, mostly range between 8.6 ± 1.4 and 11.4 ± 1.9 , and between
513 0.9 ± 0.1 and 1.4 ± 0.2 based on Pole 2 and Pole 1, respectively. Some ratios of highly
514 incompatible trace elements (Ba/Th, Rb/Th, Nb/U, Ce/Pb) also show significant variations
515 (Appendix 3). For instance, Ba/Th ranges between 53 ± 7 and 83 ± 13 in Pole 3 and Pole 1,
516 respectively (Fig. 8e). On the contrary, other ratios, as La/Th and Th/U, do not show significant
517 variations ($= 8.5 \pm 1$ and 3.6 ± 0.5 , respectively), with the exception of the matrix glasses, which
518 are variably depleted in U independently on their composition in major elements (Fig. 7, Fig.
519 8f).

520

521 **Volatiles**

522 The composition of high magnesian melt inclusions (TB, PdC) covers also a relatively wide
523 range of volatiles contents ($\text{H}_2\text{O} = 0.4\text{-}1.4$ wt%; $\text{S} = 585\text{-}3709$ ppm; $\text{Cl} = 210\text{-}418$ ppm). This
524 range is similar to that previously defined by Bureau et al. (1998b) in melt inclusions in high
525 magnesian olivine crystals from Hudson basalts (Fig. 9). Melt inclusions with a well-marked
526 alkaline affinity (Pole 1) show high ratios of both highly and variably incompatible trace
527 elements (Fig. 8), and also high Cl content ($\text{Cl} = 338$ ppm; Fig. 9a). On the opposite, those with
528 a tholeiitic affinity (Pole 2 and PdC), having low ratios of highly and variably incompatible
529 trace elements, display low Cl content ($\text{Cl} = 184$ ppm). Unfortunately, no data are available on
530 dissolved volatiles and major elements for Pole 3, defined only on the basis of trace elements.
531 However, melt inclusions with a composition in trace elements close to Pole 3 range
532 intermediate between Pole 1 and Pole 2 in both Cl content and alkalinity. No noticeable
533 systematic difference is observed regarding other volatiles such as H_2O (1.2 ± 0.2 wt%) and S
534 (1797 ± 91 ppm) between these endmembers (Fig. 9b, c). A large variability of S (from 1706 to
535 585 ppm) and H_2O contents (from 1.4 to 0.4 wt%) can occur (TB) but without noticeable
536 variations in K_2O (0.9-0.8 wt%) and FeO_T (10.6-11.6 wt%) contents or PEC (4-6 %) and
537 independently of the size of the melt inclusions.

538 The compositions of more evolved melt inclusions in olivine crystals from magmas
539 emitted in both NWRZ (peripheral) and central areas broadly range between the compositions
540 of the two endmembers Pole 1 and Pole 2 described above (Fig. 9a). An overall decrease in Cl,
541 H_2O , and S contents is observed from evolved melt inclusions in olivine crystals from lavas
542 from NWRZ vents ($\text{Cl} = 306\pm 77$ ppm; $\text{H}_2\text{O} = 1.1\pm 0.2$ wt%; $\text{S} = 1556\pm 321$ ppm) to those from
543 central vents ($\text{Cl} = 269\pm 50$ ppm; $\text{H}_2\text{O} = 0.8\pm 0.2$ wt%; $\text{S} = 1032\pm 237$ ppm) and matrix glasses
544 ($\text{Cl} = 237\pm 29$ ppm; $\text{H}_2\text{O} = 0.2\pm 0.1$ wt%; $\text{S} = 288\pm 309$ ppm). We estimate that the extent of Cl-
545 degassing from matrix melts does not exceed 23% with respect to the pre-eruptive
546 concentrations recorded by melt inclusions (black arrows on Fig. 9). Early sulphide saturation
547 buffers the maximum pre-eruptive S concentration in the mafic melts from eccentric vents (Fig.
548 9c).

549 Dissolved CO_2 content in melt inclusions is highly variable for a same product (Table
550 3). For high magnesian melt inclusions (TB), recalculated CO_2 contents spans a very large range
551 from 13 ± 156 ppm to 3029 ± 564 ppm overlapping previous estimates made at Piton de la
552 Fournaise without taking into account the CO_2 content trapped by the shrinkage bubble (PdC,

553 775-1765 ppm; Hudson, up to 4317 ppm; Bureau et al. 1998b, 1999). For more evolved melt
554 inclusions, recalculated CO₂ contents also covers a large range from 283±259 to 1518±152
555 ppm.

556

557 **DISCUSSION**

558 **New insights on magma ponding and degassing at Piton de la Fournaise**

559 Both pressures from CO₂-rich fluid inclusions or recalculated CO₂-H₂O saturation pressures in
560 melt inclusions (Duan 2014) are undistinguishable between melts emitted along the NWRZ
561 having either an alkaline or a tholeiitic affinity (Table 2, 4).

562 The pressure distribution recorded by fluid inclusions brings the evidence of two main
563 barometric modes, and points out two levels of entrapment or reequilibration, and multi-stage
564 magma ascent (Figs. 10a, 11; Hansteen and Klügel 2008). The first and shallowest mode, at 0.3
565 kbar, is mostly recorded by fluid inclusions in olivine crystals from central lavas and highlights
566 a well-constrained storage zone located close to sea level below the summit cone (Gallard et al.
567 1999; Peltier et al. 2009; Lénat et al. 2012; Di Muro et al. 2014; Fontaine et al. 2015; Lengliné
568 et al. 2016). The second and main mode at 4.2 kbar is represented by the majority of fluid
569 inclusions investigated in olivine crystals from eccentric magmas emitted along the NWRZ and
570 corresponds to the depth of the underplating layer at the mantle-crust transition (10-15 km
571 below sea level; Fig. 10a, 11). A similar pressure mode was also identified by clinopyroxene
572 barometry performed on lava samples erupted within the NWRZ (Boudoire et al. 2019). Higher
573 pressures are rarely recorded by fluid inclusions (up to 5.7 kbar; PdC; Table 2) but testify of
574 even deeper levels of crystallization located in the mantle (Fig. 10a, 11; Hansteen and Klügel
575 2008) as suggested by geophysical studies (Fontaine et al. 2015). The mantle-crust underplating
576 layer is thus the main level recorded in the most magnesian olivine crystals cargo at Piton de la
577 Fournaise and is the site of entrapment of a CO₂-rich fluid phase and of fluid inclusion
578 reequilibration at high pressure.

579 We note that the average saturation pressure from melt inclusions is in most cases lower
580 than that estimated from fluid inclusions in crystals from lavas emitted along the NWRZ (Fig.
581 10b; Table 4). The fluid inclusion barometry fully confirms that the CO₂-H₂O saturation
582 pressures calculated on the basis of the dissolved contents of these two species are variably and
583 potentially strongly underestimated, because of the diffusion of CO₂ into the shrinkage bubble
584 (Fig. 6c; Wallace et al. 2015; Aster et al. 2016; Tuohy et al. 2016). A good agreement between
585 fluid and melt inclusion barometry is observed in rare melt inclusions either hosted in

586 magnesian olivine crystals from Hudson basalts (Bureau et al. 1999) or recalculated by taking
587 into consideration the effect of the shrinkage bubble (our study). We found that the effect of
588 the shrinkage bubble induces a pressure shift up to 3 kbar in our reconstructed melt inclusions
589 as observed for some melt inclusions from Hudson olivine crystals (black arrow on Fig. 10b).
590 Our results are consistent with previous findings of MacLennan (2017) who suggested that the
591 underestimation of the CO₂-H₂O saturation pressure in melt inclusions is generally <2.5 kbar
592 and that less than 5% of melt inclusions are unaffected by this effect.

593 Our data on melt inclusions with an alkaline affinity provide a new opportunity to
594 reconstruct magma evolution and degassing at Piton de la Fournaise starting from the mantle
595 depth, below the NWRZ. We modeled the evolution of dissolved CO₂-H₂O in melts following
596 both open- and closed-system degassing (1) assuming an initial content of dissolved CO₂ of 1.1
597 wt% in Piton de la Fournaise melts ponding at 10 kbar (Boudoire et al. 2018), and (2) starting
598 from an initial dissolved water content of 1.4 wt% (Fig. 10b). This H₂O content corresponds to
599 that measured in the most magnesian melt inclusions with an alkaline affinity (Pole 1; Table 4)
600 and strikingly matches those with a tholeiitic one (1.3 wt%; Di Muro et al. 2014). Even if melt
601 inclusions show a variable H₂O/K₂O ratio that cannot totally rule out the role of some source
602 effects (as discussed further), the H₂O/K₂O variability mainly results from quite constant H₂O
603 contents at variable K₂O contents (maximum of 1.3 and 1.4 wt% H₂O for high magnesian melt
604 inclusions with a tholeiitic or an alkaline affinity, respectively). This behavior mirrors a possible
605 buffering of the H₂O content dissolved in melts ponding at the depth of the mantle-crust
606 underplating layer, i.e. at the level of important reequilibration processes as suggested above
607 (Boudoire et al. 2019).

608 Our data from recalculated CO₂-H₂O contents in melt inclusions with an alkaline
609 affinity deviate from the modeled degassing trends because of higher CO₂/H₂O ratios. This
610 behavior is commonly reported in many basaltic volcanic systems (see Metrich and Wallace
611 (2008) for a review) where high CO₂/H₂O ratios in melt inclusions, with respect to predicted
612 degassing models, have been attributed to (1) post-trapping processes as H₂O-loss by diffusion
613 across olivine crystals (Massare et al. 2002; Gaetani and Watson 2000; Bucholz et al. 2013),
614 and/or (2) pre-trapping processes affecting magmas (Baker et al. 2005; Spillaert et al. 2006;
615 Witham 2011). In our case,

616 (1) we show that no correlation is observed between H₂O content, FeO_T content, and
617 PEC percentage for the melt inclusions discussed in our study (6a). That minimizes the role of
618 H₂O-loss by re-equilibration and/or diffusion across olivine crystals (Danyushevski et al. 2000;
619 Gaetani and Watson 2000; Massare et al. 2002; Bucholz et al. 2013; Albert et al. 2019). Water

620 loss can be discarded in our study by considering the absence of daughter minerals in melt
621 inclusions and the primitive nature of the entrapped melt (Sobolev and Chaussidon, 1996).

622 (2) The initial dehydration can thus preferentially be attributed to processes affecting
623 melts before their trapping as melt inclusions like: (i) mixing between deep primitive gas-rich
624 melts and shallower more evolved degassed melts (Witham 2011; Myers et al. 2014), (ii)
625 disequilibrium degassing (Baker et al. 2005; Gonnermann and Manga 2005; Pichavant et al
626 2013; Yoshimura 2015), and/or (iii) CO₂-flushing affecting dissolved volatile ratios (Spillaert
627 et al. 2006; Collins et al 2009; Blundy and Cashman 2008; Métrich et al. 2010). On the one
628 hand, the absence in our dataset on melts emitted along the NWRZ of reverse correlation
629 between H₂O and K₂O contents (Fig. 9b) permits to exclude the role of magma mixing between
630 primitive gas-rich melts and more evolved-degassed melts. On the other hand, the slight
631 correlation ($R^2 = 0.55$) between H₂O and S contents (Appendix 6b) does not support the
632 hypothesis of a disequilibrium degassing due to the low diffusivity of S with respect to H₂O
633 and CO₂ in basaltic melts (Pichavant et al. 2013). Conversely, Boudoire et al. (2018) have
634 recently highlighted that about 68 wt% of the initial CO₂ in Piton de la Fournaise melts is
635 already exsolved in the upper mantle, as expected owing the CO₂-dependence on the pressure
636 (Métrich et al. 2014; Di Muro et al. 2016). We stress that massive early-stage CO₂ exsolution
637 may percolate within the upper part of the plumbing system and increase the CO₂/H₂O ratio
638 with respect to the equilibrium degassing trends.

639 Our combined barometric results from fluid inclusions and melt inclusions make the
640 mantle-crust underplating layer the locus of magma ponding and degassing beneath the NWRZ
641 (Bureau et al. 1998; Liuzzo et al. 2015; Boudoire et al. 2018), favouring CO₂/H₂O increase in
642 magmas as recorded by melt inclusions in olivine crystals from peripheral lavas emitted along
643 the NWRZ.

644

645 **Alkali enrichment by degassing-induced clinopyroxene crystallisation** 646 **and mixing between variably evolved melts**

647 The role of clinopyroxene in controlling the deep and early evolution of Piton de la Fournaise
648 magmas has been extensively discussed since the pioneering work of Albarède et al. (1997).
649 We have shown that the trend of alkali enrichment recorded by evolved melt inclusions and
650 mostly in matrix glasses occurs together with CaO/Al₂O₃ decrease and SiO₂ enrichment. A
651 similar trend of evolution was documented for bulk rocks at Piton de la Fournaise and attributed
652 to the role of late clinopyroxene fractionation (Boudoire et al. 2019). Our study of melt

653 inclusions and in particular of dissolved volatile elements provide a new look on the conditions
654 of deep clinopyroxene stability in the range 3-6 kbar and its link with the observed enrichment
655 in alkalis in Piton de la Fournaise residual melts.

656 With respect to the initial melt composition calculated by Boudoire et al. (2019) and
657 used to model melt evolution at Piton de la Fournaise (AlphaMELTS code; Asimow et al. 2003;
658 Ghiorso et al. 2001; Ghiorso & Sack 1995), we model here the evolution of PdF magmas
659 starting from most magnesian melt inclusion documented in our study (inclusion Fo_{87.7} in TB;
660 H₂O = 1.4 wt%; Table 4). Oxygen fugacity was fixed at NNO-0.5 based on the review of Piton
661 de la Fournaise melts made by Pichavant et al. (2016). Nevertheless, oxygen fugacity variability
662 is not expected to play a major role in the Piton de la Fournaise fractionation sequence
663 (Boudoire et al. 2019). Initial temperature (1280 °C) was estimated from the MgO-thermometer
664 of Helz and Thornber (1987) modified by Putirka (2008). The results of a simulated simple
665 fractionation process (red lines on Fig. 11a) highlight that olivine is the first phase to crystallize,
666 clinopyroxene appears at 1190 °C (6 kbar) and 1140 °C (3 kbar) in H₂O-rich melts (1.6 wt%).
667 While these trends fit the composition of lavas from NWRZ peripheral vents (Boudoire et al.
668 2019), they do not fit the composition of more evolved melt inclusions and matrix glasses
669 described in this study. Conversely, in more dehydrated melts, i.e. H₂O = 0.4-0.5 wt%, like the
670 most water-poor melt inclusions (Fig. 10b), clinopyroxene starts fractionating at higher
671 temperature, between 1230°C (at 6 kbar) and 1180°C (at 3 kbar). Modelled differentiation
672 trends (blue lines on Fig. 11a) better fit with most of water-poor evolved melt inclusions and
673 matrix glasses, and overlap uncorrected composition of melt inclusions in TB clinopyroxenes
674 (Fig. 11a; Appendix 3). Our modelling supports the idea that decrease in dissolved water at
675 high pressure, possibly induced by extensive CO₂-flushing at the depth of the mantle-crust
676 underplating layer can play a major role to control the stability of clinopyroxene with respect
677 to olivine. Conversely, inputs of hot volatile rich melts can destabilize and dissolve the early
678 formed clinopyroxene (Boudoire et al. 2019). This process is fully consistent with recent
679 experimental studies made on Piton de la Fournaise melts (Brugier 2016) and resorption
680 textures observed in clinopyroxene crystals (Fig. 5). We thus corroborate our early findings
681 (Boudoire et al. 2019) showing that the alkali enrichment associated with low CaO/Al₂O₃
682 observed in some low magnesian (MgO <9 wt%) matrix glasses and melt inclusions (Fig. 7b,
683 10a) is triggered by clinopyroxene fractionation in dehydrated melts ponding at the depth of the
684 mantle-crust underplating layer and at shallower levels.

685

686 We also document a few melt inclusions with variable $\text{CaO}/\text{Al}_2\text{O}_3$ at intermediate MgO
687 contents (9-10 wt%) that do not fall on the crystallisation trends modelled in the 3-6 kbar
688 pressure range (Fig. 11a). These compositions may track (1) a deeper step of clinopyroxene
689 dissolution/crystallisation (>6 kbar), and/or (2) the effect of magma mixing between deep
690 primitive and more evolved melts (Fig. 11a):

691 (1) high $\text{CaO}/\text{Al}_2\text{O}_3$ ratios (mostly linked to an impoverishment in Al_2O_3) are found in
692 melt inclusions hosted in Fo_{83-86} olivine crystals rather than in $\text{Fo}_{>86}$ olivine crystals. This does
693 not support the hypothesis of an early effect of deep (>6 kbar) clinopyroxene crystallisation but
694 instead of the process of clinopyroxene dissolution during magma recharges discussed in
695 Boudoire et al. (2019) (Fig. 11a; Appendix 6c).

696 (2) Our petrological investigation on olivine and clinopyroxene crystals zoning better
697 supports the hypothesis of magma mixing between deep primitive and more evolved melts (Fig.
698 11a). Actually, the most evolved melts measured (cf PC) and modelled in our study are in
699 equilibrium with Fo_{80-82} olivine crystal (Fig. 11a) recording reverse zoning (Fig. 3). Moreover,
700 their clinopyroxenes display melt inclusions with a negative PEC, and both resorption surface
701 (Fig. 5) and Mg-Cr-rich bands (Fig. 5) hardly attributable to (i) crystal growth effect (Mg and
702 Cr being anticorrelated in sector zoned clinopyroxenes; Kouchi et al. 1983; Mollo et al. 2010;
703 Hammer et al. 2016), and/or (ii) convective thermal effect expected to generate an alternance
704 between Mg-Cr-rich bands (hotter conditions) and Ti-rich zones (cooler conditions;
705 Danyushevski et al. 2000; Welsch et al. 2009; Johnson et al. 2010; Head et al. 2011). We
706 attribute the oscillatory zoned clinopyroxene rims (Fig. 5) to a convective effect during magma
707 transfer (in dykes or sills) as suggested by the progressive Ti/Al ratio increase in rims
708 (barometric proxy; Boudoire et al. 2019).

709 Thus, magma ponding at the depth of the mantle-crust underplating layer (ca. 4 kbar)
710 plays a major role in the geochemical melt variability at Piton de la Fournaise by combining
711 the competitive effects of crystallisation-dehydration processes, magma mixing between
712 variably evolved melts and recycling or even assimilation of pre-existing crystals (Fig. 11b).
713 There, melt dehydration and clinopyroxene crystallization/dissolution controls the alkali
714 enrichment and $\text{CaO}/\text{Al}_2\text{O}_3$ variability reported in lavas emitted in the NWRZ (Lénat et al.
715 2012; Valer et al. 2017; Boudoire et al. 2019). Conversely, intermediate melt compositions, as
716 recorded by steady state basalts emitted in the central area, record effective mixing of magmas
717 injected in the central and shallower crustal reservoir. Mixing is responsible of hiding potential
718 pristine alkali enrichment (Fig. 11b), and leads to the emission of evolved and hybrid lavas as

719 commonly observed at oceanic basaltic volcanoes (Di Muro et al. 2014; Schipper et al. 2015,
720 2016; Jennings et al. 2017).

721

722 **Alkali enrichment by partial melting of a slightly heterogeneous mantle** 723 **source and contamination with low-degree partial melts derived from** 724 **cumulates**

725 Besides the main trend in alkali enrichment, together with CaO/Al₂O₃ decrease and SiO₂
726 enrichment, measured in some evolved melts and matrix glasses and tracking the effect of late-
727 stage clinopyroxene fractionation, we have also documented at Piton de la Fournaise alkali
728 enrichment in high magnesian melt inclusions, i.e., for primitive melts. Interestingly, these high
729 magnesian (MgO >10 wt%) melt inclusions have a highly variable alkali and SiO₂ contents at
730 constant CaO/Al₂O₃ ratio (inclusions Fo_{87.7} and Fo_{87.1}; Fig. 7). They support the role of
731 magmatic processes in the chemical variability of the melts of Piton de la Fournaise other than
732 the crystallization of deep clinopyroxene invoked by Albarède et al. (1997).

733 On one side, the enrichment in major (K₂O, Na₂O) and trace (Ba, Rb) alkali elements
734 in some inclusions (Pole 1) correlates with an enrichment in incompatible trace elements (La,
735 Th) and Cl, and a depletion in SiO₂ (Fig. 7, 8, 9) that could be attributed to a lower melting
736 degree of the source (Reines and Nelson 1998; Bourdon et al. 2005). We used a batch melting
737 model based on the composition of a primitive mantle (Wang et al. 2018) at 3 GPa composed
738 by 53% olivine, 34% clinopyroxene, 8% orthopyroxene, and 5% garnet as previously discussed
739 by Valer et al. (2017) for La Réunion magmas. Crystal-melt partition coefficients are from
740 Salters and Stracke (2004). Through this model we estimate that part of the observed
741 geochemical variability (as recorded by melt inclusions in magnesian inclusions like Pole 1 on
742 Fig. 12a, b) could be explained by a variation of the degree of partial melting from 5% to 7%.
743 Higher melting degrees, 8-10%, are recorded in melt inclusions with a tholeiitic signature (Pole
744 2) and in central magmas like those of the large April 2007 eruption. This range of degrees of
745 partial melting is fully consistent with those previously reported at La Réunion (Ludden 1978;
746 Albarède and Tamagnan 1988; Valer et al. 2017; Vlastelic et al. 2018).

747 On the other side, variable degrees of partial melting of a homogenous mantle source
748 fail to explain the variations of some canonical ratios of highly incompatible elements like
749 Nb/U, Ba/Th and Ce/Pb in melts with either a tholeiitic or an alkaline affinity (Fig. 12c, d;
750 Appendix 7). This is even more true considering that such variations may occur at variable
751 Ba/Th contents and La/Yb ratios (between Pole 1 and Pole 2 on Fig. 8 and 12, respectively).

752 Together with the enrichment in radiogenic Sr of some lavas with an alkaline affinity (Table
753 1), these observations better support the role of some minor source heterogeneities (Huang et
754 al. 2005; Workman and Hart 2005; Jackson and Dasgupta 2008; Vlastelic et al. 2018). At La
755 Réunion, source heterogeneity was invoked to explain (i) the correlation between Sr isotopes
756 and ratios of variably incompatible trace elements as Th/Yb (Fig. 12b; Di Muro et al. 2014;
757 Vlastelic et al. 2018; Boudoire et al. 2019), and (ii) the Nd-Hf-Pb variability observed between
758 the old dormant Piton des Neiges volcano and the currently active Piton de la Fournaise volcano
759 (Bosch et al. 2008; Nauret et al. 2019). Melt extraction from small-scale distinct mantle
760 components within the plume characterized by variable canonical ratios (Ba/Th, Th/Yb on Fig.
761 8 and 12) and Sr isotope signature may explain the geochemical variability in trace elements
762 described in high magnesian melt inclusions (black arrows on Fig. 12). Interestingly, a
763 compilation of bulk rock data from basalts and mugearites of Piton de la Fournaise and Piton
764 des Neiges late-stage activity (Younger Lavas; Smietana 2011), shows that, on average (i) high
765 magnesian melt inclusions with a tholeiitic affinity and marked by low Nb/U, Nb/Y and Ba/Th
766 ratios (Pole 1) mirror the composition of the lava emitted during the large 2007 caldera-forming
767 event of the 5th April 2007 at Piton de la Fournaise whereas (ii) those with an alkaline affinity
768 and marked by high La/Yb, Nb/U, Nb/Y and Ba/Th ratios (inclusion F_{O87.1}) are geochemically
769 close to the composition of late-stage lavas emitted by the Piton des Neiges volcano (Fig. 12c;
770 Appendix 7). As peripheral products investigated in our study lie on the NWRZ, i.e. the rift
771 zone linking Piton des Neiges and Piton de la Fournaise volcanoes, these geochemical
772 signatures raise major issues regarding the current magmatic activity of both PdF and PdN
773 plumbing systems (Fig. 11b).

774

775 Noteworthy, we have documented some evolved melt inclusions (Pole 3) that are even more
776 enriched in some highly incompatible elements (Th, U, La) than the geochemical poles
777 described above (Fig. 8). In these melt inclusions, the selective increase in Th and U leads to
778 lower Ba/Th and Nb/U ratios, respectively (Fig. 12c, d). Such observation contrasts with
779 previous findings attributing the variability of these ratios only to source processes at Piton de
780 la Fournaise (Valer et al. 2017). Instead, this behaviour reflects a possible effect of mixing
781 between mantle-derived magmas and low-degree partial melts derived from cumulates having
782 experienced an enrichment in fluid mobile elements (Bureau et al. 1998; Pietruszka et al. 2009;
783 Boudoire et al. 2019). To test this hypothesis, we have calculated the mixing of:

784 (1) a melt with a composition intermediate between those found in Pole 1 and Pole 2
785 ('TB047-8-2' in Appendix 3) as expected by the formation of hybrid magmas at the depth of
786 the mantle-crust underplating layer (Fig. 10a), and

787 (2) a melt produced by low degree ($\approx 1\%$; Pietruszka et al. 2009; Boudoire et al. 2019)
788 of partial melting from a standard (i.e. not enriched) wehrlite cumulate ('94.G1'; Upton et al.
789 2000; Boudoire et al. 2019). Wehrlite cumulates, together with dunite cumulates, are expected
790 to represent the composition of the dense layer of underplated material (Pietruszka et al. 2009)
791 related to the crystallization of magmas ponding at the mantle-crust interface (Charvis et al.
792 1999; Fontaine et al. 2015).

793 The simulated mixing curves show that the contamination of mantle-derived melts with
794 such partial melts from wehrlite cumulates may lead to an enrichment in incompatible trace
795 elements (Th, U) together with a decrease of the Ba/Th and Nb/U ratios as observed in part of
796 the evolved melt inclusions (Pole 3) and secondary melt inclusions (Fig. 12b, c, d; Bureau et
797 al. 1998; Di Muro et al. 2014). We calculate that less than 2% of assimilation of partial melt
798 from wehrlite cumulates is required to explain most of enrichment in trace elements of these
799 evolved melt inclusions. Meanwhile, a larger degree of assimilation (up to 10%) of partial melt
800 is required to explain the most enriched compositions. As suggested by Pietruszka et al. (2009),
801 enrichment in Th and U of the assimilated melts can explain these extreme compositions.

802

803 Previous studies at Piton de la Fournaise have highlighted the role of U remobilization
804 in bulk rock (Albarède et al. 1997; Pietruszka et al. 2009; Vlastelic et al. 2011) and melt
805 inclusions (Di Muro et al. 2014) related to a leaching and transfer by either meteoritic-
806 hydrothermal water or by magmatic fluids. On the one hand, we measure the highest Nb/U
807 ratios in part of the matrix glasses (Fig. 12d) and this process could be related to late-stage U-
808 leaching by meteoritic waters during weathering (Aiuppa et al. 2000) or by acidic gas in
809 fumarolic environment (Vlastelic et al. 2011). However, the unaltered appearance of those
810 glasses and the absence of systematic SiO₂ enrichment together with a decrease in Al₂O₃, MgO,
811 FeO_T minimize the role of late-stage hydrothermal alteration (Vlastelic et al. 2011). On the
812 other hand, the U-enrichment recorded in some evolved melt inclusions occurs (i)
813 independently of the composition in major elements, and (ii) together with an enrichment in Th
814 and La leading to constant U/Th and La/Th ratios (Fig. 8). These results exclude a
815 contamination by late-stage acidic gases or by meteoritic waters to explain the observed
816 chemical variability (Aiuppa et al. 2000; Vlastelic et al. 2011). Instead, we suggest that
817 cumulates located at the depth of the mantle-crust underplating layer (i.e., preserved from

818 interaction with meteoritic waters) may be metasomatized by deep (mantle) magmatic fluids
819 enriched in a range of elements as U, Th, La (Hedenquist and Lowenstern 1994; Pietruszka et
820 al. 2009) and radiogenic He produced by U-Th radioactive decay (Boudoire et al. 2020). We
821 speculate that this process may be favoured by the occurrence of CO₂-rich fluids in the upper
822 mantle beneath Piton de la Fournaise and in particular at the depth of mantle-crust underplating
823 layer (Lowenstern (2001) and references therein).

824

825 **CONCLUSION**

826 In this study, we characterize for the first-time crystal zoning, fluid and melt inclusions
827 composition and barometry in olivine and clinopyroxene crystals from fast quenched mafic
828 basalts having an alkaline affinity, and emitted along the North-West rift zone linking Piton de
829 la Fournaise and Piton des Neiges volcanoes. We document here the most primitive melt
830 inclusions found so far on La Réunion (MgO = 11.3 wt%), and we show that alkaline
831 enrichment in melts emitted from peripheral vents is mainly related to the competitive effects
832 between: (1) various degrees of partial melting of a slightly heterogenous mantle source, and
833 (2) late-stage clinopyroxene crystallisation at the depth of the mantle-crust underplating layer.
834 We suggest that deep pyroxene stability is controlled by magma water content and melt
835 dehydration by extensive CO₂-flushing.

836 We show that the mantle-crust underplating layer is a major site of magma evolution
837 and that magma ponding at that depth (10-15 km below sea level) experiences: (i) CO₂-flushing
838 and increase in CO₂/H₂O ratios, (ii) mixing between melts recording slight source
839 heterogeneity, (iii) crystal recycling/assimilation, and (iv) contamination with low-degree
840 partial melts from metasomatized cumulates. Magma mixing is even more advanced for melts
841 ascending in the crustal part of the volcano plumbing system, leading to the apparent
842 geochemical homogeneity of evolved and degassed melts emitted in the central area during the
843 frequent activity of Piton de la Fournaise.

844 Primitive melts in magnesian-rich unzoned olivine (Fo_{>86}) provide snapshots of the
845 mantellic part of the plumbing system (>15 km below sea level) and record a range of
846 geochemical signatures that lies between that of the current activity of Piton de la Fournaise
847 and that of the late-stage activity of the quiescent neighbouring Piton des Neiges volcano. These
848 findings raise major issue regarding the activity of the NWRZ linking both volcanoes and more
849 generally on the current magmatic activity below the whole island.

850

851 **ACKNOWLEDGMENTS**

852 We acknowledge helpful technical support from N. Braukmüller, A. Kronz, G. Labeau, G.
853 Montagnac, and D. Neuville during microanalysis. V. Zanon is gratefully acknowledged for
854 stimulating discussions on ocean island magmatism. We are in debt with C. Sundermeyer and
855 G. Wörner for the analytical support and constructive discussions. The Université de La
856 Réunion and the ANR “STRAP” (ANR-14-CE03-0004) have funded this work. We also
857 acknowledge the French government IDEX-ISITE initiative 16-IDEX-0001 (CAP 20-25). This
858 is Laboratory of Excellence ClerVolc contribution number XXX.

859 **REFERENCES**

- 860 Adam J, Green T (2006) Trace element partitioning between mica- and amphibole-bearing
861 garnet lherzolite and hydrous basanitic melt: 1. Experimental results and the investigation
862 of controls on partitioning behaviour. *Contrib. Mineral. Petrol.* **152**:1-17.
863 DOI:10.1007/s00410-006-0085-4.
- 864 Aiuppa A, Allard P, D’Alessandro W, Michel A, Parello F, Treuil M, Valenza M (2000)
865 Mobility and fluxes of major, minor and trace metals during basalt weathering and
866 groundwater transport at Mt. Etna volcano (Sicily). *Geochimica et Cosmochimica Acta*
867 **64**:1827-1841. DOI:10.1016/S0016-7037(00)00345-8.
- 868 Albarède F, Luais B, Fitton G, Semet M, Kaminski E, Upton BGJ, Bachèlery P, Cheminée JL
869 (1997) The geochemical regimes of Piton de la Fournaise volcano (Réunion) during the
870 last 530000 years. *Journal of Petrology* **38**:171-201.
871 <https://doi.org/10.1093/petroj/38.2.171>.
- 872 Albarède F, Tamagnan V (1988) Modeling the recent geochemical evolution of the Piton de la
873 Fournaise volcano, Reunion Island. *Journal of Petrology* **29**:997-1030.
874 <https://doi.org/10.1093/petrology/29.5.997>.
- 875 Albert H, Costa F, Di Muro A, Herrin J, Métrich N, Deloule E (2019) Magma interactions,
876 crystal mush formation, timescales, and unrest during caldera collapse and lateral
877 eruption at ocean island basaltic volcanoes (Piton de la Fournaise, La Réunion). *Earth*
878 *and Planetary Science Letters* **515**:187-199. <https://doi.org/10.1016/j.epsl.2019.02.035>.
- 879 Asimow PD, Dixon JE, Langmuir CH (2003) A hydrous melting and fractionation model for
880 mid-ocean ridge basalts: Application to the Mid-Atlantic Ridge near the Azores.
881 *Geochemistry Geophysics Geosystems* **5**:Q01E16. doi:10.1029/2003GC000568.
882 <https://doi.org/10.1029/2003GC000568>.

883 Aster EM, Wallace PJ, Moore LR, Watkins J, Gazel E, Bodnar RJ (2016) Reconstructing CO₂
884 concentrations in basaltic melt inclusions using Raman analysis of vapor bubbles. *J.*
885 *Volcanol. Geotherm. Res.* **323**:148–162. DOI:10.1016/j.jvolgeores.2016.04.028.

886 Baker DR, Freda C, Brooker RA, Scarlato P (2005) Volatile diffusion in silicate melts and its
887 effects on melt inclusions. *Annals of Geophysics* **48**(4-5). [https://doi.org/10.4401/ag-](https://doi.org/10.4401/ag-3227)
888 3227.

889 Bakker RJ (2003) Package FLUIDS 1. Computer programs for analysis of fluid inclusion data
890 and for modelling bulk fluid properties. *Chemical Geology* **194**(1-3):3-23.
891 [https://doi.org/10.1016/S0009-2541\(02\)00268-1](https://doi.org/10.1016/S0009-2541(02)00268-1).

892 Blundy J, Cashman K (2008) Petrologic reconstruction of magmatic system variables and
893 processes. *Reviews in Mineralogy & Geochemistry* **69**:179-239.
894 DOI:10.2138/rmg.2008.69.6.

895 Boivin P, Bachèlery P (2009) Petrology of 1977 to 1998 eruptions of Piton de la Fournaise, La
896 Réunion Island. *Journal of Volcanology and Geothermal Research*. Elsevier B.V.
897 **184**:109–125. DOI:10.1016/j.jvolgeores.2009.01.012.

898 Bosch D, Blichert-Toft J, Moynier F, Nelson BK, Telouk P, Gillot PY, Albarède F (2008) Pb,
899 Hf and Nd isotope compositions of the two Réunion volcanoes (Indian Ocean): a tale of
900 two small-scale mantle blobs. *Earth and Planetary Science Letters* **265**:748-768.

901 Boudoire G (2017) Architecture et dynamique des systèmes magmatiques associés aux volcans
902 basaltiques : exemple du Piton de la Fournaise. PhD thesis, Université de La Réunion –
903 OVPF/IPGP, Sainte-Clotilde, La Réunion, France, 500pp.

904 Boudoire G, Brugier YA, Di Muro A, Wörner G, Arienzo I, Métrich N, Zanon V, Braukmüller
905 N, Kronz A, Le Moigne Y, Michon L (2019) Eruptive activity on the western flank of
906 Piton de la Fournaise (La Réunion Island, Indian Ocean): insights on magma transfer,
907 storage and evolution at an oceanic volcanic island. *Journal of Petrology* **60**(9):1717-
908 1752. <https://doi.org/10.1093/petrology/egz045>.

909 Boudoire G, Liuzzo M, Di Muro A, Ferrazzini V, Michon L, Grassa F, Derrien A, Villeneuve
910 N, Bourdeu A, Brunet C, Giudice G, Gurrieri S. (2017) Investigating the deepest part of
911 a volcano plumbing system: evidence for an active magma path below the western flank
912 of Piton de la Fournaise (La Réunion Island). *Journal of Volcanology and Geothermal*
913 *Research* **341**:193-207. <https://doi.org/10.1016/j.jvolgeores.2017.05.026>.

914 Boudoire G, Rizzo AL, Di Muro A, Grassa F, Liuzzo M (2018) Extensive CO₂ degassing in
915 the upper mantle beneath oceanic basaltic volcanoes: First insights from Piton de la

916 Fournaise volcano (La Réunion Island). *Geochimica et Cosmochimica Acta* **235**:376-401.
917 DOI:10.1016/j.gca.2018.06.004.

918 Boudoire G, Rizzo AL, Arienzo I, Di Muro A (2020) Major contribution of a primitive mantle
919 plume component in paroxysmal eruption of Piton de la Fournaise: inferences from
920 helium isotopes. *Scientific Reports* **10**:1-16. doi: 10.1038/s41598-020-66260-x.

921 Brugier YA (2016) Magmatologie du Piton de la Fournaise (Ile de la Réunion) - Approche
922 Volcanologique, Pétrologique et Expérimentale. Ph.D. thesis, Université d'Orléans,
923 Orléans, France, 273pp.

924 Bucholz CE, Gaetani GA, Behn MD, Shimizu N (2013) Post-entrapment modification of
925 volatiles and oxygen fugacity in olivine-hosted melt inclusions. *Earth Planet. Sci. Lett.*
926 **374**:145–155. <https://doi.org/10.1016/j.epsl.2013.05.033>.

927 Bureau H, Métrich N, Pineau F, Semet M (1998a) Magma-conduit interaction at Piton de la
928 Fournaise volcano (Réunion island): a melt and fluid inclusion study. *J Volcanol*
929 *Geotherm Res* **84**:39–60.

930 Bureau H, Métrich N, Semet MP, Staudacher T (1999) Fluid-magma decoupling in a hotspot
931 volcano. *Geophysical Research Letters* **26**:3501.

932 Bureau H, Pineau F, Métrich N, Semet M, Javoy M (1998b) A melt and fluid inclusion study
933 of the gas phase at Piton de la Fournaise volcano (Réunion island). *Chem Geol* **147**:115–
934 130.

935 Charvis P, Laesanpura A, Gallart J, Hirn A, Lepine J, de Voogd B, Minshull TA, Hello Y,
936 Pontoise B (1999). Spatial distribution of hotspot material added to the lithosphere under
937 La Reunion, from wide-angle seismic data. *Journal of Geophysical Research, Solid Earth*
938 **104(2)**:2875–2893. DOI:10.1029/98jb02841.

939 Collins SJ, Pyle DM, Maclennan J (2009) Melt inclusions track pre-eruption storage and
940 dehydration of magmas at Etna. *Geology* **37(6)**:571-574.
941 <https://doi.org/10.1130/G30040A.1>.

942 Danyushevsky LV, Della-Pasqua FN, Sokolov S (2000) Re-equilibration of melt inclusions
943 trapped by magnesian olivine phenocrysts from subduction-related magmas: petrological
944 implications. *Contrib. Mineral. Petrol.* **138**:68–83. <https://doi.org/10.1007/PL00007664>.

945 Dasgupta R, Jackson MG, Lee CTA (2010) Major element chemistry of ocean island basalts -
946 Conditions of mantle melting and heterogeneity of mantle source. *Earth and Planetary*
947 *Science Letters* **289**:377–392. <https://doi.org/10.1016/j.epsl.2009.11.027>.

- 948 Davis MG, Garcia MO, Wallace P (2003) Volatiles in glasses from Mauna Loa Volcano,
949 Hawai'i : implications for magma degassing and contamination, and growth of Hawaiian
950 volcanoes. *Contrib. Mineral. Petrol.* **144**:570–591. DOI:10.1007/s00410-002-0416-z.
- 951 De Maesschalck R, Jouan-Rimbaud D, Massart DL (2000) The Mahalanobis distance.
952 *Chemometrics and Intelligent Laboratory Systems* **50**:1–18.
953 [https://doi.org/10.1016/S0169-7439\(99\)00047-7](https://doi.org/10.1016/S0169-7439(99)00047-7).
- 954 Di Muro A, Métrich N, Allard P, Aiuppa A, Burton M, Galle B, Staudacher T (2016) Magma
955 Degassing at Piton de la Fournaise Volcano. In: Bachèlery P, Lénat JF, Di Muro A,
956 Michon L (ed) *Active volcanoes of the southwest Indian Ocean: Piton de la Fournaise
957 and Karthala*. Active volcanoes of the world. Springer, Berlin, pp. 203-222.
- 958 Di Muro A, Métrich N, Vergani D, Rosi M, Armienti P, Fougeroux T, Deloule E, Arienzo I,
959 Civetta L (2014) The shallow plumbing system of Piton de la Fournaise Volcano (La
960 Réunion Island, Indian Ocean) revealed by the major 2007 caldera-forming eruption.
961 *Journal of Petrology* **55**:1287–1315. <https://doi.org/10.1093/petrology/egu025>.
- 962 Di Muro A, Staudacher T, Ferrazzini V, Métrich M, Villemant B, Besson P, Garofalo C (2015)
963 Shallow magma storage at Piton de la Fournaise Volcano after the 2007 summit caldera
964 collapse tracked in Pele's hairs. In: Carey R, Cayol V, Poland M, Weis D (ed) *Hawaiian
965 volcanoes: from source to surface*. Geophysical monograph series 208, pp. 189–212.
- 966 Duan X (2014) A general model for predicting the solubility behavior of H₂O-CO₂ fluids in
967 silicate melts over a wide range of pressure, temperature and compositions. *Geochimica
968 et Cosmochimica Acta* **125**:582-609. DOI:10.1016/j.gca.2013.10.018.
- 969 Duan Z, Møller N, Weare JH (1992) Molecular dynamics simulation of PVT properties of
970 geological fluids and a general equation of state of nonpolar and weakly polar gases up
971 to 2000 K and 20,000 bar. *Geochim. Cosmochim. Acta* **56**:3839 – 3845.
972 [https://doi.org/10.1016/0016-7037\(92\)90175-I](https://doi.org/10.1016/0016-7037(92)90175-I).
- 973 Duan Z, Møller N, Weare JH (1996) A general equation of state for supercritical fluid mixtures
974 and molecular dynamics simulation of mixture PVTX properties. *Geochim. Cosmochim.
975 Acta* **60**:1209 – 1216. [https://doi.org/10.1016/0016-7037\(96\)00004-X](https://doi.org/10.1016/0016-7037(96)00004-X).
- 976 Fall A, Tattitch B, Bodnar RJ (2011) Combined microthermometric and Raman spectroscopic
977 technique to determine the salinity of H₂O–CO₂–NaCl fluid inclusions based on clathrate
978 melting. *Geochimica et Cosmochimica Acta* **75**(4):951-964.
979 DOI:10.1016/j.gca.2010.11.021

- 980 Famin V, Welsch B, Okumura S, Bachèlery P, Nakashima S (2009) Three differentiation stages
981 of a single magma at Piton de la Fournaise volcano (Reunion hot spot). *Geochemistry,*
982 *Geophysics, Geosystems* **10**(1). DOI:10.1029/2008gc002015.
- 983 Fisk MR, Upton BGJ, Ford CE, White M (1988) Geochemical and experimental study of the
984 genesis of magmas of Réunion island, Indian Ocean. *Journal of Geophysical Research*
985 **93**:4933-4950. <https://doi.org/10.1029/JB093iB05p04933>.
- 986 Fontaine FR, Barruol G, Tkalčić H, Wölbern I, Rümpler G, Bodin T, Haugmard M (2015)
987 Crustal and uppermost mantle structure variation beneath La Réunion hotspot track.
988 *Geophysical Journal International* **203**:107–126. <https://doi.org/10.1093/gji/ggv279>.
- 989 Frezzotti ML, Tecce F, Casagli A (2012) Raman spectroscopy for fluid inclusion analysis.
990 *Journal of Geochemical Exploration* **112**:1–20.
991 <https://doi.org/10.1016/j.gexplo.2011.09.009>.
- 992 Gaetani GA, Watson EB (2002) Modelling the major-element evolution of olivine-hosted melt
993 inclusions. *Chem. Geol.* **183**:25–41. doi:10.1016/S0009-2541(01)00370-9.
- 994 Gallart J, Driad L, Charvis P, Sapin M, Hirn A, Diaz J, de Voogd B, Sachpazi M (1999)
995 Perturbation to the lithosphere along the hotspot track of La Réunion from an offshore-
996 onshore seismic transect. *Journal of Geophysical Research* **104**:2895-2908.
997 DOI:10.1029/98JB02840.
- 998 Gerbault M, Fontaine FJ, Rabinowicz M, Bystricky M (2017) Elastic flexure controls magma
999 trajectories and explains the offset of primary volcanic activity upstream of mantle plume
1000 axis at La Réunion and Hawaii hotspot islands. *Earth and Planetary Science Letters*
1001 **462**:142-156. DOI:10.1016/j.epsl.2017.01.013.
- 1002 Ghiorso MS, Sack RO (1995) Chemical Mass-Transfer in Magmatic Processes IV. A Revised
1003 and Internally Consistent Thermodynamic Model for the Interpolation and Extrapolation
1004 of Liquid-Solid Equilibria in Magmatic Systems at Elevated-Temperatures and Pressures.
1005 *Contributions to Mineralogy and Petrology* **119**:197-212.
1006 <https://doi.org/10.1007/BF00307281>.
- 1007 Ghiorso MS, Hirschmann MM, Reiners PW, Kress VC (2001) The pMELTS: A revision of
1008 MELTS for improved calculation of phase relations and major element partitioning
1009 related to partial melting of the mantle to 3 GPa. *Geochemistry Geophysics Geosystems*
1010 **3**:1030. doi:10.1029/2001GC000217.
- 1011 Gonnermann HM, Manga M (2005) Nonequilibrium magma degassing: results from modeling
1012 of the ca. 1340 AD eruption of Mono Craters, California. *Earth and Planetary Science*
1013 *Letters* **238**(1):1-16. DOI:10.1016/j.epsl.2005.07.021.

1014 Hagiwara Y, Yoshida K, Yoneda A, Torimoto J, Yamamoto J (2021) Experimental variable
1015 effects on laser heating of inclusions during Raman spectroscopic analysis. *Chemical*
1016 *Geology* **559**:119928. DOI:10.1016/j.chemgeo.2020.119928.

1017 Hart S, Gaetani GA (2016) Experimental determination of Pb partitioning between sulfide melt
1018 and basalt melt as a function of P, T and X. *Geochim. Cosmochim. Acta* **185**:9–20.
1019 DOI:10.1016/j.gca.2016.01.030.

1020 Hammer JE, Jacob S, Welsch B, Hellebrand E, Sinton J (2016) Clinopyroxene in postshield
1021 Haleakala ankaramite. 1. Efficacy of thermobarometry. *Contributions to Mineralogy and*
1022 *Petrology***171**. doi:10.1007/s00410-015-1212-x.

1023 Hansteen TH, Klügel A (2008) Fluid inclusion thermobarometry as a tracer for magmatic
1024 processes. *Reviews in Mineralogy & Geochemistry* **69**:143-177.
1025 <https://doi.org/10.2138/rmg.2008.69.5>.

1026 Hanyu T, Dunai TJ, Davies GR, Kaneoka I, Nohda S, Uto K (2001) Noble gas study of the
1027 Reunion hot spot: evidence for distinct less-degassed mantle sources. *Earth Planet. Sci.*
1028 *Lett.* **193**:83–98. doi:10.1016/S0012-821X(01)00489-7.

1029 Hanyu T, Yamamoto J, Kimoto K, Shimizu K, Ushikubo T (2020) Determination of total CO₂
1030 in melt inclusions with shrinkage bubbles. *Chemical Geology* **557**:119855.
1031 <https://doi.org/10.1016/j.chemgeo.2020.119855>.

1032 Hartley ME, Maclennan J, Edmonds M, Thordarson T (2014) Reconstructing the deep CO₂
1033 degassing behaviour of large basaltic fissure eruptions. *Earth Planet. Sci. Lett.* **393**:120–
1034 131. DOI:10.1016/j.epsl.2014.02.031.

1035 Head E, Shaw AM, Wallace PJ, Sims KWW, Carn SA (2011) Insight into volatile behavior at
1036 Nyamuragira volcano (D.R. Congo, Africa) through olivine-hosted melt inclusions.
1037 *Geochem. Geophys. Geosyst.* **12**:Q0AB11. doi:10.1029/2011GC003699.

1038 Hedenquist JW, Lowenstern JB (1994) The role of magmas in the formation of hydrothermal
1039 ore deposits. *Nature* **370**:519-527. <https://doi.org/10.1038/370519a0>.

1040 Helz RT, Clague DA, Sisson TW, Thornber CR (2014) Petrologic insights into basaltic
1041 volcanism at historically active Hawaiian volcanoes. In: Poland MP, Takahashi TJ,
1042 Landowski CM (ed) *Characteristics of Hawaiian Volcanoes*, pp. 237-292.

1043 Helz RT, Thornber CR (1987) Geothermometry of Kilauea Iki lava lake, Hawaii. *Bulletin of*
1044 *Volcanology* **49**:651-668. <https://doi.org/10.1007/BF01080357>.

1045 Hildner E, Klügel A, Hauff F (2011) Magma storage and ascent during the 1995 eruption of
1046 Fogo, Cape Verde Archipelago. *Contributions to Mineralogy and Petrology* **162**(4):751.
1047 DOI:10.1007/s00410-011-0623-6.

1048 Huang S, Frey FA, Blichert-Toft J, Fodor RV, Bauer GR, Xu G (2005) Enriched components
1049 in the Hawaiian plume: evidence from Kahoolawe Volcano, Hawaii. *Geochemistry,*
1050 *Geophysics, Geosystems* **6**(11). doi:10.1029/2005GC001012.

1051 Jackson MG, Dasgupta R (2008) Compositions of HIMU, EM1, and EM2 from global trends
1052 between radiogenic isotopes and major elements in ocean island basalts. *Earth and*
1053 *Planetary Science Letters* **276**(1):175-186. DOI:10.1016/j.epsl.2008.09.023.

1054 Jenner FE, O'Neill HSC (2012) Major and trace analysis of basaltic glasses by laser-ablation
1055 ICP-MS. *Geochemistry, Geophysics, Geosystems* **13**:3. doi:10.1029/2011GC003890.

1056 Jennings ES, Gibson SA, Maclennan J, Heinonen JS (2017) Deep mixing of mantle melts
1057 beneath continental flood basalt provinces: constraints from olivine-hosted melt
1058 inclusions in primitive magmas. *Geochimica et Cosmochimica Acta* **196**:36-57.
1059 DOI:10.1016/j.gca.2016.09.015.

1060 Johnson ER, Wallace PJ, Cashman KV, Granados HD (2010) Degassing of volatiles (H₂O,
1061 CO₂, S, Cl) during ascent, crystallization, and eruption at mafic monogenetic volcanoes
1062 in central Mexico. *J. Volcanol. Geotherm. Res.* **197**:225–238.
1063 doi:10.1016/j.jvolgeores.2010.02.017.

1064 Klügel A, Longpré MA, Garcia-Canada L, Stix J (2015) Deep intrusions, lateral magma
1065 transport and related uplift at ocean island volcanoes. *Earth and Planet. Sci. Lett.*
1066 **431**:140-149. DOI:10.1016/j.epsl.2015.09.031.

1067 Kobayashi T, Yamamoto J, Hirajima T, Ishibashi H, Hirano N, Lai Y, Prikhod'Ko V, Arai S
1068 (2012) Conformity and precision of CO₂ densimetry in CO₂ inclusions:
1069 Microthermometry versus Raman microspectroscopic densimetry. *Journal of Raman*
1070 *Spectroscopy* **43**(8):1126-1133. <https://doi.org/10.1002/jrs.3134>.

1071 Kouchi A, Sugawara Y, Kashima K, Sunagawa I (1983) Laboratory growth of sector zoned
1072 clinopyroxenes in the system CaMgSi₂O₆-CaTiAl₂O₆. *Contributions to Mineralogy and*
1073 *Petrology* **83**:177-184. <https://doi.org/10.1007/BF00373091>.

1074 Lamadrid HM, Moore LR, Moncada D, Rimstidt JD, Burruss RC, Bodnar RJ (2017)
1075 Reassessment of the Raman CO₂ densimeter. *Chemical Geology.* doi:
1076 10.1016/j.chemgeo.2016.12.034.

1077 Lange RA, Carmichael IS (1987) Densities of Na₂O-K₂O-CaO-MgO-FeO-Fe₂O₃-Al₂O₃-
1078 TiO₂-SiO₂ liquids: new measurements and derived partial molar properties. *Geochimica*
1079 *et Cosmochimica Acta* **51**(11):2931-2946. [https://doi.org/10.1016/0016-7037\(87\)90368-](https://doi.org/10.1016/0016-7037(87)90368-1)
1080 1.

- 1081 Lénat JF, Bachèlery P, Merle O (2012) Anatomy of Piton de la Fournaise volcano (La Réunion,
1082 Indian Ocean). *Bulletin of Volcanology* **74**:1945-1961. DOI:10.1007/s00445-012-0640-
1083 y.
- 1084 Lengliné O, Duputel Z, Ferrazzini V (2016) Uncovering the hidden signature of a magmatic
1085 recharge at Piton de la Fournaise volcano using small earthquakes. *Geophys. Res. Lett.*
1086 **43**. <http://dx.doi.org/10.1002/2016GL068383>.
- 1087 Liuzzo M, Di Muro A, Giudice G, Michon L, Ferrazzini V, Gurrieri S (2015) New evidence of
1088 CO₂ soil degassing anomalies on Piton de la Fournaise volcano and the link with volcano
1089 tectonics structures. *Geochem. Geophys. Geosys.* **16**:4388–4404.
1090 doi:10.1002/2015GC006032.
- 1091 Lowenstern JB (2001) Carbon dioxide in magmas and implications for hydrothermal systems.
1092 *Mineralium Deposita* **36**:480-502. DOI:10.1007/s001260100185.
- 1093 Ludden JN (1978) Magmatic evolution of the basaltic shield volcanoes of Reunion Island.
1094 *Journal of Volcanology and Geothermal Research* **4**(1-2):171-198.
1095 [https://doi.org/10.1016/0377-0273\(78\)90035-5](https://doi.org/10.1016/0377-0273(78)90035-5).
- 1096 MacLennan J (2017) Bubble formation and decrepitation control the CO₂ content of olivine-
1097 hosted melt inclusions. *Geochemistry, Geophysics, Geosystems* **18**.
1098 doi:10.1002/2016GC006633.
- 1099 Mahoney JJ, Duncan RA, Khan W, Gnos E, McCormick GR (2002) Cretaceous volcanic rocks
1100 of the South Tethyan suture zone, Pakistan: implications for the Réunion hotspot and
1101 Deccan Traps. *Earth Planet. Sci. Lett.* **203**:295-310. [https://doi.org/10.1016/S0012-
1102 821X\(02\)00840-3](https://doi.org/10.1016/S0012-821X(02)00840-3).
- 1103 Massare D, Métrich N, Clocchiatti R (2002) High-temperature experiments on silicate melt
1104 inclusions in olivine at 1 atm: inference on temperatures of homogenization and H₂O
1105 concentrations. *Chemical Geology* **183**:87-98.
- 1106 Maurel C, Maurel P (1982) Etude expérimentale de l'équilibre Fe²⁺–Fe³⁺ dans les spinelles
1107 chromifères et les liquides silicatés basiques coexistant à 1 atm. *CR Acad Sci* **295**:209–
1108 212.
- 1109 Mercier M, Di Muro A, Giordano D, Métrich N, Lesne P, Pichavant M, Scaillet B, Clocchiatti
1110 R, Montagnac G (2009) Influence of glass polymerisation and oxidation on microRaman
1111 water analysis in alumino-silicate glasses. *Geochimica et Cosmochimica Acta* **73**:197-
1112 217. Doi:10.1016/j.gca.2008.09.030.
- 1113 Mercier M, Di Muro A, Métrich N, Giordano D, Belhadj O, Mandeville CW (2010)
1114 Spectroscopic analysis (FTIR, RAMAN) of water in mafic and intermediate glasses and

1115 glass inclusions. *Geochimica et Cosmochimica Acta* **74**:5641-5656.
1116 doi:10.1016/j.gca.2010.06.020.

1117 Métrich N, Allard P, Aiuppa A, Bani P, Bertagnini A, Shinohara H, Massare D (2011) Magma
1118 and volatile supply to post-collapse volcanism and block resurgence in Siwi Caldera
1119 (Tanna Island, Vanuatu Arc). *Journal of Petrology* **52**(6):1077-1105.
1120 <https://doi.org/10.1093/petrology/egr019>.

1121 Métrich N, Bertagnini A, Di Muro A (2010) Conditions of magma storage, degassing and ascent
1122 at Stromboli: New insights into the volcano plumbing system with inferences on the
1123 eruptive dynamics. *J. Petrol.* **51**:603–626. doi:10.1093/petrology/egp083.

1124 Métrich N, Wallace PJ (2008) Volatile abundances in basaltic magmas and their degassing
1125 paths tracked by melt inclusions. Minerals, Inclusions, and Volcanic Processes. *Reviews*
1126 *in Mineralogy and Geochemistry* **69**:363–402. DOI:10.2138/rmg.2008.69.10.

1127 Métrich N, Zanon V, Créon L, Hildenbrand A, Moreira M, Marques FO (2014) Is the ‘Azores
1128 hotspot’ a wet spot? Insights from the geochemistry of fluid and melt inclusions in olivine
1129 of Pico basalts. *Journal of Petrology* **55**(2):377-393.
1130 <https://doi.org/10.1093/petrology/egt071>.

1131 Michon L, Ferrazzini V, Di Muro A, Villeneuve N, Famin V (2015) Rift zones and magma
1132 plumbing system of Piton de la Fournaise volcano: how do they differ from Hawaii and
1133 Etna. *J Volcanol Geotherm Res* **303**:112-129. Doi:10.1016/j.jvolgeores.2015.07.031.

1134 Mollo S, Del Gaudio P, Ventura G, Iezzi G, Scarlato P (2010) Dependence of clinopyroxene
1135 composition on cooling rate in basaltic magmas: Implications for thermobarometry.
1136 *Lithos* **118**:302–312. DOI:10.1016/j.lithos.2010.05.006.

1137 Moore LR, Gazel E, Tuohy R, Lloyd AS, Esposito R, Steele-Macinnis M, Hauri ER, Wallace
1138 PJ, Plank T, Bodnar RJ (2015) Bubbles matter: an assessment of the contribution of vapor
1139 bubbles to melt inclusion budgets. *Am. Mineral.* **100**(4):806–823.
1140 <https://doi.org/10.2138/am-2015-5036>.

1141 Myers ML, Geist DJ, Rowe MC, Harpp KS, Wallace PJ, Dufek J (2014) Replenishment of
1142 volatile-rich mafic magma into a degassed chamber drives mixing and eruption of
1143 Tungurahua volcano. *Bulletin of Volcanology* **76**(11):872. DOI:10.1007/s00445-014-
1144 0872-0.

1145 Nauret F, Famin V, Vlastélic I, Gannoun A (2019) A trace of recycled continental crust in the
1146 Réunion hotspot. *Chemical Geology* **524**:67-76. DOI:10.1016/j.chemgeo.2019.06.009.

- 1147 Papale P, Moretti R, Barbato D (2006) The compositional dependence of the saturation surface
1148 of H₂O + CO₂ fluids in silicate melts. *Chem. Geol.* **229**:78–95.
1149 DOI:10.1016/j.chemgeo.2006.01.013.
- 1150 Peltier A, Bachèlery P, Staudacher T (2009) Magma transport and storage at Piton de la
1151 Fournaise (La Réunion) between 1972 and 2007: A review of geophysical and
1152 geochemical data. *J. Volcanol. Geotherm. Res.* **184**:93-108.
1153 <http://dx.doi.org/10.1016/j.jvolgeores.2008.12.008>.
- 1154 Peltier A, Poland M, Staudacher T (2015) Are Piton de la Fournaise (La Réunion) and Kīlauea
1155 (Hawai'i) Really Analog Volcanoes? In: Carey R, Cayol V, Poland M, Weis D (ed)
1156 Hawaiian Volcanoes: From Source to Surface. John Wiley & Sons, Inc, (ch 23). doi:
1157 10.1002/9781118872079.
- 1158 Peters BJ, Day JMD, Taylor LA (2016) Early mantle heterogeneities in the Réunion hotspot
1159 source inferred from highly siderophiles elements in cumulate xenoliths. *Earth and*
1160 *Planetary Science Letters* **448**:150-160. <https://doi.org/10.1016/j.epsl.2016.05.015>.
- 1161 Pichavant M, Brugier Y, Di Muro A (2016) Petrological and experimental constraints. In:
1162 Bachèlery P, Lénat JF, Di Muro A, Michon L (ed) *Active volcanoes of the southwest*
1163 *Indian Ocean: Piton de la Fournaise and Karthala*. Active volcanoes of the world.
1164 Springer, Berlin.
- 1165 Pichavant M, Di Carlo I, Rotolo SG, Scaillet B, Burgisser A, Le Gall N, Martel C (2013)
1166 Generation of CO₂-rich melts during basalt magma ascent and degassing. *Contributions*
1167 *to Mineralogy and Petrology* **166**(2):545-561. Doi:10.1007/s00410-013-0890-5.
- 1168 Pietruszka AJ, Hauri EH, Blichert-Toft J (2009) Crustal contamination of mantle-derived
1169 magmas within Piton de la Fournaise volcano, Reunion island. *Journal of Petrology*
1170 **50**:661–684. <https://doi.org/10.1093/petrology/egp016>.
- 1171 Poland MP, Miklius A, Montgomery-Brown EK (2014) Magma supply, storage, and transport
1172 at shield-stage Hawaiian volcanoes, In Poland MP, Takahashi TJ, Landowski CM (ed)
1173 Characteristics of Hawaiian Volcanoes, U.S. Geol. Surv., pp. 179-234.
1174 doi:10.3133/pp18015.
- 1175 Putirka KD (2008) Thermometers and Barometers for Volcanic Systems. *Reviews in*
1176 *Mineralogy and Geochemistry* **69**:61–120. <https://doi.org/10.2138/rmg.2008.69.3>.
- 1177 Quidelleur X, Holt JW, Salvany T, Bouquerel H (2010) The double structure of the Réunion
1178 geomagnetic event based on new K-Ar ages from the type locality, massif de la
1179 Montagne, Réunion Island (Indian Ocean) and assessment of the global context. *Geophys*
1180 *J Int* **182**:699–710.

1181 Rae AS, Edmonds M, Maclennan J, Morgan D, Houghton B, Hartley ME, Sides I (2016) Time
1182 scales of magma transport and mixing at Kīlauea Volcano, Hawai'i. *Geology* **44**(6):463-
1183 466. <https://doi.org/10.1130/G37800.1>.

1184 Reiners PW, Nelson BK (1998) Temporal-compositional-isotopic trends in rejuvenated-stage
1185 magmas of Kauai, Hawaii, and implications for mantle melting processes. *Geochimica et*
1186 *Cosmochimica Acta* **62**(13):2347–2368. doi:10.1016/S0016-7037(98)00141-0.

1187 Rhodes JM (1996) Geochemical stratigraphy of lava flows sampled by the Hawaii Scientific
1188 Drilling Project. *Journal of Geophysical Research* **101**:11729.
1189 <https://doi.org/10.1029/95JB03704>.

1190 Riker J (2005) The 1859 Eruption of Mauna Loa Volcano, Hawai'i: controls on the
1191 development of long lava channels. Master's thesis, University of Oregon, pp. 117–127.

1192 Salaün A, Villemant B, Semet MP, Staudacher T (2010) Cannibalism of olivine-rich cumulate
1193 xenoliths during the 1998 eruption of Piton de la Fournaise (La Réunion hotspot):
1194 Implications for the generation of magma diversity. *Journal of Volcanology and*
1195 *Geothermal Research* **198**:187–204. <https://doi.org/10.1016/j.jvolgeores.2010.08.022>.

1196 Salters VJ, Stracke A (2004) Composition of the depleted mantle. *Geochemistry, Geophysics,*
1197 *Geosystems* **5**(5). <https://doi.org/10.1029/2003GC000597>.

1198 Salvany T, Lahitte P, Nativel P, Gillot PY (2012) Geomorphic evolution of the Piton des Neiges
1199 volcano (Réunion Island, Indian Ocean): competition between volcanic construction and
1200 erosion since 1.4 Ma. *Geomorphology* **136**:132–147.
1201 doi:10.1016/j.geomorph.2011.06.009.

1202 Schipper CI, Jakobsson SP, White JDL, Palin JM, Bush-Marcinowski T (2015) The Surtsey
1203 Magma Series. *Sci. Rep.* **5**:11498. doi: 10.1038/srep11498.

1204 Schipper, CI, Le Voyer M, Moussallam Y, White JDL, Thordarson T, Kimura JI, Chang Q
1205 (2016) Degassing and magma mixing during the eruption of Surtsey Volcano (Iceland,
1206 1963-1967): the signatures of a dynamic and discrete rift propagation event. *Bull*
1207 *Volcanol* **78**. Doi :10.1007/s00445-016-1025-4.

1208 Smietana M (2011) Pétrologie, géochronologie (K–Ar) et géochimie élémentaire et isotopique
1209 (Sr, Nd, Hf, Pb) des laves anciennes de La Réunion. Implications sur la construction de
1210 l'édifice volcanique. PhD thesis, Université de La Réunion, Sainte-Clotilde, France,
1211 362pp.

1212 Sobolev AV, Chaussidon M (1996) H₂O concentrations in primary melts from supra-
1213 subduction zones and mid-ocean ridges: implications for H₂O storage and recycling in

1214 the mantle. *Earth and Planetary Science Letters* **137**(1-4):45-55. DOI:10.1016/0012-
1215 821X(95)00203-O.

1216 Song Y, Chou IM, Hu W, Robert B, Lu W (2009) CO₂ density-Raman shift relation derived
1217 from synthetic inclusions in fused silica capillaries and its application. *Acta Geologica*
1218 *Sinica* **83**(5):932-938.

1219 Spilliaert N, Allard P, Métrich N, Sobolev A (2006) Melt inclusion record of the conditions
1220 of ascent, degassing and extrusion of volatile-rich alkali basalt during the powerful 2002
1221 flank eruption of Mount Etna (Italy). *Journal of Geophysical Research* **111**:B04203.
1222 doi:10.1029/2005/JB003934.

1223 Sublett JDM, Sendula E, Lamadrid H, Steele-MacInnis M, Spiekermann G, Burruss RC,
1224 Bodnar RJ (2020) Shift in the Raman symmetric stretching band of N₂, CO₂, and CH₄
1225 as a function of temperature, pressure, and density. *Journal of Raman Spectroscopy*
1226 **51**(3):555-568. <https://doi.org/10.1002/jrs.5805>.

1227 Tuohy RM, Wallace PJ, Loewen MW, Swanson DA, Kent AJR (2016) Magma transport and
1228 olivine crystallization depths in Kīlauea's east rift zone inferred from experimentally
1229 rehomogenized melt inclusions. *Geochim. Cosmochim. Acta.* **185**:232–250.
1230 <https://doi.org/10.1016/j.gca.2016.04.020>.

1231 Upton BGJ, Semet M, Joron JL (2000) Cumulate clasts in the Bellecombe Ash Member, Piton
1232 de la Fournaise, Réunion Island, and their bearing on cumulative processes in the
1233 petrogenesis of the Réunion lavas. *J Volcanol Geotherm Res* **104**:297–318.
1234 [https://doi.org/10.1016/S0377-0273\(00\)00212-2](https://doi.org/10.1016/S0377-0273(00)00212-2).

1235 Upton BGJ, Wadsworth WJ (1966) The basalts of Réunion island, Indian Ocean. *Bull Volcanol*
1236 **29**:7–23. <https://doi.org/10.1007/BF02597136>.

1237 Valer M (2016) Caractérisation des stockages magmatiques profonds sous les volcans de La
1238 Réunion: apports de la pétro-géochimie des inclusions magmatiques. PhD thesis,
1239 Université Clermont Auvergne, Clermont-Ferrand, France.

1240 Valer M, Schiano P, Bachèlery P (2017) Geochemical characteristics of La Réunion mantle
1241 plume source inferred from olivine-hosted melt inclusions from the adventive cones of
1242 Piton de la Fournaise volcano (La Réunion Island). *Contribution. Mineral. Petrol.*
1243 **172**:74. DOI:10.1007/s00410-017-1397-2.

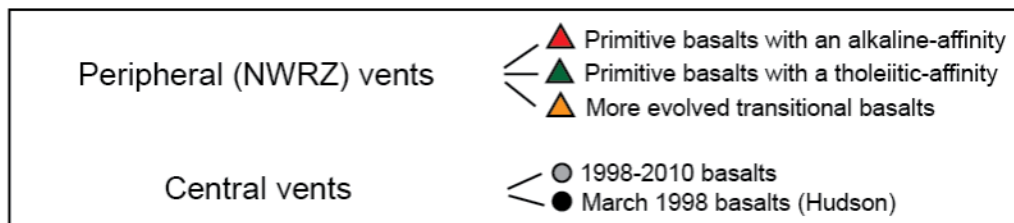
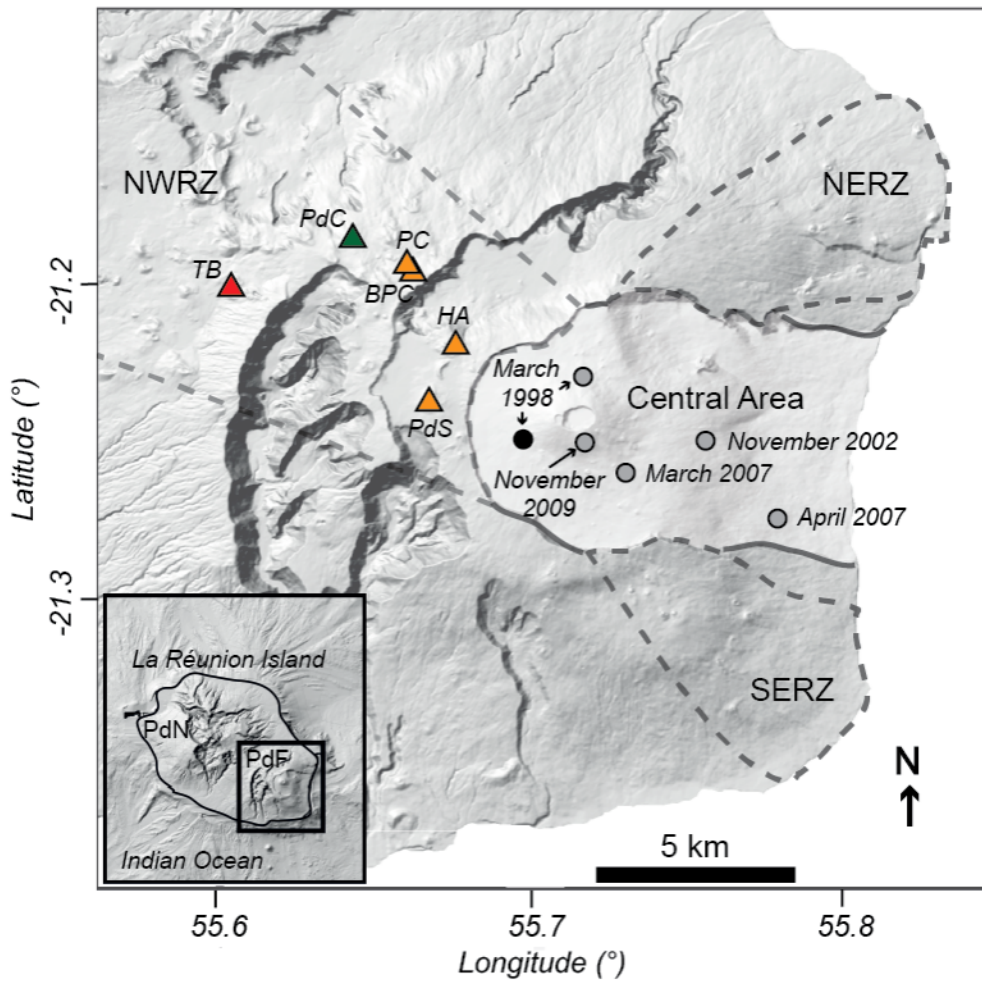
1244 van den Kerkhof AM, Olsen SN (1990) A natural example of superdense CO₂ inclusions:
1245 microthermometry and Raman analysis. *Geochimica et Cosmochimica Acta* **55**:2533–
1246 2544.

- 1247 Venugopal S, Schiavi F, Moune S, Bolfan-Casanova N, Druitt T, Williams-Jones G (2020)
1248 Melt inclusion vapour bubbles: the hidden reservoir for major and volatile elements.
1249 *Scientific reports* **10**(1):1-14. DOI:10.1038/s41598-020-65226-3.
- 1250 Vigouroux N, Williams-Jones AE, Wallace P, Staudacher T (2009) The November 2002
1251 eruption of Piton de la Fournaise, Réunion: tracking the pre-eruptive thermal evolution
1252 of magma using melt inclusions. *Bull Volcanol* **71**:1077–1089. DOI:10.1007/s00445-
1253 009-0287-5.
- 1254 Villemant B, Salaün A, Staudacher T (2009) Evidence for a homogeneous primary magma at
1255 Piton de la Fournaise (La Réunion): A geochemical study of matrix glass, melt inclusions
1256 and Pélé's hairs of the 1998–2008 eruptive activity. *Journal of Volcanology and*
1257 *Geothermal Research* **184**:79-92. doi: 10.1016/j.jvolgeores.2009.03.015.
- 1258 Villeneuve N, Bachèlery P (2006) Revue de la typologie des éruptions au Piton de la Fournaise,
1259 processus et risques volcaniques associés. *CyberGeo* [http://](http://dx.doi.org/10.4000/cybergeo.2536)
1260 dx.doi.org/10.4000/cybergeo.2536 (<http://cybergeo.revues.org/2536>).
- 1261 Vlastélic I, Di Muro A, Bachèlery P, Gurioli L, Auclair D, Gannoun A (2018) Control of source
1262 fertility on the eruptive activity of Piton de la Fournaise volcano, La Réunion. *Scientific*
1263 *Reports* **8**:14478. <https://doi.org/10.1038/s41598-018-32809-0>.
- 1264 Vlastélic I, Staudacher T, Bachèlery P, Télouk P, Neuville D, Benbakkar M (2011) Lithium
1265 isotope fractionation during magma degassing: Constraints from silicic differentiates and
1266 natural gas condensates from Piton de la Fournaise. *Chemical Geology* **284**:26-34.
1267 DOI:10.1016/j.chemgeo.2011.02.002.
- 1268 Wallace PJ, Kamenetsky VS, Cervantes P (2015) Melt inclusion CO₂ contents, pressures of
1269 olivine crystallization, and the problem of shrinkage bubbles. *Am. Mineral.* **100**(4):787–
1270 794. doi:10.2138/am-2015-5029.
- 1271 Walther G, Frese I, Di Muro A, Kueppers U, Michon L, Métrich N (2015) The eruptive history
1272 of the Trous Blancs pit craters, La Réunion Island: the origin of a 24 km long lava flow.
1273 *EGU General Assembly 2015*.
- 1274 Welsch B, Faure F, Bachèlery P, Famin V (2009) Microcrysts record transient convection at
1275 Piton de la Fournaise Volcano (La Réunion hotspot). *Journal of Petrology* **50**:2287-2305.
1276 <https://doi.org/10.1093/petrology/egp076>.

- 1277 Welsch B, Famin V, Baronnet A, Bachèlery P (2013) Dendritic crystallization: a single process
1278 for all textures of olivine in basalts? *Journal of Petrology* **54**:539-574.
1279 DOI:10.1093/petrology/egs077.
- 1280 Witham F (2011) Conduit convection, magma mixing, and melt inclusion trends at persistent
1281 degassing volcanoes. *Earth and Planetary Science Letters* **301**:345-352.
1282 DOI:10.1016/j.epsl.2010.11.017.
- 1283 Workman RK, Hart SR (2005) Major and trace element composition of the depleted MORB
1284 mantle (DMM). *Earth and Planetary Science Letters* **231**:53-72.
1285 doi:10.1016/j.epsl.2004.12.005.
- 1286 Yoshimura S (2015) Diffusive fractionation of H₂O and CO₂ during magma degassing.
1287 *Chemical Geology* **411**:172-181. <https://doi.org/10.1016/j.chemgeo.2015.07.003>.
- 1288 Wang X, Chou IM, Hu W, Burruss RC, Sun Q, Song Y (2011) Raman spectroscopic
1289 measurements of CO₂ density: Experimental calibration with high-pressure optical cell
1290 (HPOC) and fused silica capillary capsule (FSCC) with application to fluid inclusion
1291 observations. *Geochimica et Cosmochimica Acta* **75**(14):4080-4093.
1292 DOI:10.1016/j.gca.2011.04.028.
- 1293 Wang HS, Lineweaver CH, Ireland TR (2018). The elemental abundances (with uncertainties)
1294 of the most Earth-like planet. *Icarus* **299**:460-474.
1295 <https://doi.org/10.1016/j.icarus.2017.08.024>.
- 1296 Zanon V, Frezzotti ML, Peccerillo A (2003) Magmatic feeding system and crustal magma
1297 accumulation beneath Vulcano Island (Italy): Evidence from CO₂ fluid inclusions in
1298 quartz xenoliths. *J Geophys Res* **108**. doi:10.1029/2002JB002140.

1299

1300 **FIGURES**



1301

1302

1303

1304

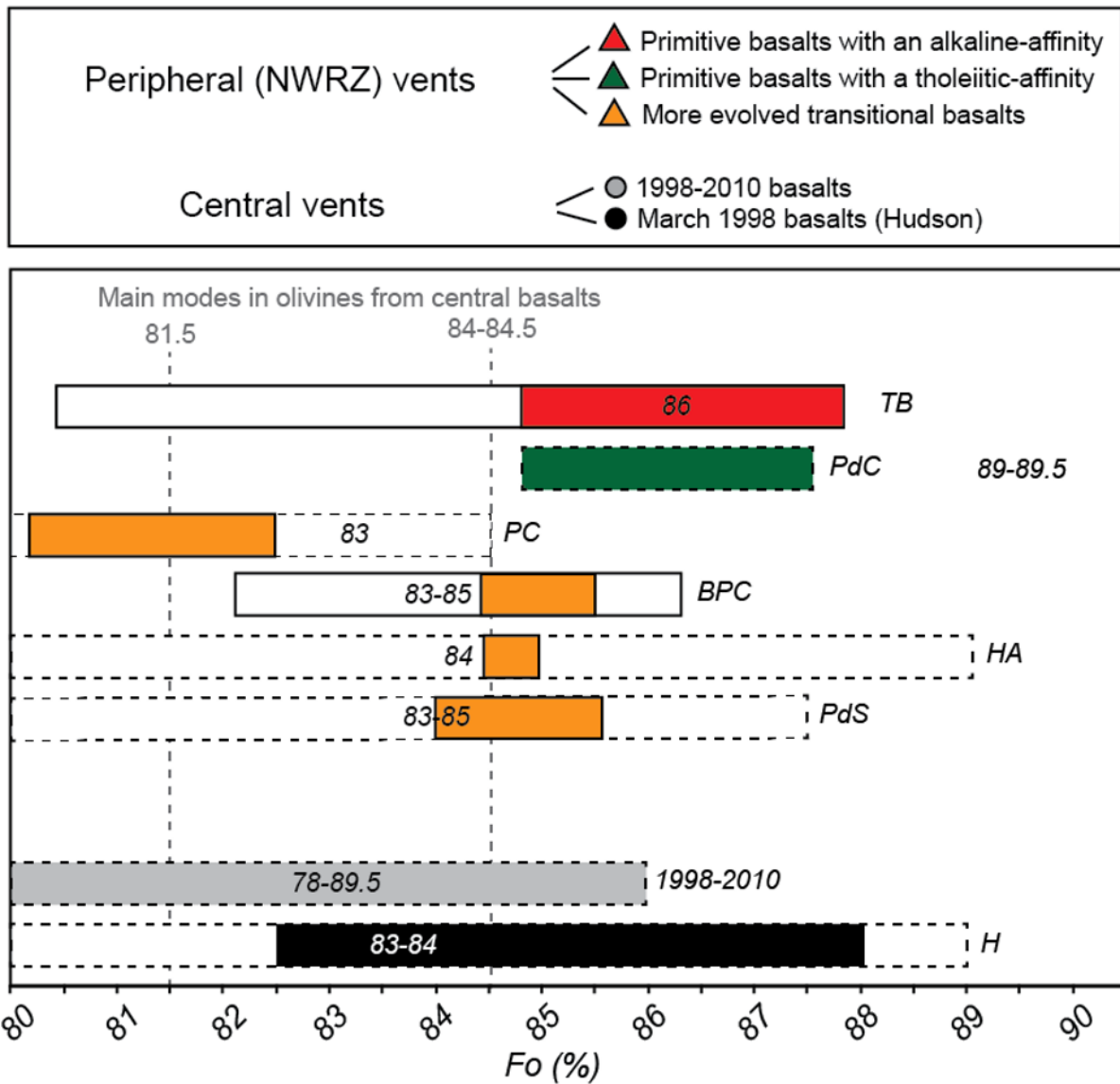
1305

1306

1307

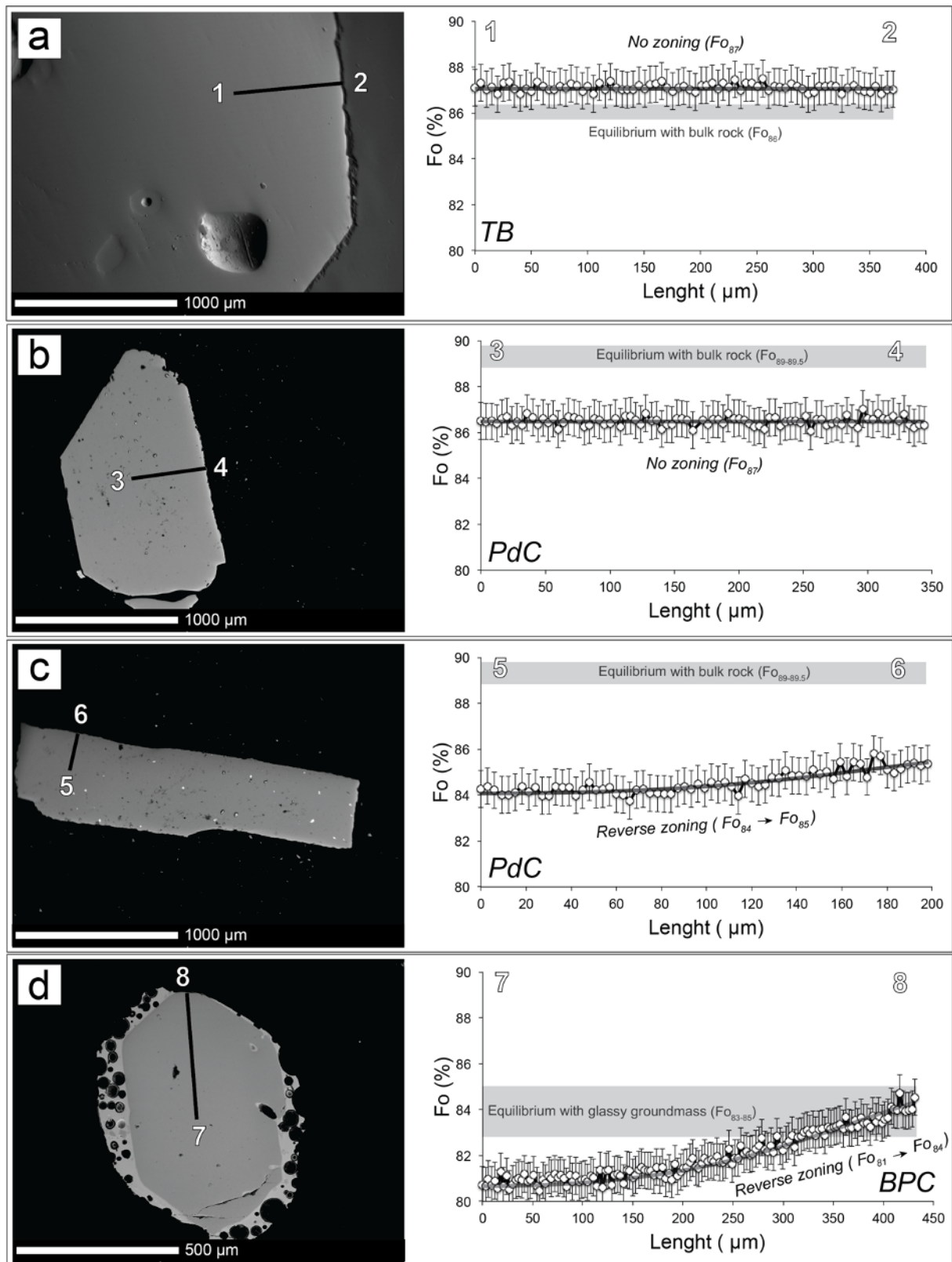
1308

Fig. 1: Digital Elevation Model of the Piton de La Fournaise volcano, showing the main geomorphological features and the general trends of the three rift systems (dashed lines) converging towards the central area (Enclos Fouqué caldera): the North-West rift zone (NWRZ), the South-East rift zone (SERZ), and the North-East rift zone (NERZ). The location of the eruptive products described in this study is reported (TB: Trous Blancs; PdC: Piton de Caille; PC: Petit Cratère; BPC: Below Petit Cratère; PdS: Plaine des Sables fall; HA: Piton Haüy). BPC refers to a lapilli layer close to the eruption of Petit Cratère.



1309
 1310
 1311
 1312
 1313
 1314
 1315

Fig. 2: Range of forsterite contents of olivine crystals in transitional basalts of Piton de la Fournaise. The analyses performed in this study are highlighted with continuous lines. Analyses from previous studies are highlighted with dashed lines (see Boudoire et al. (2019) for a review). Numbers are the composition of olivine composition at the equilibrium with the hosting bulk rock. Filled coloured ranges are the composition of olivine crystals hosting melt inclusions described in this study.



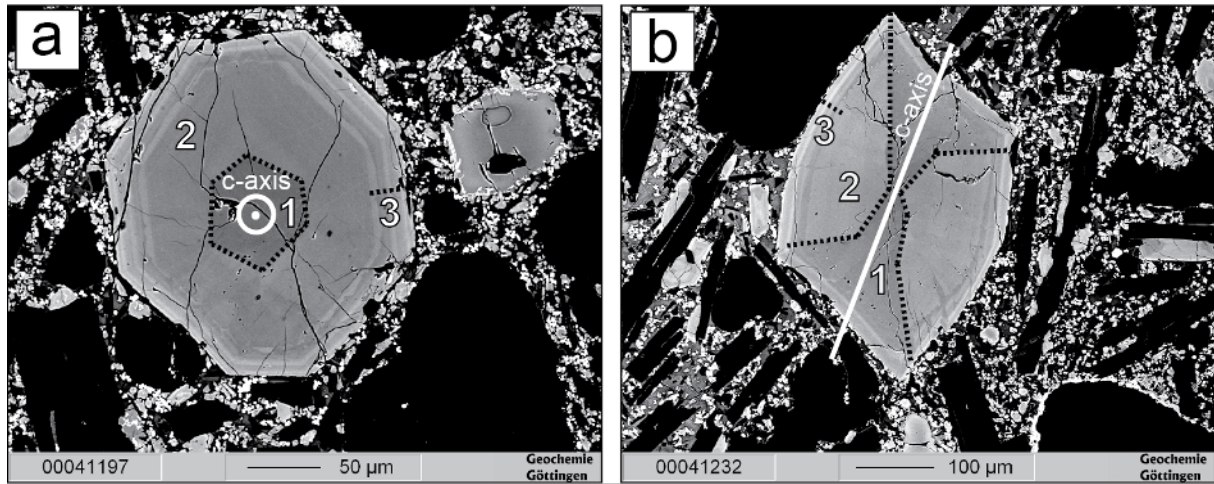
1316

1317 **Fig. 3: Representative zoning profiles (Fo-content) in olivine crystals from (a) Trous Blancs (TB), (b, c)**

1318 **Piton de Caille (PdC), and (d) Below Petit Cratère (BPC) transitional basalts. The analytical uncertainty is**

1319 **reported for each point of the profiles.**

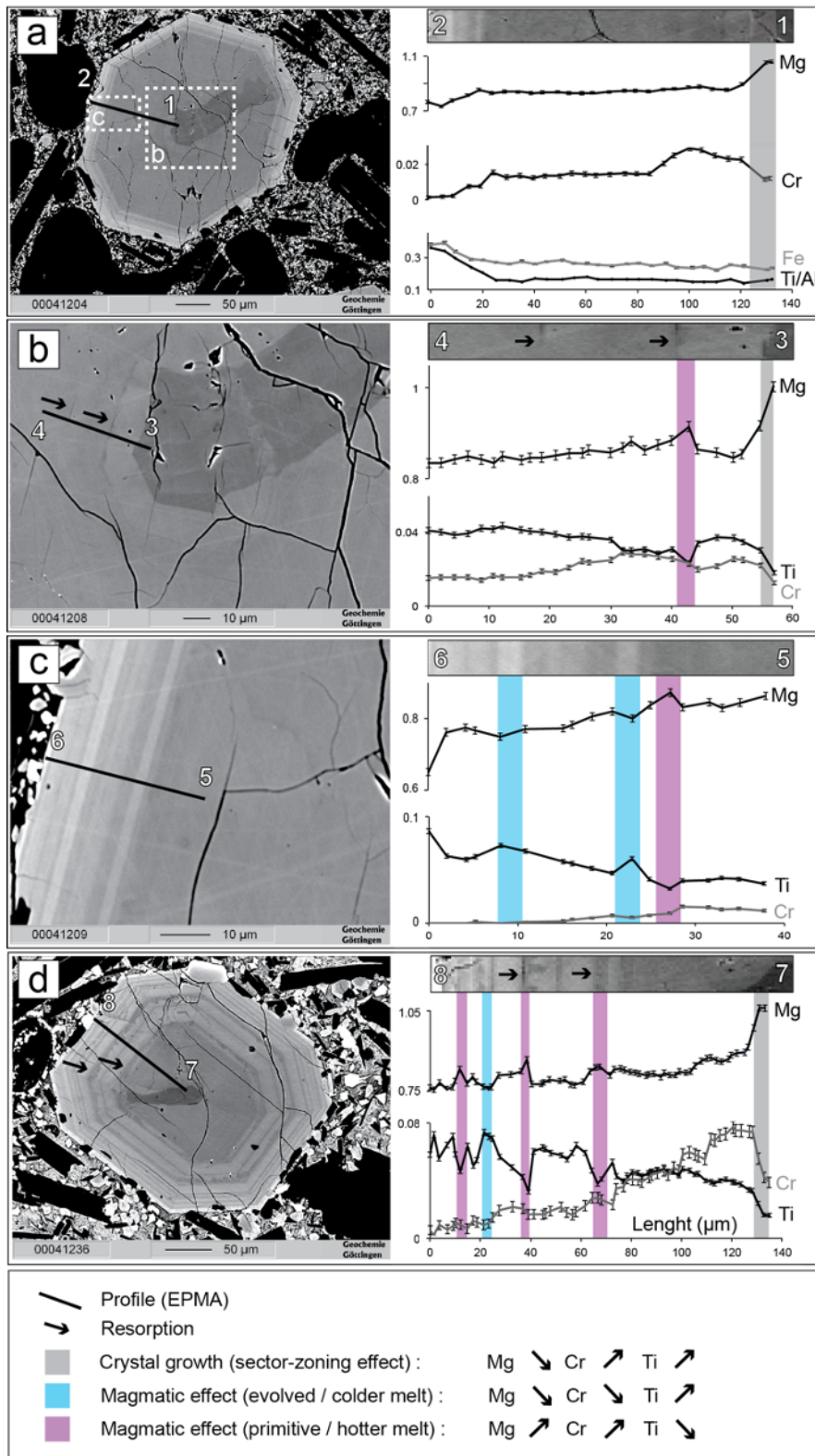
1320



1321

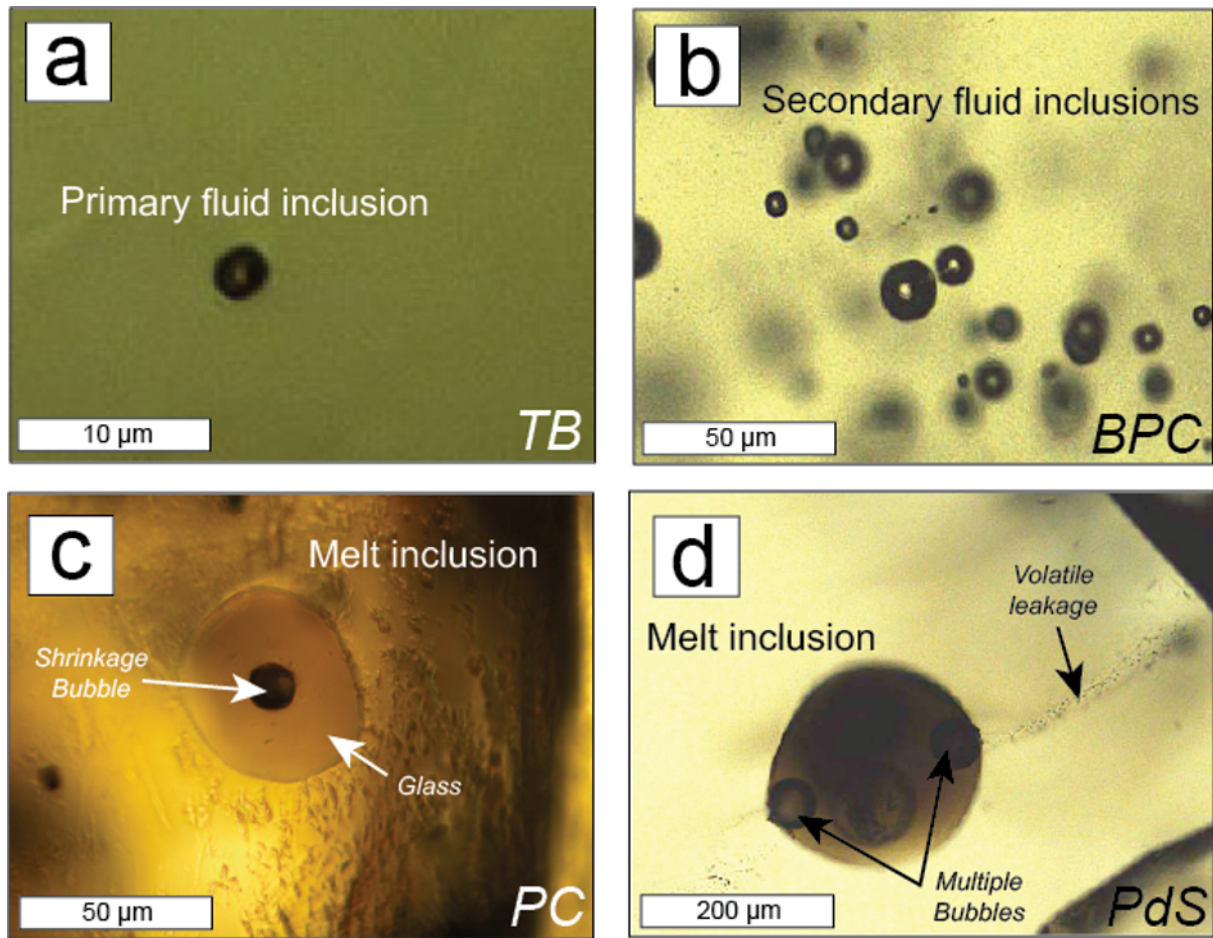
1322 **Fig. 4: BSE imagery of sector-zoned clinopyroxene in Petit Cratère (PC) bombs. (a) Section cut across the**
 1323 **c-axis. (b) Section cut along the c-axis. 1: Mg-Si-rich and Ti-Al-Cr-poor core dark sectors. 2: Ti-Al-Cr-rich**
 1324 **and Mg-Si-poor core bright sectors. 3: oscillatory zoning at the rims.**

1325



1326

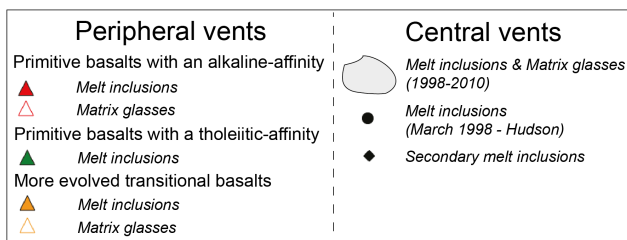
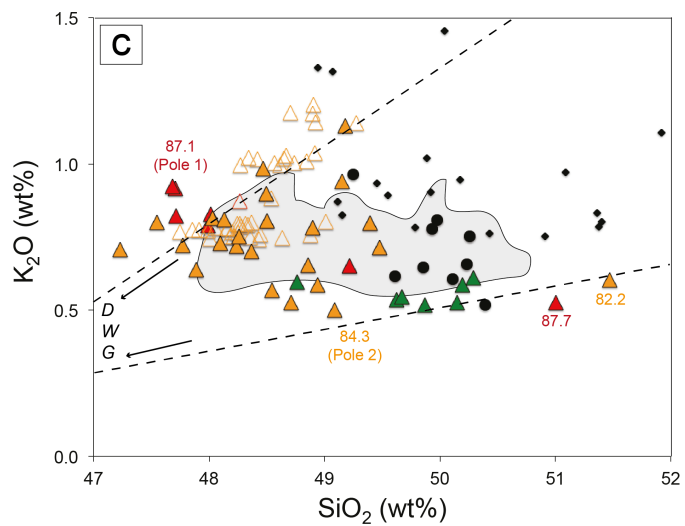
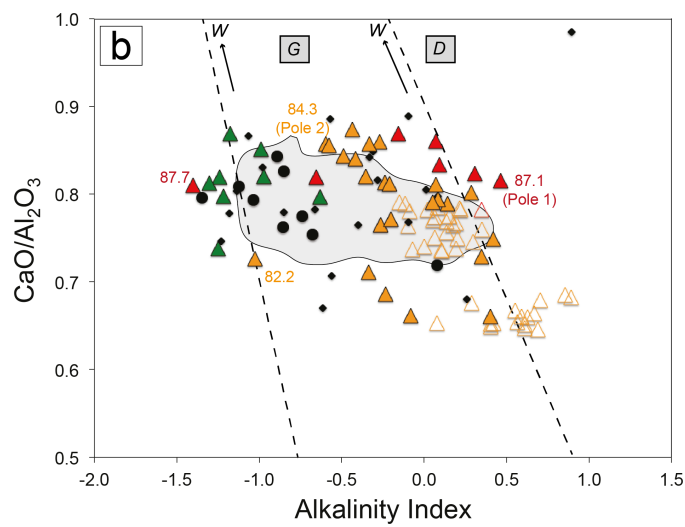
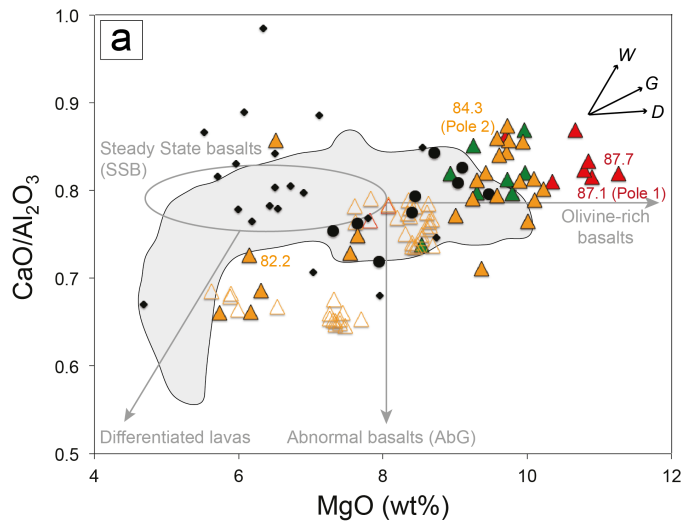
1327 **Fig. 5: Backscattered electrons (BSE) imagery of sector-zoned clinopyroxene in Petit Cratère (PC) bombs.**
 1328 **Elemental profiles are reported with an analytical spacing varying between 5 µm (a) and 2 µm (b, c, d). The**
 1329 **analytical uncertainty is reported for each point of the profiles. Only the most significant compositional**
 1330 **changes are highlighted (filled coloured bars).**



1332

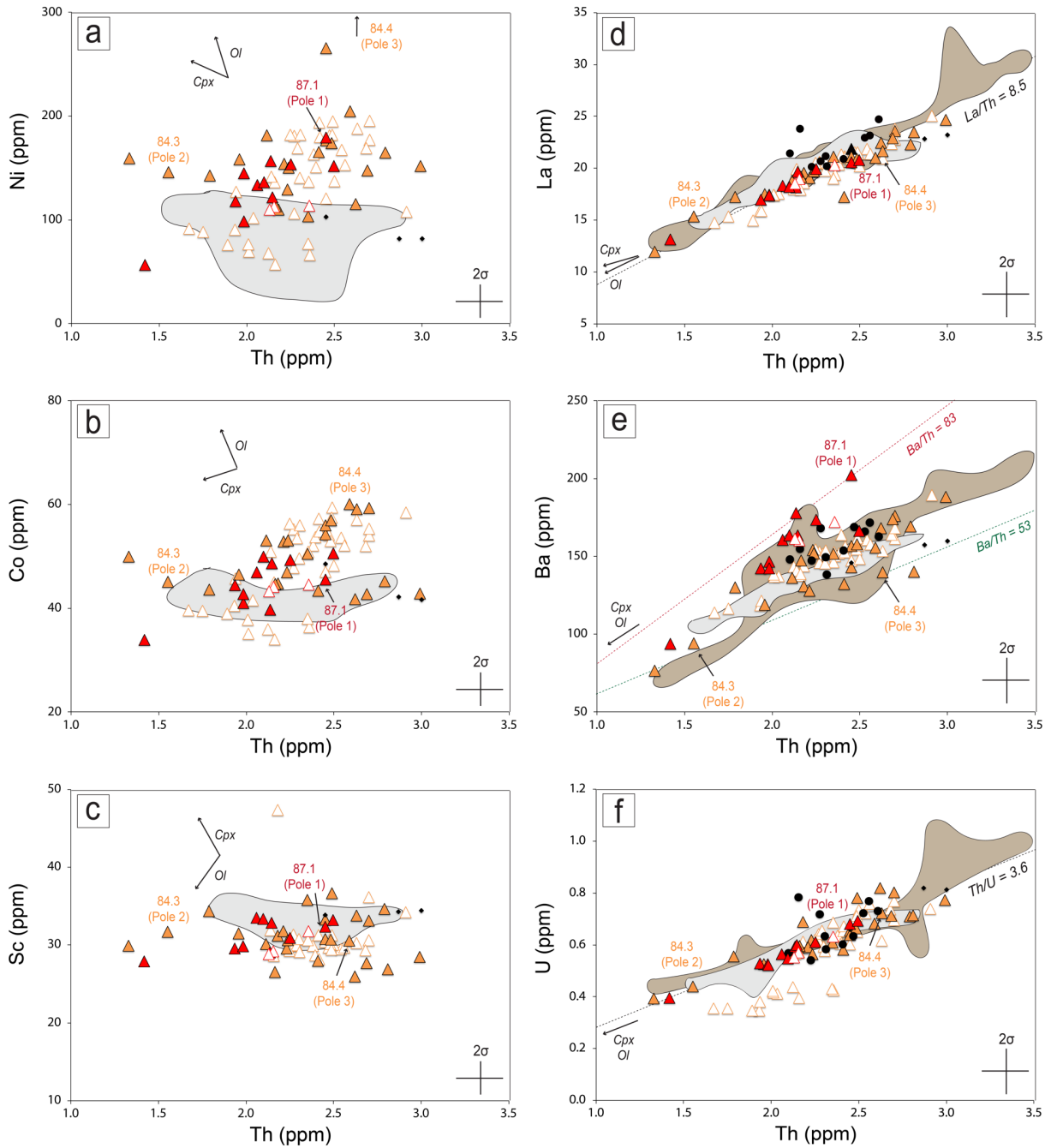
1333 **Fig. 6: Transmitted light microphotographs of fluid and melt inclusions in olivine crystals from basalts**
 1334 **emitted along the NWRZ. (a) Primary fluid inclusion in an olivine crystal from Trous Blancs basalts (TB).**
 1335 **(b) Trail of secondary fluid inclusions in an olivine crystal from Below Petit Cratère (BPC).**
 1336 **Well-rounded primary melt inclusions in olivine from Petit Cratère basalts (PC). A single gas bubble is present.**
 1337 **(d) Well-rounded primary melt inclusions in an olivine crystal from the Plaine des Sables fall (PdS). Many**
 1338 **gas bubbles are present in the melt inclusion. Note the presence of a trail of micro fluid inclusions that could**
 1339 **indicate a potential volatile leakage from melt inclusions.**

1340



1342 **Fig. 7: Composition in major elements of melt inclusions, corrected for PEC, and matrix glasses. (a) MgO**
 1343 **vs. CaO/Al₂O₃ with the distinct groups of lavas usually defined at Piton de la Fournaise (Lénat et al. 2012;**
 1344 **Valer et al. 2017). (b) Alkalinity index AI vs. CaO/Al₂O₃. (c) SiO₂ vs. K₂O. Gray area is the composition**
 1345 **range of melt inclusions and matrix glasses from the 1998-2010 central products. The compositions of**
 1346 **ultramafic enclaves are pointed by the black arrows (D: Dunite; W: Wehrlite; G: Gabbros). Dashed lines**
 1347 **are predicted melt evolution following a wehrlitic differentiation trends. Numbers represent the forsterite**
 1348 **contents of the hosting olivine crystals.**

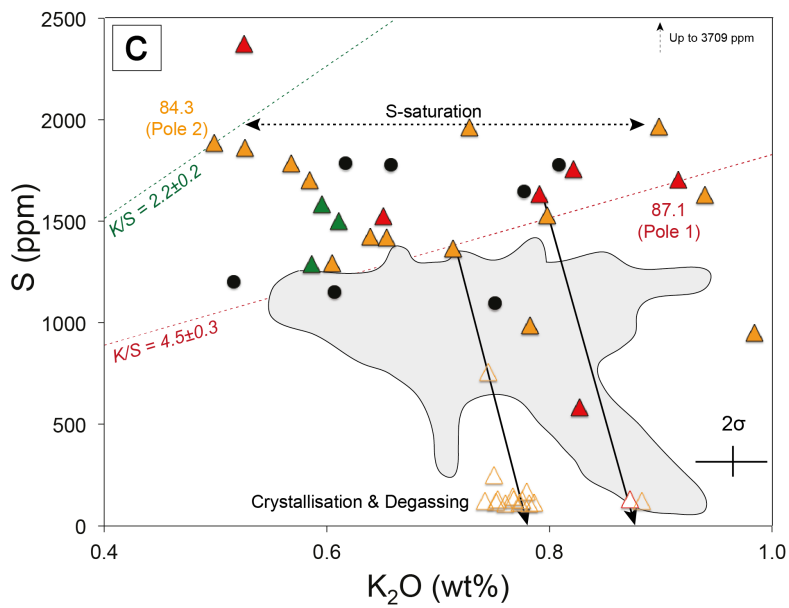
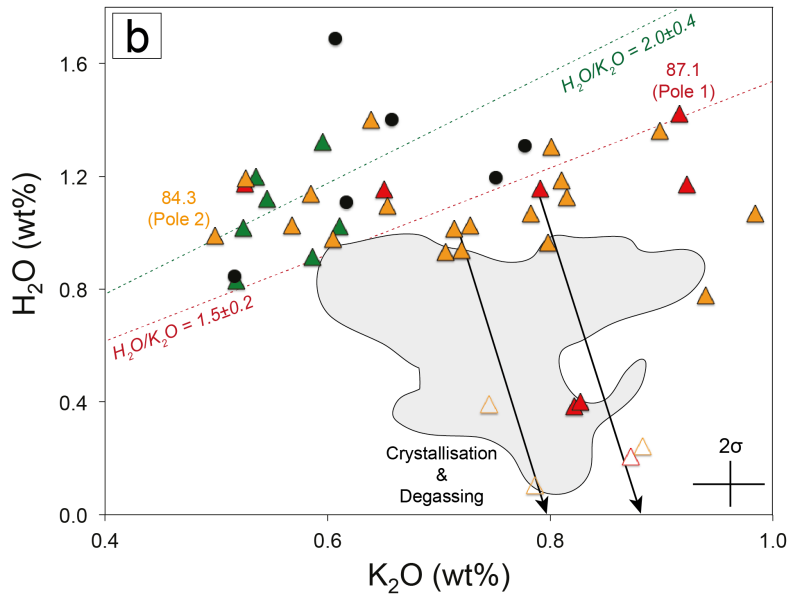
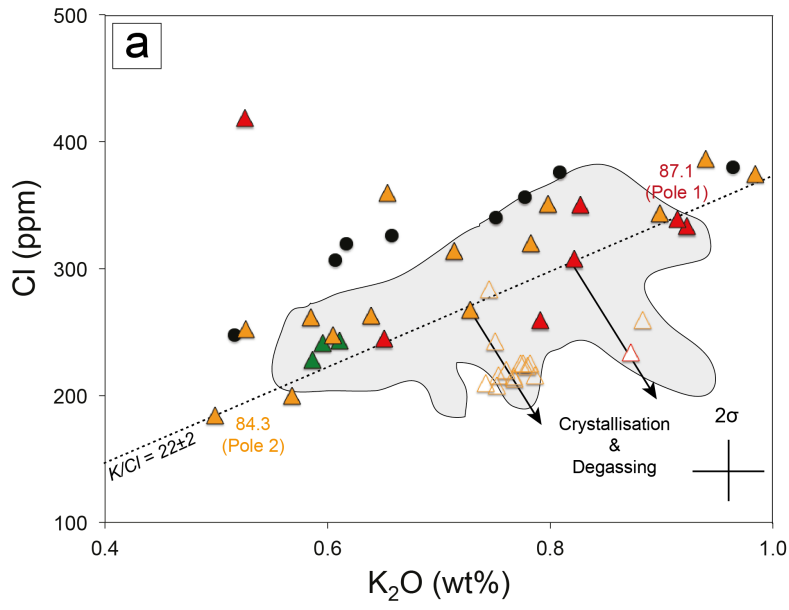
1349



1350

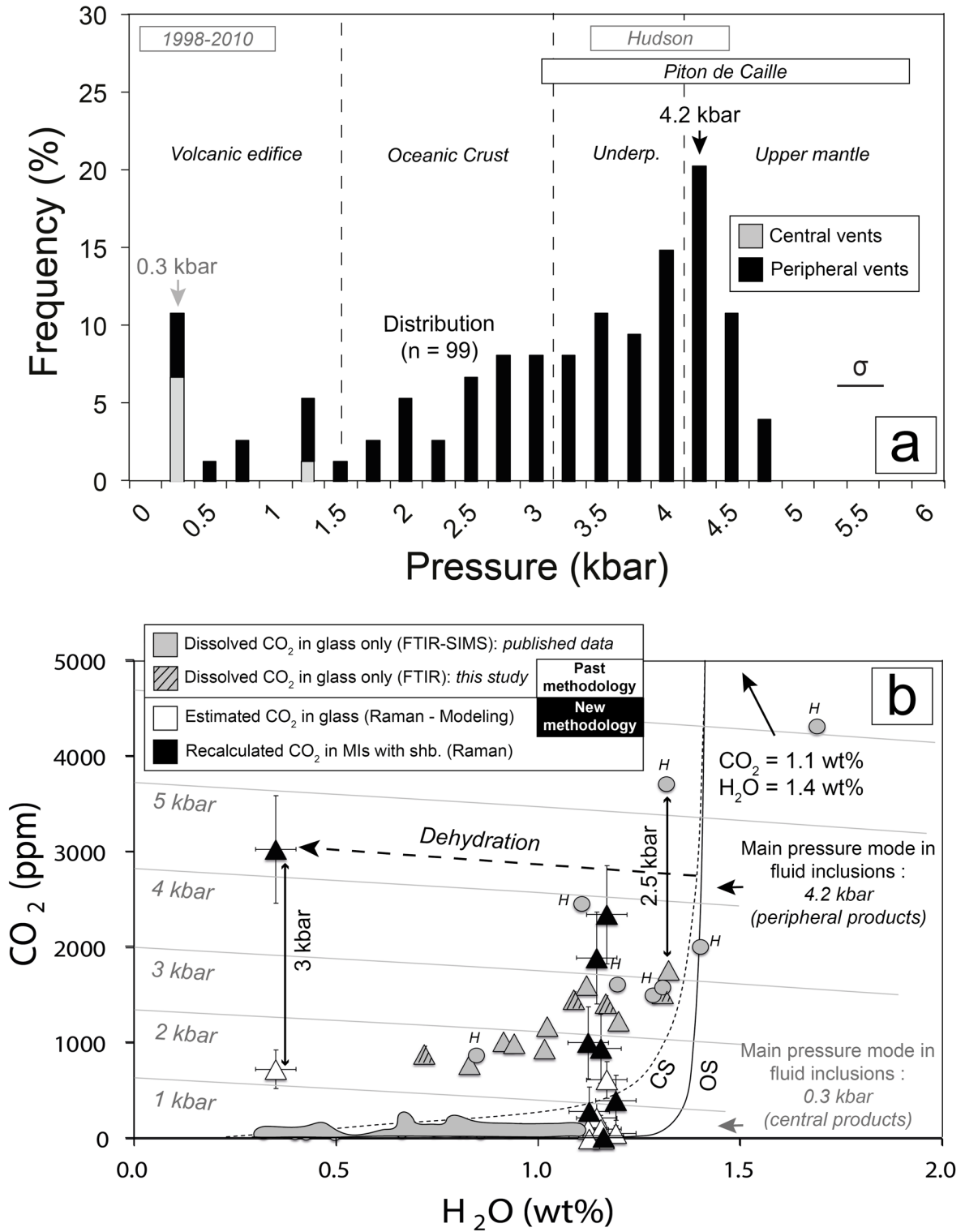
1351 **Fig. 8: Composition in compatible (a, b, c) and incompatible (d, e, f) trace elements from melt inclusions**
1352 **and matrix glasses. Clinopyroxene and olivine composition from the Appendix 1. Intersection of dotted lines**
1353 **at 0. Same symbols and legend as in Fig. 7. Gray areas are melt inclusion compositions in olivine crystals**
1354 **from central products from the major April 2007 caldera forming eruption (Di Muro et al. 2014). Data from**
1355 **peripheral basalts (brown filled area) from Valer (2016) and Valer et al. (2017).**

1356



1358 Fig. 9: Dissolved volatile contents in melt inclusions (recalculated) and matrix glasses. (a) K₂O vs. Cl. (b)
 1359 K₂O vs. H₂O. (c) K₂O vs. S. Gray area is the composition range of melt inclusions and matrix glasses from
 1360 the 1998-2010 central products. Same symbols as in Fig. 7. The “crystallisation & degassing” trends fit melt
 1361 inclusions and related matrix glasses with respect to the differentiation trends defined in Fig. 7.

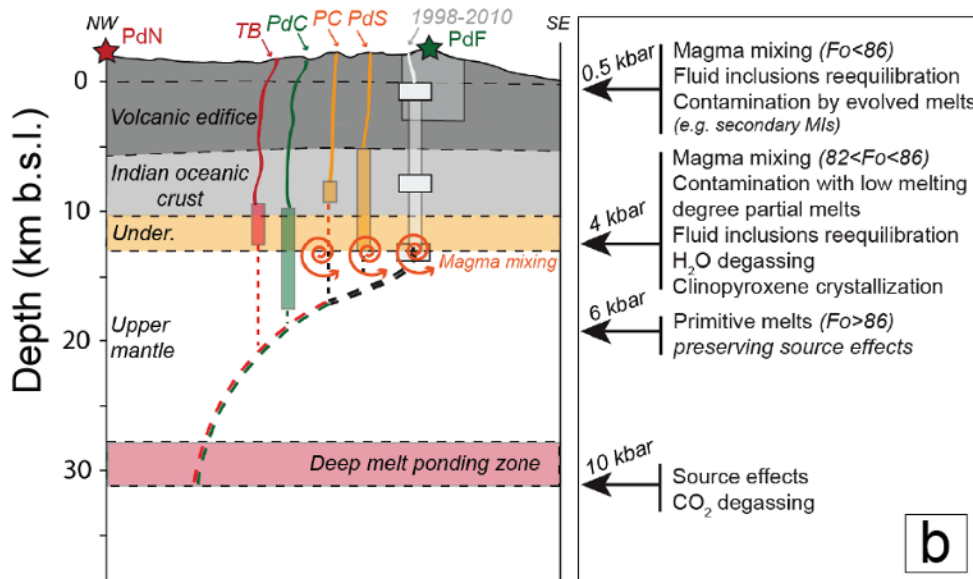
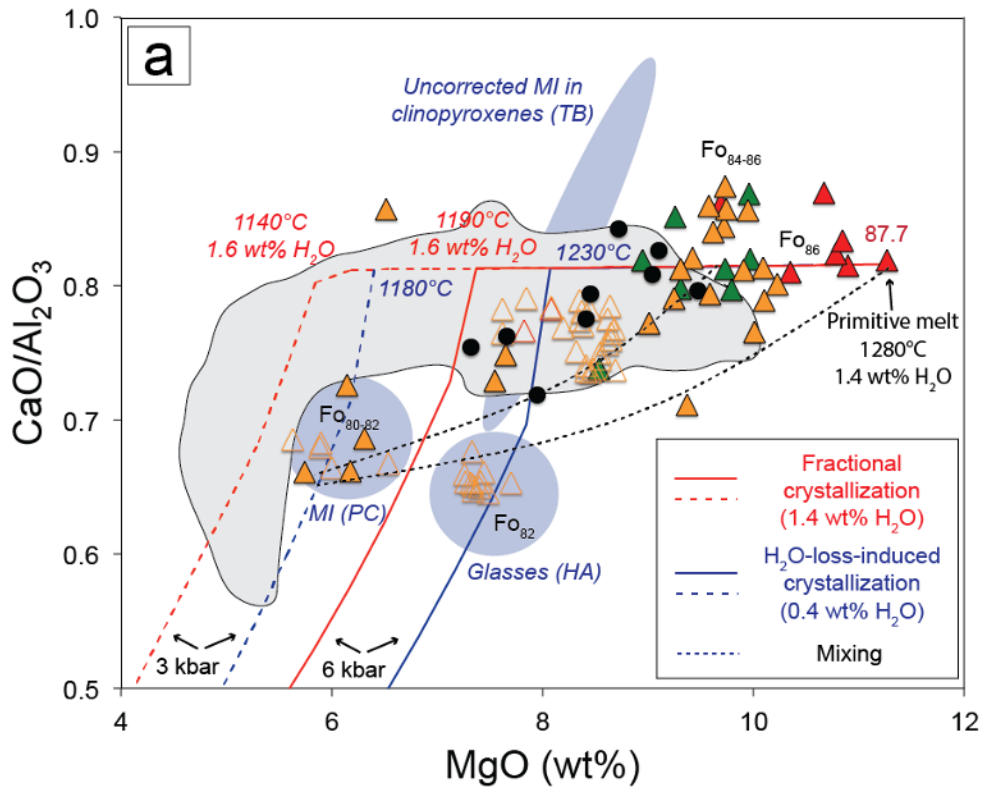
1362



1363

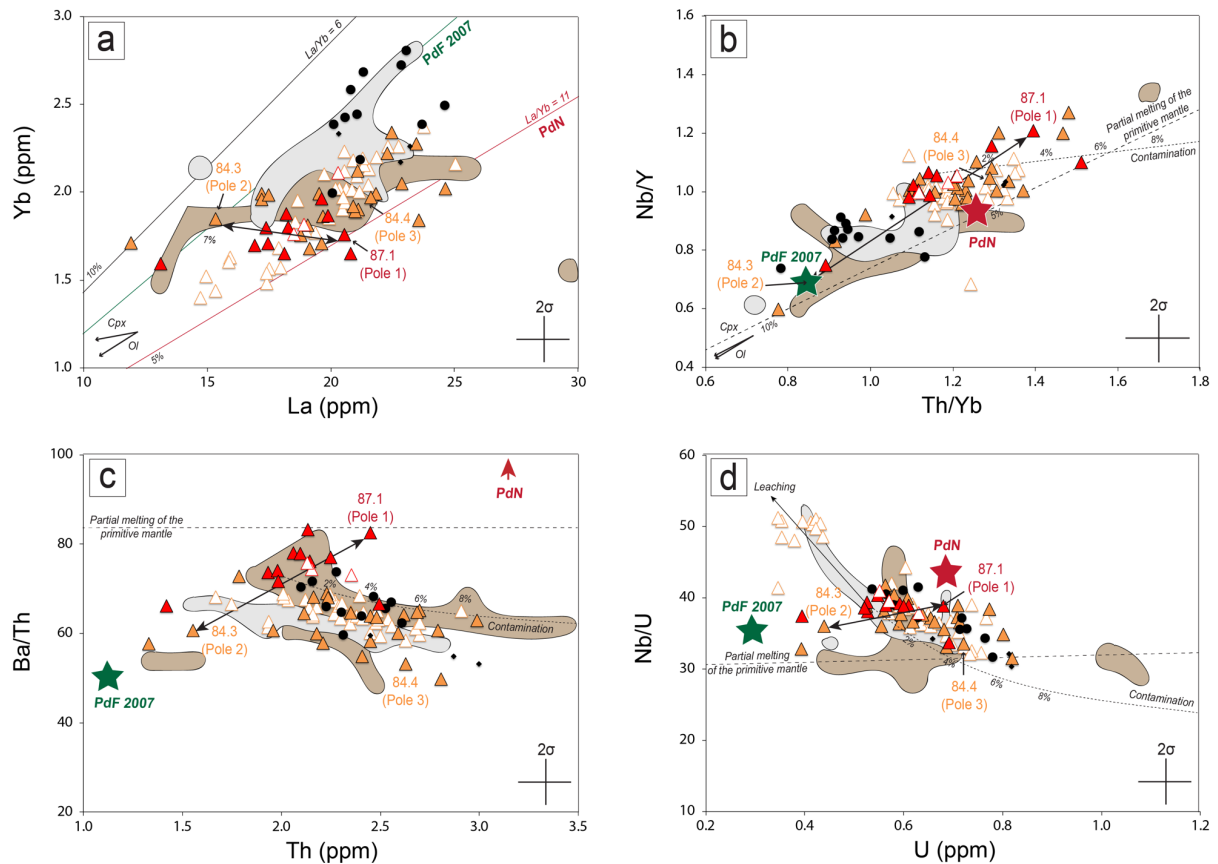
1364 **Fig. 10: (a) Pressure distribution from fluid inclusions barometry from NWRZ (in black) and central (in**
1365 **gray) lavas (46 measurements in this study and 53 analysis from Boudoire et al. (2019)). Piton de la**
1366 **Fournaise interfaces depth as in Boudoire et al. (2019). Underp.: underplating layer at the mantle-crust**
1367 **transition. (b) Comparison between melt inclusion compositions (H₂O and CO₂) analysed and recalculated**
1368 **with different methods (FTIR-SIMS, Raman; see text for explanation and Table 3). Grey area is the**
1369 **composition range of melt inclusions and matrix glasses from the 1998-2010 central products: only melt**
1370 **inclusions in olivine crystals from the central Hudson (H) basalts are enriched in CO₂ (highlighted with grey**
1371 **circles). Melt inclusions in olivine crystals from NWRZ lavas are shown with triangles. Main pressure modes**
1372 **from fluid inclusion barometry in crystals from NWRZ and central basalts are reported (see (a)). Isobars**
1373 **calculated from Duan (2014). Open- (OS) / closed-system (CS; exsolved vapor phase = 1 wt%) degassing**
1374 **paths calculated from Duan (2014). Error bars for each melt inclusion represent the cumulated uncertainty**
1375 **resulting from the methodology of CO₂ recalculation developed in this study (see Methods).**

1376



1377

1378 **Fig. 11: (a) AlphaMELTS simulations of crystallization paths taking into account the potential effect of**
 1379 **water-loss induced by CO₂-flushing. Starting melt corresponds to the most primitive composition found in**
 1380 **the melt inclusions described in this study (inclusion Fo_{87.7}; TB). Oxygen fugacity is fixed at NNO-0.5 in**
 1381 **accordance with recent review of Piton de la Fournaise melts (Pichavant et al. 2016). Starting temperature**
 1382 **was computed from the hydrous melt composition by the equation of Putirka (2008). Same symbols as in**
 1383 **Fig. 7. (b) Conceptual model of the deep plumbing system of Piton de la Fournaise showing magma evolution**
 1384 **below the NW rift zone (modified from Boudoire et al. (2019); see Boudoire et al. (2019) for explanation**
 1385 **about the relation between pressure and depth beneath La Réunion island).**



1387

1388 **Fig. 12: Concentration and ratios in variably incompatible trace elements in melt inclusions and matrix**
 1389 **glasses. Same symbols and legend as in Fig. 7 and 8. The clinopyroxene and olivine composition are**
 1390 **indicated by the black arrows (Appendix 1). Continuous green line (intersection at 0) and star fit the bulk**
 1391 **rock composition of the low radiogenic Sr basalt of the 5th April 2007 (Di Muro et al. 2014). Continuous red**
 1392 **line (intersection at 0) and star fit the average composition of the late-stage lavas of the currently dormant**
 1393 **Piton des Neiges volcano (Younger Lavas <350 kyrs; Smietana 2011). Black dashed line represents the**
 1394 **expected variability of melt compositions related to various degrees of partial melting (italic numbers)**
 1395 **applying a modal batch melting of a garnet peridotite solid source (see Valer et al. (2017) for explanations:**
 1396 **starting conditions from Wang et al. (2018)). Black arrows highlight the effect of source heterogeneities on**
 1397 **the variability in trace elements. The black dotted line represents the expected variability of melt**
 1398 **compositions related to assimilation of 1% of a partial melt from wehrlite cumulate (see text for**
 1399 **explanations; initial conditions: ‘94.G1’ cumulate from Upton et al. (2000); coefficients of partition between**
 1400 **melt-olivine-clinopyroxene at 1 GPa and 1200 °C from Adam and Green (2006)).**

1401

1402 **TABLES**

1403

Sites	Age (BP)	Sample type	Alkalinity Index (AI)	CaO/Al ₂ O ₃	Mg#	⁸⁷ Sr/ ⁸⁶ Sr	Analysis (This study)	Analysis (Literature)
<i>Peripheral eruptions : primitive transitional basalts</i>								
Trous Blancs: TB	> 9430	Lapilli	+0.4 ± 0.1 (n = 2)	0.74	0.65	0.704260±0.000012	Melt inclusions - Fluid inclusions - Mineralogy - Zoning	Bulk rock (Walther et al., 2015; Boudoire et al., 2019)
Piton de Caille: PdC	< 4175	Free minerals in fallout bed	-0.3 (n = 1)	0.79	0.72	0.704151	Zoning	Bulk rock (Valer, 2016) - Fluid inclusions/Mineralogy/Melt Inclusions (Bureau et al., 1998)
<i>Peripheral eruptions : more evolved transitional basalts</i>								
Bélow Petit Cratère: BPC *	< 4175	Lapilli	+0.1 ± 0.2 (n = 11)	0.77 ± 0.02	0.60 ± 0.02	-	Melt inclusions - Fluid inclusions - Mineralogy - Zoning	-
Plaine des Sables fall: Pds	≈ 380 (Last stage)	Lapilli	+0.9 ± 0.1 (n = 8)	0.68 ± 0.01	0.61 ± 0.01	0.704272	Melt inclusions	Bulk rock/Fluid inclusions/Mineralogy (Boudoire et al., 2019)
Piton Haüy: HA	< 4175 (Early stage)	Lapilli	+0.9 (n = 1)	0.68	0.62	0.704291	Melt inclusions	Bulk rock/Fluid inclusions/Mineralogy (Boudoire et al., 2019)
Petit Cratère: PC	470	Bomb	-0.3 (n = 1)	0.76	0.60	0.704133	Melt inclusions - Zoning	Bulk rock /Fluid inclusions/Mineralogy (Boudoire et al., 2019)
<i>Central eruptions : transitional basalts</i>								
998-2010 "classical" eruptions: 98-1 1998-2010 AD Lava - Lapilli -0.4 ± 0.2 (n = 146) 0.79 ± 0.02 0.62 ± 0.10 0.704162±0.000022 - Melt inclusions (Bureau et al., 1999; Vigouroux et al., 2009; Villemant et al., 2009; Di Muro et al., 2014) - Bulk rock/Mineralogy (Boudoire et al., 2019; for a review) - Fluid inclusions (Famin et al., 2005; Boudoire et al., 2019)								
March 1998 (Hudson): H	1998 AD	Lava - Lapilli	+0.2 ± 0.1 (n = 7)	0.71 ± 0.01	0.61 ± 0.01	0.704248	-	Melt inclusions (Bureau et al., 1999; Valer et al., 2017) - Mineralogy/Fluid inclusions (Bureau et al., 1999)
<i>Enclaves</i>								
Dumite (Piton Chisny): D	-	Granular enclaves	+0.4 ± 0.2 (n = 30)	1.08 ± 0.62	0.86 ± 0.04	-	-	Bulk rocks (Upton et al., 2000; Peters et al., 2016) - Fluid inclusions/Mineralogy (Boudoire et al., 2019)
Wehrüte (Piton Chisny): W	-	Granular enclaves	-1.2 ± 0.9 (n = 3)	4.09 ± 0.26	0.86 ± 0.02	-	-	Bulk rocks (Upton et al., 2000; Peters et al., 2016) - Fluid inclusions/Mineralogy (Boudoire et al., 2019)
Gabbros (Bellescombe: Be): G	-	Granular enclaves	-0.7 ± 0.4 (n = 10)	1.15 ± 0.33	0.84 ± 0.03	-	-	Bulk rocks (Upton et al., 2000)

1406 **Table 1: Main characteristics (eruption age, sample type, geochemical markers) of products studied in this**
 1407 **research. Fonts in regular for basalts with an alkaline affinity and in italic for basalts with a tholeiitic**
 1408 **affinity. Geochemical markers (Alkalinity Index = $K_2O + Na_2O - 0.37 \times (SiO_2 - 39)$, Mg#, CaO/Al_2O_3) were**
 1409 **determined for bulk rocks (Valer 2016; Valer et al. 2017; Boudoire et al. 2019), except for BPC (*) for which**
 1410 **only glassy groundmass has been analysed. The description of analyses performed in this study and sources**
 1411 **of complementary published data are reported.**
 1412

	Number of Analysis	FI density ($g.cm^{-3}$)		Pressure (kbar)	
		d_{min}	d_{max}	P_{min}	P_{max}
Peripheral eruptions : primitive transitional basalts	5	0.66	0.77	3.3	4.4
Trous Blancs (TB)* <i>Piton de Caille (PdC)</i>				3.1	5.7
Peripheral eruptions : more evolved transitional basalts	41	0.06	0.79	0.4	4.6
Below Petit Cratère (BPC)* Piton Chisny (CHI)				3.8	4.3
Piton Haüy (HA)				4.2	4.7
<i>Petit Cratère (PC)</i>				3.3	3.3
Central eruptions : transitional basalts				0	0.9
<i>Classical eruptions (98-10)</i> March 1998 (H)				3.5	4.4

1414 **Table 2: Range of CO₂ densities and pressure calculation from fluid inclusions. * for new data documented**
 1415 **in this study. The uncertainty on the estimated pressures using the Fermi dyad (Raman spectroscopy) is**
 1416 **lower to 0.4 kbar (see Methods).**

1417

	Peripheral eruptions											
	Primitive transitional basalts with an alkaline-affinity						More evolved transitional basalts					
	Trous Blancs (TB)			Below Petit Cratère (BPC)			BPC-2			BPC-3		
Shrinkage Bubble Volume	Sites											Piton Chirny (CHI)
	Label											PDS2-126
	T _{melt inclusion} (°C)											
	T _{glass} (°C)											
	Calculated Volume Shrinkage Bubble (%)											
	Estimated Volume Shrinkage Bubble (%)											
	CO ₂ density in Shrinkage Bubble (g.cm ⁻³)											
CO₂ exsolved in Shrinkage Bubble	Glass density (g.cm ⁻³)											
	CO ₂ in Shrinkage Bubble (ppm)											
	CO ₂ uncertainty (ppm)											
Estimated CO₂ dissolved in Glass	Pressure of Exsolution (bar)											
	H ₂ O in glass (wt%)											
	CO ₂ in glass (ppm)											
	CO ₂ uncertainty (ppm)											
Initial CO₂ in Melt Inclusion	CO ₂ content (ppm)											
	Total uncertainty (ppm)											
	TB047-4-M11	TB047-5-M11	TB047-5-M12	TB047-7-M11	TB047-7-M12	TB047-9-M12	PCR1 ec.51a	BPC-1	BPC-2	BPC-3	BPC-4	PDS2-126
	1207	1319	1319	1291	1280	1281	1250	1217	1217	1213	1237	1237
	1204	1204	1204	1204	1204	1204	1217	1217	1217	1213	1237	1237
	0.0	1.9	1.9	1.4	1.2	1.2	0.5	0.5	0.5	0.4	0.4	0.4
	0.07	0.24	0.33	0.17	0.18	0.36	0.17	0.17	0.17	0.10	0.10	0.10
	2.66	2.72	2.72	2.67	2.66	2.63	2.66	2.66	2.66	2.67	2.67	2.67
	13	1675	2303	907	840	1729	343	343	343	283	283	283
	86	458	526	349	333	478	229	229	229	224	224	224
	200	850	1250	560	590	1360	550	550	550	300	300	300
	1.1	1.0	0.4	1.1	1.2	1.2	1.2	1.2	1.3	1.2	0.7	1.1
	0	217	726	99	103	616	57	57	1518	1416	884	0
	130	152	203	140	140	192	136	136	152	142	88	130
	13	1892	3029	1006	943	2345	400	400	1518	1416	884	283
	156	482	564	376	361	515	266	266	152	142	88	259

1418

1419 **Table 3: Reconstruction of the initially dissolved CO₂ content in melt inclusion taking into account single**
 1420 **shrinkage gas bubbles (see Methods for details about the protocol). Font in italic, measurements performed**
 1421 **by FTIR.**

1422

Major element (wt%)	Alkaline affinity (TB)				Tholeiitic affinity (PdC)		(BPC - PC - HA - PdS)		March 1998 (H)		1998-2010 (88-10)		
	Average (n = 5)	σ	87.1 (Pole1)	86.2	87.7	Average (n = 12)	σ	Average (n = 24)	σ	84.3 (HA (Pole 2))	82.2 (PC)	Average (n = 116)	σ
SiO ₂	48.1	0.6	47.7	48.0	49.2	50.1	0.9	48.6	0.8	49.1	51.5	49.5	1.1
K ₂ O	0.8	0.1	0.9	0.8	0.7	0.6	0.2	0.7	0.1	0.5	0.6	0.8	0.2
Na ₂ O	2.5	0.2	2.8	2.5	2.5	2.4	0.1	2.4	0.2	2.6	3.0	2.6	0.2
TiO ₂	2.6	0.1	2.6	2.5	2.4	2.7	0.2	2.7	0.2	2.4	2.5	2.7	0.2
MgO	10.7	0.6	10.9	10.9	11.3	9.4	0.5	8.8	1.5	9.7	6.1	7.9	1.0
FeO _T	11.1	0.8	10.6	11.4	10.2	10.8	0.8	11.4	0.9	11.6	9.1	9.6	1.1
CaO	10.9	0.3	10.8	10.7	10.5	10.2	0.2	10.9	0.5	10.9	11.3	11.2	0.5
Al ₂ O ₃	12.9	0.3	13.3	12.8	12.8	13.3	0.6	13.9	1.1	12.7	15.5	14.3	0.7
P ₂ O ₅	0.3	0.0	0.3	0.3	0.3	0.2	0.1	0.3	0.1	0.3	0.2	0.8	0.2
Volatiles													
Cl (ppm)	299	46	338	259	244	249	42	306	77	184	247	269	50
S (ppm)	1842	1142	1706	1636	1525	1333	318	1556	322	1889	1297	1032	237
H ₂ O (wt%)	0.9	0.4	1.4	1.2	1.2	1.0	0.2	1.1	0.2	1.0	1.0	0.8	0.2
CO ₂ (ppm)	1671	1081	nd	1188	nd	1188	337	1024	519	nd	nd	95	39
Ratios													
CaO/Al ₂ O ₃	0.84	0.02	0.82	0.83	0.82	0.81	0.04	0.79	0.06	0.86	0.73	0.77	0.04
Alkalinity Index	0.15	0.39	0.47	0.09	-0.65	-1.10	0.22	-0.18	0.35	-0.59	-1.02	-0.56	0.36
K ₂ O/TiO ₂	0.33	0.03	0.35	0.31	0.27	0.24	0.11	0.28	0.04	0.21	0.25	0.28	0.06
K ₂ O/P ₂ O ₅	2.73	0.11	2.69	2.64	2.34	2.17	0.21	2.40	0.57	1.74	2.90	2.01	0.33
Cl/K ₂ O	0.039	0.005	0.037	0.033	0.021	0.038	0.004	0.042	0.005	0.037	0.041	0.037	0.007
Pressure													
Melt Inclusions	2.7	1.4	nd	2.1	nd	3.0	0.5	2.1	0.9	nd	nd	0.4	0.1
Fluid Inclusions	3.5	0.4	nd	nd	nd	4.5	nd	2.7	1.1	nd	nd	<1.5	nd

1423

1424 **Table 4: Recalculated melt inclusions composition, CO₂-H₂O saturation pressure (Duan, 2014) and**
1425 **elemental ratios. 87.1, 86.2, 87.7, 84.3, and 82.2 individual composition mark geochemical poles described**
1426 **in this study (see text for explanation). Font in bold, CO₂-H₂O saturation pressure calculations in melt**
1427 **inclusions that take into account the CO₂ content trapped only in single gas bubbles hosted in melt inclusions**
1428 **(see Table 3). nd: not determined.**

1429

1430 **APPENDIXES**

1431

1432 **Electronic Appendix 1: Composition (major and trace elements) of minerals analysed in this study.**

1433

1434 **Electronic Appendix 2: Fluid inclusions densimetry (see Methods for explanations). nd: not determined.**

1435

1436 **Electronic Appendix 3: Composition (major, trace and volatiles) of melt inclusions and glassy groundmass**
1437 **analysed in this study. Blank for not analysed elements and n.d. for measurements not correctly analysed**
1438 **(i.e. for which the content does not exceed 3 times the detection limit).**

1439

1440 **Electronic Appendix 4: Distribution of forsterite contents in olivine crystals from primitive (TB) and more**
1441 **evolved (BPC) basalts.**

1442

1443 **Electronic Appendix 5: Average composition of clinopyroxenes in transitional basalts from NWRZ vents,**
1444 **transitional basalts emitted in the central area, enclaves from La Plaine des Sables and gabbroic clasts from**
1445 **the Bellecombe explosive event (BC). Alkalinity Index defined in Table 1. CaO/Al₂O₃ is a weight percent**
1446 **oxides ratio whereas Ti/Al is an atomic ratio.**

1447

1448 **Electronic Appendix 6: Dissolved volatiles contents in melt inclusions (recalculated) and matrix glasses. (a)**
1449 **FeO_T vs. H₂O. No evidence of post-trapping H₂O-loss is evidenced. (b) S vs. H₂O. Gray area is the**
1450 **composition range of melt inclusions and matrix glasses from the 1998-2010 central products. Same symbols**
1451 **as in Fig. 7. The « degassing » trends fit melt inclusions and related matrix glasses with respect to the**
1452 **« differentiation » trends defined in Fig. 7. (c) Variability of the CaO/Al₂O₃ ratio in melt inclusions as a**
1453 **function of the forsterite content of the hosted olivine. Same symbols as in Fig. 7.**

1454

1455 **Electronic Appendix 7: Comparison between contents of trace elements, ratios of variably incompatible**
1456 **elements, canonical ratios of highly incompatible elements and Sr-Nd isotopy in Piton des Neiges Younger**
1457 **Lavas (Smietana 2011), Piton de la Fournaise 5th April 2007 melt (Di Muro et al. 2014) and our melt**
1458 **inclusions with either an alkaline affinity (F_{087.1}) or a tholeiitic one (F_{084.4}). Sr-Nd isotopy in Piton des Neiges**
1459 **Younger Lavas (Smietana 2011) are consistent with previous findings of Bosch et al. (2008). *Sr-Nd isotopy**

1460 from bulk rock of TB (alkaline affinity; Boudoire et al. 2019) and of PdC (tholeiitic affinity; Valer 2016;
1461 Valer et al. 2017).