

1 An evaluation of the C/N ratio of the mantle from natural CO₂-rich gas analysis:
2 Geochemical and cosmochemical implications

3

4 Bernard Marty^{1*}, Matthieu Almayrac¹, Peter H. Barry², David V. Bekaert^{1,2}, Michael W.
5 Broadley¹, David J. Byrne¹, Christopher J. Ballentine³ and Antonio Caracausi⁴

6

7

8 ¹Université de Lorraine, CNRS, CRPG, F-54000 Nancy, France

9 ²Marine Chemistry and Geochemistry Dept., Woods Hole Oceanographic Institution,

10 Woods Hole, MA 02543, USA

11 ³Department of Earth Sciences, University of Oxford, OX1 3AN Oxford, United Kingdom

12 ⁴Instituto Nazionale di Geofisica e Vulcanologia, Sezione di Palermo, 90146 Palermo, Italy

13

14 *Corresponding author - Email: bernard.marty@univ-lorraine.fr

15

16

17 **Abstract**

18

19 The terrestrial carbon to nitrogen ratio is a key geochemical parameter that can provide
20 information on the nature of Earth's precursors, accretion/differentiation processes of our
21 planet, as well as on the volatile budget of Earth. In principle, this ratio can be determined
22 from the analysis of volatile elements trapped in mantle-derived rocks like mid-ocean ridge
23 basalts (MORB), corrected for fractional degassing during eruption. However, this correction
24 is critical and previous attempts have adopted different approaches which led to contrasting
25 *C/N* estimates for the bulk silicate Earth (BSE) (Marty and Zimmermann, 1999; Bergin et al.,
26 2015). Here we consider the analysis of CO₂-rich gases worldwide for which a mantle origin
27 has been determined using noble gas isotopes in order to evaluate the *C/N* ratio of the mantle
28 source regions. These gases experienced little fractionation due to degassing, as indicated by
29 radiogenic ⁴He/⁴⁰Ar* values (where ⁴He and ⁴⁰Ar* are produced by the decay of U+Th, and
30 ⁴⁰K isotopes, respectively) close to the mantle production/accumulation values. The *C/N* and
31 *C*^β/*He* ratios of gases investigated here are within the range of values previously observed in
32 oceanic basalts. They point to an elevated mantle *C/N* ratio (~350-470, molar) higher than
33 those of potential cosmochemical accretionary endmembers. For example, the BSE *C/N* and
34 ³⁶Ar/*N* ratios (160-220 and 75 x 10⁻⁷, respectively) are higher than those of CM-CI chondrites
35 but within the range of CV-CO groups. This similarity suggests that the Earth accreted from
36 evolved planetary precursors depleted in volatile and moderately volatile elements. Hence the
37 high *C/N* composition of the BSE may be an inherited feature rather than the result of
38 terrestrial differentiation. The *C/N* and ³⁶Ar/*N* ratios of the surface (atmosphere plus crust) and
39 of the mantle cannot be easily linked to any known chondritic composition. However, these
40 compositions are consistent with early sequestration of carbon into the mantle (but not N and
41 noble gases), permitting the establishment of clement temperatures at the surface of our planet.

42 **1. Introduction**

43

44 Carbon and nitrogen are essential ingredients to build a habitable Earth-like planet (Alexander
45 et al., 2012; Marty, 2012; Bergin et al., 2015; Johnson and Goldblatt, 2015; Hirschmann,
46 2018; Grewal et al., 2020). Together with water, these elements were delivered to Earth by
47 accretionary materials mainly originating from the inner solar system, whose remnants can be
48 found in primitive meteorites (chondrites). Stable isotope ratios point to an asteroidal-like
49 origin for terrestrial water, carbon and nitrogen (Marty, 2012; Alexander et al., 2012),
50 whereas other potential cosmochemical sources such as cometary material were likely to be
51 minor (Bekaert et al., 2020). Of particular interest is C, one of the main greenhouse forcing
52 elements in the atmosphere. Its present-day abundance at Earth's surface is reasonably well
53 known from the geological record (27 ± 5 ppmw, normalized to the mass of the BSE, $4.0 \times$
54 10^{27} g; Hirschmann, 2018), but the abundance of carbon in the other major terrestrial
55 reservoirs (e.g., the core and the mantle) is not well-constrained and somewhat model-
56 dependent (Dalou et al., 2017; Hirschmann, 2018; Grewal et al, 2020). For example, C is
57 often advocated to be an important trace element in the core, but quantitative estimates of its
58 concentration depend on several poorly constrained parameters such as pressure,
59 composition-dependent metal-silicate partitioning coefficients and chronology of core
60 formation (e.g., Dalou et al., 2017; Li et al., 2020). As both carbon and nitrogen are
61 incompatible during partial melting under present-day mantle oxygen fugacity conditions,
62 their contents in the mantle-derived lavas can thus, in principle, be used to estimate the
63 composition of the mantle source. However, CO₂, the main carbon-bearing species in present-
64 day magmas, has a low solubility in basaltic melts and lavas reaching Earth's surface are
65 readily degassed, including those erupted at high hydrostatic pressure and rapidly quenched
66 on the ocean floor. Some authors have considered oceanic basalt glasses with minimally-

67 degassed basalts (Hirschmann, 2018) to estimate the ratio of CO₂ to non-volatile elements
68 with known geochemical behaviors such as Ba or Nd (Rosenthal et al., 2015), for which the
69 mantle abundance are known. Assuming that the CO₂/Ba ratio (100±20 ppmw/ppmw;
70 (Hirschmann, 2018) is representative of the mantle sources of analyzed lavas, and taking a Ba
71 concentration of the convective mantle to be 4.0±0.4 ppmw (Palme and O'Neill, 2013), the
72 concentration of C in the mantle was hence estimated at 110±40 ppmw for parent reservoirs
73 (Hirschmann, 2018). Volcanism at the Earth's surface is volumetrically dominated by mid
74 ocean ridge (MOR) magmatism (~16 km³/yr; Le Voyer et al., 2019) which is sourced directly
75 from the convecting depleted MORB mantle (DMM). The DMM carbon concentration can be
76 estimated independently from the total flux of CO₂ from ridges, the magma production rate,
77 and the average MOR melting rate (~12%). MOR CO₂ flux values computed following
78 different approaches yield comparable values: [1.3±0.8] x 10¹² mol/yr from MOR data
79 compilation (Le Voyer et al., 2019), [2.2±0.9] x 10¹² mol/yr (Marty and Tolstikhin, 1998) and
80 [1.6±0.2] x 10¹² mol/yr (Tucker et al., 2018) from calibration to ³He. The corresponding
81 MOR mantle C content would be about 20-30 ppmw (Tucker et al., 2018), significantly lower
82 than the one obtained from the Ba calibration. Current constraints on the C content of the
83 mantle therefore remain uncertain, especially if regions that are not involved in mantle
84 convection (e.g., the plume mantle source) are significant repositories for C, as independently
85 suggested by noble gas isotopes (e.g., Mukhopadhyay and Parai, 2019).

86

87 Nitrogen in the atmosphere-mantle system can be estimated on a global scale. N₂ and
88 radiogenic ⁴⁰Ar abundances correlate in MORB glasses and ocean island basalt (OIB) glasses
89 from different localities worldwide (Marty and Zimmermann, 1999; Marty and Dauphas,
90 2003). Consequently, the N content of the mantle can be estimated from that of ⁴⁰Ar produced
91 by the decay of ⁴⁰K because the potassium content of the BSE is reasonably well estimated

92 (280±60 ppm K, Arevalo et al., 2009). Terrestrial ^{40}Ar has been produced by the radioactive
93 decay of ^{40}K in the BSE, and about half (42^{+32}_{-18} %) of radiogenic ^{40}Ar (noted $^{40}\text{Ar}^*$) is now
94 in the atmosphere as a result of terrestrial degassing (Allègre et al., 1996). The surface
95 reservoirs (atmosphere: 0.97 ppmw N, sediments and crust) can account for about 1.56 ± 0.06
96 ppmw N (Johnson and Goldblatt, 2015; Hirschmann, 2018; after correction in error
97 propagation). Using a mass balance approach, the mantle N concentration was estimated to be
98 around 1 ppmw (Marty, 2012) and more recently to be 1.10 ± 0.55 ppmw (Hirschmann, 2018).
99 Available constraints on the global inventory of N in the Earth's mantle and surface constitute
100 a solid basis for estimating the volatile content of the BSE (Marty, 2012; Johnson and
101 Goldblatt, 2015; Bergin et al., 2015; Hirschmann, 2018).

102

103 By combining the mantle carbon concentration obtained from calibration to Ba together with
104 the above mantle N concentration, Bergin et al. (2015) estimated a mantle *C/N* ratio of ~90 (in
105 the following elemental ratios are molar and written in italics while C and N concentrations
106 are in ppmw of the BSE having a mass of 4×10^{27} g). Using their estimated surface *C/N* ratio
107 (21 ± 6), Bergin et al. proposed a BSE *C/N* ratio of 61 ± 23 . In contrast, Marty (2012) estimated
108 a significantly higher BSE *C/N* ratio of 365 ± 233 from the analysis of CO_2 and N_2 in MORBs
109 and OIBs (Marty and Zimmermann, 1999; Marty and Dauphas, 2003). The implications are
110 not trivial, as a high *C/N* ratio of the BSE would imply a high mantle C concentration and
111 drastic volatile fractionation either in the planetesimals that formed our planet or during
112 Earth's formation.

113

114 The bulk mantle *C/N* estimate of Marty (2012) is based on the C-N-noble gases composition
115 of a suite of 47 MORB and OIB samples extracted following vacuum crushing to remove
116 secondary contributions within basaltic glasses. Gases trapped in oceanic glasses always

117 constitute a residual fraction of the initial volatile content due to pre-eruption and syn-
118 eruption degassing. Hence residual volatiles are elementally fractionated according to their
119 solubilities in basaltic melts. The solubilities of N₂ and Ar are comparable ($K_{N_2} \sim K_{Ar} \sim 2\text{-}3 \times$
120 10^{-12} mol/g.hPa; Jambon et al., 1985; Libourel et al., 2003), and their degassing fractionation
121 is minimal. However, CO₂ and He are ~3 and ~10 times more soluble than N₂ or Ar
122 respectively ($K_{CO_2} = 9 \times 10^{-12}$ mol/g.hPa, Dixon et al., 1995; $K_{He} = 2.5 \times 10^{-11}$ mol/g.hPa,
123 Jambon et al., 1985). Accordingly, He and CO₂ will be depleted (relative to N₂-Ar) in the first
124 fractions of gas escaping from a magma and enriched in residual gas trapped in erupted lavas.
125 One way to correct for degassing fractionation is to use the radiogenic ${}^4\text{He}/{}^{40}\text{Ar}^*$ ratio (where
126 ${}^{40}\text{Ar}^*$ is the amount of ${}^{40}\text{Ar}$ corrected for atmospheric contamination). Its
127 production/accumulation ratio in the mantle is estimated from the parent $(U+Th)/K$
128 abundance ratio to be ~2 (range: 1.8-3.0; e.g., Marty, 2012; Mukhopadhyay and Parai, 2019).
129 Oceanic basalt glasses present systematically higher values in the range 3-96 (Marty and
130 Zimmermann, 1999), demonstrating the effect of elemental fractionation that could be
131 modeled following a Rayleigh-type distillation process. The initial $(C/N)_i$ ratio of the mantle
132 source could be computed from the measured $(C/N)_{meas}$ ratio according to Eqn. 1:

133

$$134 \quad (C/N)_i = (C/N)_m \times \left[\frac{({}^4\text{He}/{}^{40}\text{Ar})_m}{({}^4\text{He}/{}^{40}\text{Ar})_i} \right]^{\left[\frac{(K_{N_2}^{-1} - K_{CO_2}^{-1})}{(K_{He}^{-1} - K_{Ar}^{-1})} \right]} \quad (1)$$

135

136 where K_j are the basaltic melt/gas partition coefficients for species j and suffixes i and m
137 mean initial and measured, respectively. After correction, the mean MORB (C/N) ratio was
138 computed to be 535 ± 224 (Marty and Zimmermann, 1999), with E-MORBs ($(C/N) =$
139 1839 ± 641) being richer in C (relative to N) than N-MORBs ($(C/N) = 273 \pm 106$). This study
140 indicated that (i) the C/N ratio of the MORB mantle is much higher than that of the surface

141 inventory ($(C/N)_{surface} = 21 \pm 6$, computed with abundances given in Hirschmann, 2018), and
142 (ii) the C content of mantle sources is highly variable, possibly due to preferential subduction
143 of C compared to N (Marty and Zimmermann, 1999).
144
145 Bergin et al. (2015) questioned this approach by noting that the extent of correction for
146 largely variable C/N ratios in MORB complicates the reconstruction of the C/N ratio of
147 mantle sources (Bergin et al, 2015). Natural CO₂-rich gases for which isotopic ratios of C, N
148 and noble gases demonstrate a mantle origin may provide an independent means to
149 investigate the C/N ratio of the mantle. Continental CO₂-rich gases tap into large crustal
150 volumes where mantle-derived gases have accumulated over long periods of time (Ballentine
151 and Holland, 2008), making them presumably more representative of the original mantle
152 composition than residual gases left over in oceanic basaltic glasses. Atmospheric and/or
153 crustal contamination can be problematic for natural gases, yet the noble gases along with C
154 and N can provide sensitive tools to assess these effects and identify the original mantle
155 source composition. Here we have selected CO₂-rich gases from different geodynamic
156 environments where these effects can be robustly corrected for, to thoroughly estimate the
157 C/N ratios of the corresponding magma sources and provide new insights into the C/N ratio of
158 the BSE.

159

160 **2. Samples**

161

162 We have selected CO₂-rich gases for which the $^{40}Ar/^{36}Ar$ ratios are markedly higher than the
163 atmospheric value of 298.6. Other noble gas constraints were used as additional sample
164 selection criteria, namely those with (i) mantle-derived helium, rich in primordial ³He, (ii)
165 mantle-derived neon with 3-isotope compositions lying along the MORB or the mantle plume

166 correlation, and (iii) excess ^{129}Xe from the radioactive decay of extinct ^{129}I (as mantle Xe is
167 enriched in ^{129}Xe with respect to atmospheric Xe) (*Table 1*). Samples having all these features
168 are not common and as far as we know are restricted to the list presented in *Table 1*.

169

170 We include CO_2 -rich gases from continental Europe. Bräuer et al.(2013) carried out a
171 systematic survey of CO_2 -rich gases in the Eifel quaternary volcanic province (Germany), and
172 found two locations (Victoriaquelle and Schwefelquelle) with elevated $^{40}\text{Ar}/^{36}\text{Ar}$ ratios in the
173 range 1,500-3,000. Repeated sampling of these springs by the CRPG group identified even
174 higher $^{40}\text{Ar}/^{36}\text{Ar}$ ratios up to 8,300, ^{129}Xe excesses up to 7 % relative to atmospheric Xe and
175 MORB-like neon (Bekaert et al., 2019). Nitrogen isotopologues of these samples indicate
176 undetectable atmospheric N_2 contribution (Labidi et al., 2020). We also include data for a
177 CO_2 -rich gas sampled at Bublak, Eger graben, Czech Republic (Bräuer et al., 2004).

178

179 In 2018, we undertook a survey of geothermal gases in the Yellowstone National Park (USA),
180 which is located at the present-day apex of a high- ^3He mantle plume (Craig et al., 1978).

181 Some of the gases that were collected show $^{40}\text{Ar}/^{36}\text{Ar}$ ratios higher than the atmospheric value
182 (Broadley et al., 2020) and their data are reported here in three groups (*Table 1*): (i) the Mud

183 Volcano group located inside the Yellowstone caldera where the highest $^3\text{He}/^4\text{He}$ values (≥ 14

184 Ra, typical of mantle plume He signature) are observed (Craig et al., 1978); (ii) the Turbid

185 Lake group located at the border of the caldera ($^3\text{He}/^4\text{He} = 5.2$ Ra), and (iii) the Brimstone

186 Basin group outside the caldera ($^3\text{He}/^4\text{He} = 2.8$ Ra). He isotope variations are consistent with

187 dilution of a typical mantle plume signature by radiogenic He from the Precambrian basement.

188 Brimstone Basin samples also have Xe and Ne isotopes consistent with a crustal contribution

189 to plume-like noble gases (Broadley et al., 2020). Nitrogen isotopologues of Yellowstone

190 gases also point to a mantle plume origin for nitrogen, with limited contributions of
191 atmospheric or crustal N (Labidi et al., 2020).
192
193 The Naftia sample from Sicily is a CO₂ well gas in the vicinity of Mount Etna which shows
194 an excess of ¹²⁹Xe and MORB-like Ne (Nakai et al., 1997). At this site, the ³He/⁴He ratio
195 varies between 6 and 7 Ra and correlates with the eruptive activity at Mt. Etna, suggesting
196 that new injections of magma under the volcano periodically contribute fresh mantle gases to
197 this CO₂ reservoir (Caracausi et al., 2003). Volcanic gases were also sampled at Oldoinyo
198 Lengai, an active carbonatitic volcano located on the eastern branch of the East African Rift
199 Valley (Fischer et al., 2009). Atmospheric contamination is low, as indicated by ⁴⁰Ar/³⁶Ar
200 ratios up to 948 (OLD-2 sample), which is among the highest values ever measured for
201 volcanic gases. The ³He/⁴He ratio (6.9 Ra) is in the low range of MORB values, and neon has
202 a MORB-like isotopic composition. Fischer et al (2009) interpreted the C, N and noble gas
203 compositions as reflecting degassing of a new batch of magma from a MORB-like source. A
204 ⁴He/⁴⁰Ar* ratio of 0.4 (Table 2) lower than the mantle range (1.8-3.0) is consistent with this
205 interpretation because less soluble argon in silicate melts tends to degas more readily than
206 more soluble helium during the first stages of magma degassing.
207
208 Gases extracted from two well-studied, gas-rich, basaltic glasses are also included in this
209 compilation for the purposes of comparison. 2IID43 is a MORB glass from the Mid-Atlantic
210 Ridge at 14°N (Javoy and Pineau, 1991; Moreira et al., 1998). This sample is considered to
211 best represent the composition of a MORB gas phase. DICE corresponds to a subglacial glass
212 sampled in Iceland (N64°10'31.9"/W021°02'43.0") which is also a reference sample that is
213 considered representative of unfractionated mantle plume-sourced gas (Harrison et al., 1999;

214 Mukhopadhyay, 2012). Both samples have $^4He/^40Ar^*$ close to, or within the range of
 215 accumulation/production mantle values, indicating limited, if any, degassing fractionation.

216

217 **3. Data handling**

218

219 In order to determine the mantle source compositions, gas data need to be corrected for
 220 atmospheric contribution and fractional degassing of magmas, respectively, which we explain
 221 below.

222

223 *3.1- Correction for atmospheric contamination*

224

225 The correction for atmospheric contamination is negligible for CO₂ given its low atmospheric
 226 concentration (440 ppmv) but can be important for N₂ and Ar which are abundant in air
 227 (78.08 vol % and 0.93 vol%, respectively; *Table 1*). Atmospheric contamination of natural
 228 gases is always critical for non-radiogenic noble gases (e.g., ^{36}Ar) and can overprint the
 229 original composition. This is less of a problem for ^{40}Ar because this isotope is more abundant
 230 in the mantle than primordial ^{36}Ar relative to the atmosphere ($^{40}Ar/^{36}Ar = 298.6$ in air versus
 231 $\geq 10,000-40,000$ in the mantle, see below). Correction for atmospheric contamination is
 232 achieved using measured $^{40}Ar/^{36}Ar$ ratios:

233

$$234 \quad ^{40}Ar_{ca} = \frac{^{40}Ar_{meas}}{1+\alpha} \quad (2)$$

235 with:

$$236 \quad \alpha = \left(\frac{^{36}Ar}{^{40}Ar}\right)_{ca} = \frac{\left(\frac{^{36}Ar}{^{40}Ar}\right)_{meas} - \left(\frac{^{36}Ar}{^{40}Ar}\right)_{atm}}{\left(\frac{^{36}Ar}{^{40}Ar}\right)_{atm} - \left(\frac{^{36}Ar}{^{40}Ar}\right)_{mantle}} \quad (3)$$

237

238 Where suffixes *ca*, *meas*, *atm* and *mantle* refer to $^{40}\text{Ar}/^{36}\text{Ar}$ values corrected for atmospheric
 239 contamination, measured in the sample, in the atmosphere and the mantle end-member,
 240 respectively. For the latter, we chose 40,000 and 10,000 for convective mantle and mantle
 241 plume sources, respectively (Bekaert et al., 2019; Mukhopadhyay, 2012). Taking 27,000 as
 242 the mantle source endmember of 2ΠD43 (Moreira et al., 2018) would have a minor effect on
 243 the correction. The corrections for the ^{40}Ar concentrations are in most cases marginal (within
 244 0.1- 20 %), except for the Mud Volcano and Turbid Lake samples (65-71%) and for the
 245 Bublak gas sample (54%).

246

247 For N_2 which is also less affected by atmospheric contamination than primordial noble gases
 248 ($(\text{N}_2/^{36}\text{Ar}) = 1.5\text{-}3 \times 10^4$ in air and air-saturated water, and $\geq 10^6$ in the mantle), the correction
 249 is achieved by using the $\text{N}_2/^{40}\text{Ar}$ ratio, with corrected N_2 being computed using the corrected
 250 $^{40}\text{Ar}_{\text{CA}}$ concentration:

251

$$252 \quad \left(\frac{\text{N}_2}{^{40}\text{Ar}}\right)_{ca} = \left(\frac{\text{N}_2}{^{40}\text{Ar}}\right)_{meas} + \alpha \times \left[\left(\frac{\text{N}_2}{^{40}\text{Ar}}\right)_{meas} - \left(\frac{\text{N}_2}{^{40}\text{Ar}}\right)_{atm} \right] \quad (4)$$

253

$$254 \quad \text{N}_{2,ca} = \left(\frac{\text{N}_2}{^{40}\text{Ar}}\right)_{ca} \times ^{40}\text{Ar}_{ca} \quad (5)$$

255

256 where α is computed from Eqn. 3. The atmospheric contaminant for geothermal and volcanic
 257 gases is generally carried by underground aquifers (Giggenbach, 1980). For the atmospheric
 258 $(\text{N}_2/^{40}\text{Ar})_{atm}$ ratio, we assume a value 39.2, which is a mean for air-saturated water at
 259 equilibrium temperatures of 10°C (37.9) and 100°C (40.5) (computed with the Bunsen
 260 coefficients given in Ozima and Podosek, 2002).

261

262 3.2- Correction for fractional degassing

263

264 As for oceanic basalts, the correction for fractional degassing is based on the measurement of
265 the $^4\text{He}/^{40}\text{Ar}^*$ ratio using *Eqn. (1)*, where $^{40}\text{Ar}^*$ is approximated by $^{40}\text{Ar}_{\text{ca}}$. The abundance of
266 helium does not need to be corrected for atmospheric contamination given its low
267 atmospheric content (5.24 ppmv) and the generally low level of air contamination for the
268 gases selected here (correction for the He isotopic ratios are also negligible for these samples).
269 The $^4\text{He}/^{40}\text{Ar}^*$ ratios of the gases (including gases trapped in DICE and 2IID43 vesicles)
270 range between 0.4 (Oldoinyo Lengai) and 2.3 (DICE) (*Table 2*). These values are much closer
271 to the mantle production/accumulation range (1.8-3.0) than ratios measured in oceanic basalt
272 glasses (3-96; Marty and Zimmermann, 1999), making correction for fractional degassing
273 limited. Contrary to MORB glasses which are extensively degassed, CO_2 -rich gases represent
274 a gas phase in equilibrium with a degassing batch of magma and may also be related to new
275 batches of rising magmas as in the case of Oldoinyo Lengai (Fischer et al., 2009) and Etna
276 (Caracausi et al., 2003). The C/N , $C^\beta\text{He}$, $N^\beta\text{He}$ and $^{36}\text{Ar}/N$ ratios corrected for atmospheric
277 contamination and fractional degassing are given in *Table 2* and *Table 3*, respectively.

278

279 4- Discussion

280

281 4.1- Distribution of C/N values in CO_2 -rich gases and MORBs

282

283 In *Fig. 1* we compare the distribution of corrected C/N ratios in CO_2 -rich gases and in the two
284 gas-rich glasses (2IID43 and DICE) with that of all corrected MORB data (Marty and
285 Zimmermann, 1999). The distribution of CO_2 -rich gas data shows a mean C/N around 500
286 whereas the distribution of MORB values is shifted towards low N-MORB values, which

287 were more numerous than carbon-rich E-MORB values in the compilation of Marty and
288 Zimmermann (1999). Hence data from natural gases confirm that the respective magma
289 sources have C/N ratios higher than Solar, Chondritic (CI and CM, Fig. 5) and the mantle
290 value previously proposed by Bergin et al (2015).

291

292 *4.2- Crustal or subcontinental lithospheric mantle (SCLM) contribution?*

293

294 Although gas compositions reported here indicate that C/N and $C^{\beta}He$ ratios are comparable
295 to those of the mantle sources of oceanic basalts (*Figs. 1 & 2; Tables 2 & 3*), the underlying
296 crust and/or subcontinental lithospheric mantle may also significantly contribute to the budget
297 of outgassing volatiles. We note that the isotopic compositions of carbon and nitrogen are
298 consistent with a mantle origin ($\delta^{13}C = -5 \pm 2\text{‰}$; $\delta^{15}N = -5 \pm 2\text{‰}$ for MORBs and diamonds,
299 $\delta^{15}N = 0 \pm 3\text{‰}$ for mantle plumes; e.g., Javoy and Pineau, 1991; Marty and Zimmermann, 1998;
300 Marty and Dauphas, 2003; Cartigny et al, 2014) for these gases (*Table 1*), although some of
301 the values are intermediate between those of carbonates ($\delta^{13}C = 0$) and of the atmosphere
302 ($\delta^{15}N = 0\text{‰}$), respectively, suggesting contributions of surface and/or crustal volatiles. The
303 $^3He/^4He$ ratios of some of the CO_2 -rich gases considered here (3.2 Ra for New Mexico gases,
304 4.4-5.9 Ra for continental Europe gases, 2.4 Ra for Yellowstone gases sampled outside the
305 caldera) are lower than the canonical MORB range of 8 ± 1 Ra, and samples from inside the
306 Yellowstone caldera. This is also consistent with contributions of radiogenic 4He from the
307 regional crust.

308

309 Here we investigate crustal C (and by extension N) contribution to mantle-derived gases by
310 using the $C^{\beta}He$ ratios, which constitute a good indicator of C provenance since the crust is
311 highly depleted in primordial 3He relative to the mantle. In a $C^{\beta}He$ versus C/N diagram (*Fig.*

312 2), several of the samples plot close to the field defined by MORBs and are intermediate
313 between the mean N-MORB and the carbon-rich E-MORB endmembers. However, some of
314 the data points also seem to depart from the mixing trend between these two MORB
315 endmembers. This tendency is best exemplified by the Eger Rift sample (*Fig. 2a*), but it could
316 also be the case for the New Mexico and Eifel samples. For the Eifel samples, Labidi et al
317 (2020) argued from N isotope systematics that crustal contamination of the gas is unlikely.
318 However, these authors did not exclude the contribution of ancient subducted crust to the
319 subcontinental mantle in Western Europe. For the Yellowstone gases, Werner and Brantley
320 (2003) estimated that ~50% of CO₂ is of sedimentary origin for the emanations outside the
321 caldera and that more than ~70% CO₂ is magmatic inside the caldera. In line with this view,
322 we note that the CO₂^βHe ratios at Mud Volcano in the center of the caldera are lower (2.5 x
323 10⁹) than those of the Brimstone basin gases (4 x 10⁹) located outside the caldera, despite
324 having comparable δ³C values (*Table 1*). Labidi et al (2020) concluded from N isotopologues
325 that the contribution of crustal nitrogen to the Yellowstone gases inside the caldera is
326 negligible. Overall the C^βHe ratio of the Yellowstone gases inside the caldera (2.7 x 10⁹) is
327 comparable to the Icelandic DICE ratio (2.8 x 10⁹) (*Table 3*), where crustal contamination is
328 minimal.

329

330 In principle it should be possible to correct C for crustal contribution using *the* CO₂^βHe ratios,
331 but this is complicated by the fact that E-type MORB glasses or plume samples also have
332 elevated CO₂^βHe ratios (up to 6 x 10⁹) and high CO₂^βHe ratios could be due to mantle source
333 heterogeneities and not crustal contamination of magmas or related gases. In a C^βHe versus
334 C/N diagram plot (*Fig. 2b*), contributions of pure crustal C (without crustal N) will produce
335 correlations anchored at the mantle end-member values and extending toward high C^βHe and
336 C/N ratios. Each data point is corrected for crustal C contribution by extrapolating measured

337 data towards zero and taking the intersection of these lines with the $CO_2^{\beta}He$ ratio inferred for
338 the mantle ($2.2\pm 0.6 \times 10^9$, Marty and Zimmermann, 1999 or $1.7\pm 0.2 \times 10^9$, Tucker et al.,
339 2018; *Fig. 2b*). With these two $CO_2^{\beta}He$ values, the mantle source C/N ratios are constrained
340 to be within 100-530 (*Fig. 2b*).

341
342 There is however two caveats with this approach. Firstly, the dispersion of data can be
343 accounted for by preferential loss of 3He while leaving unaffected C and N (near-constant
344 C/N ratio), as suggested by one reviewer. In this case, the observed C/N ratios should be
345 representative of the mantle source value(s). Secondly, the correction for crustal C addition
346 assumes that only C, and not N, is contributed by the crust. To check this assumption, the
347 $C^{\beta}He$ ratios are plotted as a function of the $N^{\beta}He$ ratios in Figure 3 (data from *Table 2*). Data
348 seem to define two domains that both originate from the N-MORB composition, one
349 encompassing the E-MORB endmember and the other one pointing toward high $N^{\beta}He$ values
350 and encompasses most of continental gas data. This correlation corresponds to a C/N ratio
351 around 400-500 (*Fig. 3*). However, a typical crustal C/N ratio would be ≤ 70 (with $C = 27\pm 5$
352 ppmw, Hirschmann, 2018, and $N = 0.50\pm 0.04$ ppmw, Johnson and Goldblatt, 2015) and
353 cannot therefore account for the elevated C/N ratio of the CO_2 -rich gases.

354
355 The sub-continental lithospheric mantle ($C/N = 390\pm 260$, computed with $C = 100\pm 20$ ppmw,
356 Hirschmann, 2018, and $N = 0.3\pm 0.2$ ppmw, Johnson and Goldblatt, 2015) could be a possible
357 endmember for some of the continental gases (*Fig. 3*), but other geochemical and geophysical
358 evidence support an asthenospheric, rather than lithospheric, origin for Cenozoic magmatism
359 in these regions. Evidence include: (i) trace element and isotope systematics of erupted lavas
360 (Hoernle et al., 1995); (ii) geochemical affinities with HIMU-type magmatism for Eifel
361 (Bekaert et al, 2019, and refs. therein, Sicily (Mt. Etna; Caracausi et al., 2003, and refs.

362 therein), Tanzania (Oldoinyo Lengai; Fischer et al., 2009, and refs. therein) and New Mexico
363 (Fitton et al., 1988), and (iii) geophysical imaging showing upwelling of asthenospheric
364 material below European magmatic provinces (Hoernle et al., 1995). Thus, the C-N-³He
365 signatures of related CO₂-rich gases are likely to reflect the composition of asthenospheric
366 sources different from those feeding mid-ocean ridges. Enrichments in C and N compared to
367 typical MORBs (*Fig. 3*) may be due to primordial heterogeneities, or the recycling of surface
368 C and N deep into the mantle (Busigny and Bebout, 2013), in line with the HIMU character of
369 some of the corresponding magmatic provinces. In support of this possibility, Labidi et al.
370 (2020) argued from N isotopologues that about 30% of nitrogen in the Eifel CO₂-rich gases
371 could be recycled in origin.

372

373 Contribution of C-rich material into the mantle sources of E-MORBs and of intraplate
374 volcanoes could potentially account for the high C/N component suggested by trends
375 displayed in *Fig. 3*. It is not clear however if the high C/N ratio of the Icelandic DICE sample
376 and of the (inside caldera) Yellowstone samples have a similar recycled origin. Labidi et al.
377 (2020) proposed that nitrogen in Yellowstone gases defines a primordial mantle plume
378 signature having a $\delta^{15}N$ value close to 0‰ (+3±2 ‰) and a N/³⁶Ar ratio of 2.4-3.9 x 10⁶.
379 Marty and Dauphas (2003) also noted that plume-related gases have $\delta^{15}N$ values close to 0‰
380 (-1.5 to 3.3 ‰ for DICE) but these signatures were instead attributed to recycling of surface N
381 into the source of mantle plumes (Marty and Dauphas, 2003; Halldorson et al, 2016). We note
382 that the C/N ratios are different between Yellowstone (400-500) and Icelandic (1200-1400)
383 samples (*Table 2*), suggesting that the C/N ratio of the deep mantle is heterogeneous as a
384 possible result of different mixings between primordial and recycled nitrogen.

385

386 *4.3- The C/N ratio of the mantle*

387

388 We attempted to estimate the C/N ratio of the BSE using a mass balance involving nitrogen
389 and mantle C/N ratios (Supplementary Material). As in previous work, this approach is based
390 on (i) the bulk K content of the BSE (Arevalo et al, 2009) and K-Ar-N systematics of MORB
391 and OIB-related samples (Marty, 1995; Marty and Dauphas, 2003; Johnson and Goldblatt,
392 2015; Bergin et al, 2015; Hirschmann, 2018), which are used to estimate the BSE N content,
393 and (ii) the C/N ratios measured in MORB and OIB basalt glasses (Marty, 1995; Marty and
394 Dauphas, 2003) and in mantle-derived CO_2 -rich gases (this study). In doing so, we assume
395 that the C/N ratios measured in surface basalts are representative of the mantle source
396 composition, after correction for fractional degassing and atmospheric contamination (see
397 Introduction). Because we selected samples from both the convective mantle (MORB glasses,
398 CO_2 -rich gases with MORB-like noble gas composition, e.g., Ne isotopes) and from OIB-like
399 mantle domains including high- $^3He/^4He$ plumes (Iceland, Yellowstone), we further posit that
400 these samples are representative of not only the convective mantle responsible for feeding
401 mid-ocean ridges, but also encompass mantle plume domains. However, this approach relies
402 on the assumption that the C/N ratio measured in mantle-derived basalts and gases is
403 representative of the whole mantle composition from its shallowest regions down to the core-
404 mantle interface, which is not guaranteed. There are some suggestions from diamond studies
405 that the C/N ratio of mantle fluids from which diamonds formed may vary with depth,
406 although such variations could result from processing during diamond formation rather than
407 from compositional variation of the mantle itself (e.g., Mikhail and Howell, 2016, and refs.
408 therein).

409

410 With these caveats in mind, we considered two data sets; the previously published MORB
411 data (Marty and Zimmermann, 1999), and the volcanic/geothermal gas data presented in this

412 study. In order to estimate the mantle and BSE C/N ratios, we carried out Monte Carlo
413 simulations following three approaches as detailed in Supplementary Material. We also
414 estimated the $^{36}\text{Ar}/N$ ratios of the surface and mantle reservoirs in order to further constrain
415 the composition of the BSE (*Table 4*).

416

417 *Approach 1* uses gas data only (without gas-rich glasses) for the mantle. The corresponding
418 samples tap large magmatic gas reservoirs which are more prone to homogenize the chemical
419 compositions, and for which the effects of fractional degassing are limited. In this approach
420 we use the mean C/N ratio of 541 and its standard deviation of 199 (computed with CO_2 -rich
421 gas data from *Table 2*; for each sample we took the average of the two numbers obtained with
422 $^4\text{He}/^{40}\text{Ar}^*$ values of either 1.8 or 3.0).

423

424 *Approach 2* also relies on the gas data only, using a median C/N value of 474, and an
425 interquartile range ((IQR= 75th percentile - 25th percentile = Q3 - Q1 = 618-399 = 219). The
426 fact that the mean and the median values are close is consistent with the observed near-
427 Gaussian distribution of gas data (*Fig. 1* and *Fig. S1*).

428

429 *Approach 3* considers all available data for MORB and gases. When taking this approach, the
430 distribution becomes non-Gaussian and we therefore use median and IQR values.

431

432 In the following we considered only *Approaches 2 and 3* (median and quartiles) given the
433 non-Gaussian distribution of MORB C/N data that prohibits using means and standard
434 deviations. For the sake of homogeneity, we also considered median and quartiles for the
435 $^{36}\text{Ar}/N$ values of the mantle and of the BSE. The outcomes of these calculations are
436 summarized in *Table 4* and displayed in *Fig. 4*. Our new range of best estimates for the

437 mantle C/N ratio is between 350-420, which is significantly higher than the ratio of ~90
438 obtained from the CO_2/Ba calibration (Bergin et al., 2015; Hirschmann, 2018), although
439 marginally compatible when taking into account the large range of uncertainties (Q1: 167,
440 G3: 618). We nevertheless assert that our estimate is representative of the mantle value(s)
441 because low mantle C/N values predicted from the CO_2/Ba approach are rarely observed in
442 MORBs or CO_2 -rich gases. Doing so, we estimated a mantle C content around 330-450 ppmw
443 (*Table 4*), which is higher than values previously suggested (e.g., 110 ± 40 ppmw; Hirschmann,
444 2018), although the range of uncertainties is also large (Q1: 157, Q3: 583).

445

446 *4.4- Origin of the BSE C/N ratio*

447

448 From our newly computed mantle C/N ratio we estimated BSE C/N ratio to be 158-219 (range
449 of IQs: 112 - 259; *Table 4*), higher than the previously proposed value of ~60 (Bergin et al.,
450 2015), further emphasizing the non-canonical composition of terrestrial volatiles pointed out
451 by Bergin et al. (2015). The high C/N ratio of the mantle could have resulted from preferential
452 sequestration of nitrogen in planetary cores compared to carbon during Earth's formation
453 events and/or during differentiation of parent planetesimals (Marty, 2012; Roskosz et al.,
454 2013). However, experimental evidence does not support a higher metal-silicate partitioning
455 of N compared to C (Dalou et al, 2017), although the role of other elements like H and
456 pressure need to be assessed (Grewal et al, 2020). One possibility is that the metal-silicate
457 partitioning behavior of C/N could be dependent on pressure, which remains to be
458 experimentally constrained"

459 Preferential degassing of nitrogen during accretionary processes and its subsequent loss to
460 space, possibly combined with metal-silicate partitioning, was proposed as an alternative
461 (Bergin et al., 2015). It is not clear if magma degassing during Earth's accretionary events

462 could have fractionated enough N relative to C to account for the BSE C/N ratio, which
463 signature might have been inherited from planetary bodies that accreted to form the Earth
464 (Hirschmann, 2016). Alternatively, Liu et al. (2019) proposed the formation of nitrogen-rich
465 immiscible liquids that would have removed N from the forming mantle material during
466 metal-silicate interaction.

467

468 Here we add a supplementary dimension to this problem by adding the noble gas/nitrogen
469 ratios. In *Fig.* the C/N ratios of meteorites and terrestrial reservoirs are plotted against the
470 $^{36}\text{Ar}/\text{N}$ ratios (Supplementary Material). The different meteorite groups appear to define a
471 cosmochemical trend from CI-CM, EC and CR groups towards CV and CO groups (and
472 possibly ureilites) having elevated C/N and $^{36}\text{Ar}/\text{N}$ ratios. The CV (and CO) chondrites are
473 depleted in volatile elements (including C, N and noble gases) and moderately volatile
474 elements compared to CM-CIs (Palme and O'Neill, 2013). CVs, which are metamorphized to
475 variable extent, are depleted in volatile elements and thought to represent the non-
476 differentiated upper layers of a differentiated planetary body (Elkins-Tanton et al, 2011). The
477 silicate Earth also presents a monotonous depletion of moderately volatile elements
478 (compared to CI-CM chondrites) that, although more pronounced, resembles that of CVs
479 (Palme and O'Neill, 2013; Braukmüller et al, 2019). This trend cannot be due solely to
480 nebular condensation and is instead consistent with volatile loss during highly energetic
481 events (e.g., impacts) and/or from metamorphism on planetary parent bodies. Hence the
482 cosmochemical trend of meteorite families (*Fig. 4*) may represent the combined effects of
483 parent body metamorphism and partial differentiation. These processes resulted in (i) the
484 depletion of nitrogen relative to C and ^{36}Ar , and (ii) the moderately volatile element depletion
485 trend observed in evolved planetary bodies including the Earth.

486

487 Because Ar is not as depleted as N (*Fig. 4*), this differential behavior of these two elements
488 may be due to variable retention rates of their respective host phases. Indeed, meteoritic noble
489 gases are trapped in a refractory phase (the poorly-defined Q phase associated with insoluble
490 organic matter - IOM) whereas a considerable fraction of nitrogen is trapped in less refractory
491 organics like the soluble organic matter (SOM). Due to the much higher thermal sensitivity of
492 the SOM relative to the IOM (Remusat 2015), the preferential loss of N over noble gases
493 could be accounted for by a preferential destruction of labile, N-bearing organic phases
494 relative to the more refractory, noble gas-bearing ones. An alternative possibility would be the
495 formation of N-rich liquids and subsequent degassing/loss to space as proposed by Lui et al.
496 (2019), which would have depleted N only, and not C nor Ar, from forming planetesimals or
497 proto-Earth.

498

499 The fact that the C-N-Ar signature of the BSE is within the field of CV-CO data does not
500 imply that the building blocks of Earth were made of this material. Conversely we propose
501 that Earth was made of an assemblage of precursors that condensed and evaporated at various
502 temperatures, which are best represented from a volatile element perspective by the CV
503 chondrites. This possibility is independently supported with the monotonous volatile
504 depletion trend of the silicate Earth that parallels that of CVs (Palme and O'Neill, 2013;
505 Braukmüller et al, 2019).

506

507 *4.5- Terrestrial fractionation*

508

509 The terrestrial reservoirs (surface and mantle) clearly plot outside the meteoritic trend (*Fig. 5*).
510 The surface of Earth has a C/N ratio compatible with the range of CI-CMs but its $^{36}\text{Ar}/\text{N}$ ratio
511 is in the range of CV-CO values, whereas the terrestrial mantle has an elevated C/N ratio akin

512 of CV-COs, but its ratio $^{36}\text{Ar}/\text{N}$ ratio is within the range of CI-CM-ECs. Hence none of these
513 reservoirs can be directly linked to known chondritic compositions. However, summing up
514 the C, N ^{36}Ar concentrations of these reservoirs, the obtained BSE composition falls on the
515 cosmic trend and plots within the field of CV-COs, as explained in *Section 4.4*. We therefore
516 propose that surface and the mantle compositions were established by chemical fractionation
517 of the initial BSE, involving preferential incorporation, or retention, of C into the silicate
518 Earth compared to volatile N and noble gases, and its corresponding depletion at the Earth's
519 surface relative to N and noble gases.

520

521 *The C/N* ratio of the mantle, as estimated in this study (350-420), is in line with that outgassed
522 at mid ocean ridges (385 ± 287), as calculated from the CO_2 (Marty and Tolstikhin, 1998) and
523 N fluxes ($5.7 \pm 3.6 \times 10^{12}$ mol/yr; Hirschmann, 2018 and references therein). The similarity
524 between the mantle *C/N* estimated in this study and that degassed at MOR suggests that C and
525 N are not significantly fractionated during degassing under the relatively oxidizing conditions
526 that prevail in the present-day mantle (Hirschmann, 2016), as well as further supporting the
527 validity of our mantle *C/N* estimates. Since both C and N are highly incompatible during
528 mantle melting this is not unexpected but it does highlight that the long-term degassing from
529 the mantle cannot account for the difference in *C/N* between the mantle and the surface.

530 Therefore to explain the difference in the *C/N* between the mantle and the surface requires
531 that C has been preferentially retained in the silicate Earth, or that volatile N has been
532 preferentially lost of the Earth's surface.

533

534 Impacts during terrestrial accretion might have severely degassed impacting planetesimals,
535 which themselves could have already been differentiated between a core, a silicate mantle and
536 a proto-atmosphere and could therefore provide a mechanism for fractionation C and N. If N

537 was preferentially released to the early atmosphere, then during periods of impact driven
538 atmospheric loss, nitrogen would be preferentially lost to space and the C/N of the Earth's
539 surface would be lowered towards its present value. This mechanism has been previously
540 suggested to account for the Earth's subchondritic C/H and Cl/F ratios (Tucker and
541 Mukhopadhyay, 2014). For this mechanism to satisfactorily explain the low C/N of the
542 Earth's surface requires that incompatibility of C and N during mantle melting was different
543 than at present, with N being preferentially degassed relative to C. This may have been the
544 case during the early Earth when the mantle was most likely dominated by reducing
545 conditions (Aulbach and Stagno, 2016). However, recent experimental studies show that even
546 under reducing conditions C and N may behave similarly with both likely being retained in
547 the silicate mantle (Grewal et al., 2020). It therefore remains unclear whether N was
548 preferentially degassed to, and subsequently lost from, the Earth's early atmosphere. Further
549 constraints can also be gained by comparing the $^{36}Ar/N$ of the mantle and the Earth's surface
550 reservoir (*Fig. 5*). Since Ar is a noble gas, its solubility in mantle melts is not subject to
551 differences in oxidation state and therefore even during reducing conditions, Ar would be
552 concentrated in the atmosphere relative to the mantle. Episodes of atmospheric loss should
553 therefore lower the $^{36}Ar/N$ in the atmosphere relative to the mantle, which is not the case. The
554 role of atmospheric loss on the C/N of the surface reservoir therefore remains unclear and
555 warrants further investigation.

556

557 The fractionation of C/N between the Earth's surface and the mantle may be the result of the
558 long-term preferential recycling of C in the crust or the mantle. Li et al. (2019) proposed that
559 the altered oceanic crust is the key carrier of C recycling into the mantle. An intermediate
560 reservoir could be the arc crust (Barry et al. (2019) or the lithosphere (Kelemen and Manning,
561 2015), although this view has been challenged (Plank and Manning, 2019). These processes

562 might have been operative since the onset of plate tectonics but little is known about the
563 tectonic regime of Early Earth. Recycling/sequestration of carbon should have been specific
564 to this element with regard to nitrogen in order to yield the elevated C/N ratio of the mantle.
565 Furthermore, recycling of C should have taken place early in Earth's history in order to avoid
566 a Venus-like greenhouse effect. Sleep et al. (2001) proposed an elegant solution to this
567 problem. These authors postulated that C might have been reintroduced into the mantle early
568 in Earth's history through the foundering of crustal blocks that formed the lid of magma
569 oceans (Hadean vertical tectonics). The carbonate formed in the crust would have formed
570 directly from the massive CO₂ early atmosphere. If such a mechanism could transport C from
571 the surface to the mantle, without venting CO₂ back to the atmosphere, then the fractionation
572 of C and N between the mantle and the surface could have been an early feature. The early
573 sequestration of C into the mantle is independently required to allow for element temperature
574 conditions and the formation of liquid water on the Earth's surface, which is known to have
575 existed early on the Earth's surface. In order to achieve this and avoid the development of a
576 CO₂ dominated atmospheres such as that that exists on Venus, large amounts of CO₂ must
577 have been transported from the surface to the mantle and sequestered there over geological
578 timescales. To account for the different C-N-³⁶Ar compositions of the surface and the mantle,
579 ~ 90 % of surface C should have been being sequestered into silicates during these early
580 episodes. Following the onset of plate tectonics subduction could have further augmented C
581 sequestration in the mantle by preferential recycling C relative to N (Busigny and Bebout,
582 2013) and noble gases (Parai and Mukhopadhyay, 2019, and refs. therein). Carbon
583 sequestration in the deep mantle either via early crustal foundering or subduction therefore
584 not only plays a significant role in fractionating the C and N between the Earth's surface and
585 interior but may play an essential role in modulating the Earth's climate and developing the
586 conditions right for the emergence of life on the early Earth. The fact that the isotopic

587 composition of mantle carbon is comparable with that of surface carbon (Mackenzie and
588 Lerman, 2006) suggests that carbon sequestration did not fractionate its isotopes. In contrast,
589 N and noble gases were less efficiently homogenized between the surface and the mantle,
590 which permitted the survival of isotopic heterogeneities for these elements.

591

592 **5- Conclusions**

593

594 All sample considered in this study show high C/N and $^{36}Ar/N$ ratios relative to the surface
595 (atmosphere plus crust) inventory, pointing towards a mantle that has a unique volatile
596 composition when compared to cosmochemical reservoirs (*Fig. 5*). The BSE composition
597 resembles that of chondritic groups (CV-CO) which are volatile-depleted and appear to have
598 experienced variable extent of metamorphism and planetary differentiation. We propose that
599 Earth formed from a range of already evolved planetary precursors that were depleted in
600 volatile and moderately volatile elements chemically (but not isotopically affected) akin to
601 CV and CO groups. Hence the high C/N character of the BSE may be inherited rather than
602 uniquely due to Earth's forming processes. In this scenario, most C (~90%) was preferentially
603 sequestered into the mantle relative to volatile elements like N and noble gases early in
604 Earth's history, permitting habitable environmental conditions to develop at the surface of our
605 planet.

606

607 **Acknowledgments**

608

609 M.A, D.V.B, M.W.B, D.J.B and B.M were supported by the European Research Council
610 (PHOTONIS project, grant agreement No. 695618 to B.M.). Samples were collected as part
611 of Study # YELL-08056 - Xenon anomalies in the Yellowstone Hotspot. We would like to

612 thank Annie Carlson and all of the rangers at the Yellowstone National Park for providing
613 invaluable advice and help when collecting the samples. This work was partially supported by
614 a grant (G-2016-7206) from the Alfred P. Sloan Foundation and the Deep Carbon
615 Observatory to P.H.B as well as NSF award 2015789 to P.H.B. . Sampling at Mt. Etna and
616 gas analysis was supported by Istituto Nazionale di Geofisica e Vulcanologia Palermo.
617 Fruitful discussions with Marc Hirschmann helped us to shape the ideas presented in this
618 work. We acknowledge detailed and insightful reviews by Sami Mikhail and an anonymous
619 reviewer, and efficient editing by Frederic Moynier. This is CRPG contribution 2741.
620

621 **References**

- 622 1. Alexander, C.M.O., Bowden, R., Fogel, M.L., Howard, K.T., Herd, C.D.K., Nittler,
623 L.R., 2012. The provenances of asteroids, and their contributions to the volatile
624 inventories of the terrestrial planets. *Science* 337, 721–3.
625 doi:10.1126/science.1223474
- 626 2. Allègre, C.J., Hofmann, A.W., O’Nions, R.K., 1996. The argon constraints on mantle
627 structure. *Geophys. Res. Lett.* 23, 3555–3557.
- 628 3. Arevalo, R., McDonough, W.F., Luong, M., 2009. The K/U ratio of the silicate Earth:
629 Insights into mantle composition, structure and thermal evolution. *Earth Planet. Sci.*
630 *Lett.* 278, 361–369. doi:10.1016/j.epsl.2008.12.023
- 631 4. Aulbach, S. & Stagno, V., 2016. Evidence for a reducing Archean ambient mantle and
632 its effects on the carbon cycle. *Geology* 44, 751–754.
- 633 5. Ballentine, C.J., Holland, G., 2008. What CO₂ well gases tell us about the origin of
634 noble gases in the mantle and their relationship to the atmosphere. *Philos. Trans. R.*
635 *Soc. A Math. Phys. Eng. Sci.* 366, 4183–4203. doi:10.1098/rsta.2008.0150
- 636 6. Barry, P.H., Hilton, D.R., Füre, E., Halldorson, S.A., Grönvold, R. 2014. Carbon
637 isotope and abundance systematics of Icelandic geothermal gases and subaerial basalts
638 with implications for mantle-plume related CO₂ fluxes. *Geochim. Cosmochim. Acta*
639 134, 74-99, doi.org/10.1016/j.gca.2014.02.038
- 640 7. Barry, P. H., de Moor, J. M., Giovannelli, D., Schrenk, M., Hummer, D. R., et al.,
641 2019. Forearc carbon sink reduces long-term volatile recycling into the mantle. *Nature*
642 568, 487-492
- 643 8. Bekaert, D. V., Broadley, M.W., Caracausi, A., Marty, B., 2019. Novel insights into
644 the degassing history of Earth’s mantle from high precision noble gas analysis of
645 magmatic gas. *Earth Planet. Sci. Lett.* 525, 115766. doi:10.1016/j.epsl.2019.115766

- 646 9. Bekaert, D. V, Broadley, M.W., Marty, B., 2020. The origin and fate of volatile
647 elements on Earth revisited in light of noble gas data obtained from comet 67P /
648 Churyumov- Gerasimenko Sci. Rep. 10:5796. doi:10.1038/s41598-020-62650-3
- 649 10. Bergin, E.A., Blake, G.A., Ciesla, F., Hirschmann, M.M., Li, J., 2015. Tracing the
650 ingredients for a habitable earth from interstellar space through planet formation. Proc.
651 Natl. Acad. Sci.. U.S.A. 112, 8965–8970. doi:10.1073/pnas.1500954112
- 652 11. Bräuer, K., Kämpf, H., Niedermann, S., Strauch, G., 2013. Indications for the
653 existence of different magmatic reservoirs beneath the Eifel area (Germany): A multi-
654 isotope (C, N, He, Ne, Ar) approach. Chem. Geol. 356, 193–208.
655 doi:10.1016/j.chemgeo.2013.08.013
- 656 12. Bräuer, K., Kämpf, H., Niedermann, S., Strauch, G., Weise, S.M., 2004. Evidence for
657 a nitrogen flux directly derived from the European subcontinental mantle in the
658 Western Eger Rift, Central Europe. Geochim. Cosmochim. Acta 68, 4935–4947.
659 doi:10.1016/j.gca.2004.05.032.
- 660 13. Braukmüller, N., Wombacher, F., Funk, C., Münker, C., 2019. Earth’s volatile
661 element depletion pattern inherited from a carbonaceous chondrite-like source. Nat.
662 Geosci. 12, 564–568. doi:10.1038/s41561-019-0375-x
- 663 14. Broadley, M.K., Barry, P.H., Bekaert, D.V., Byrne, D.J., Caracausi, A., Ballentine,
664 C.J., Marty, B., 2020. Identification of chondritic Proc. Natl. Acad. Sci. U.S.A.
665 doi/10.1073/pnas.2003907117
- 666 15. Busigny, V., Bebout, G.E., 2013. Nitrogen in the Silicate Earth: Speciation and
667 Isotopic Behavior during Mineral-Fluid Interactions. Elements 9, 353–358.
668 doi:10.2113/gselements.9.5.353
- 669 16. Caracausi, A., Favara, R., Giammanco, S., Italiano, F., Paonita, A., Pecoraino, G.,
670 Rizzo, A., Nuccio, P.M., 2003. Mount Etna: Geochemical signals of magma ascent

- 671 and unusually extensive plumbing system. *Geophys. Res. Lett.* 30, 1057.
672 doi:10.1029/2002gl015463
- 673 17. Craig, H., Lupton, J.E., Welhan, J.A., Poreda, R., 1978. Helium isotopes ratios in
674 Yellowstone and Lassen volcanic gases. *Geophys. Res. Lett.* 5, 897–900.
- 675 18. Dalou, C., Hirschmann, M.M., von der Handt, A., Mosenfelder, J., Armstrong, L.S.,
676 2017. Nitrogen and carbon fractionation during core–mantle differentiation at shallow
677 depth. *Earth Planet. Sci. Lett.* 458, 141–151. doi:10.1016/j.epsl.2016.10.026
- 678 19. Dixon, J.E., Stolper, E.M., Holloway, J.R., 1995. An experimental study of water and
679 carbon dioxide solubilities in Mid-Ocean Ridge basaltic liquids. Part I: Calibration and
680 solubility models. *J. Pet.* 36, 1607–1631.
- 681 20. Elkins-Tanton, L.T., Weiss, B.P., Zuber, M.T., 2011. Chondrites as samples of
682 differentiated planetesimals. *Earth Planet. Sci. Lett.* 305, 1–10.
683 doi:10.1016/j.epsl.2011.03.010E
- 684 21. Fischer, T P, Burnard, P., Marty, B., Hilton, D.R., Furi, E., Palhol, F., Sharp, Z.D.,
685 Mangasini, F., 2009. Upper-mantle volatile chemistry at Oldoinyo Lengai volcano and
686 the origin of carbonatites. *Nature* 459, 77–80. doi:10.1038/nature07977
- 687 22. Fitton, J.G., James, D., Kempton, P.D., Ormerod, D.S., Leeman, W.P., 1988. The Role
688 of Lithospheric Mantle in the Generation of Late Cenozoic Basic Magmas in the
689 Western United States. *J. Petrol. Special Lithospheric Issue*, 331–349.
690 doi:10.1093/petrology/Special_Volume.1.331
- 691 23. Giggenbach, W.F., 1980. Geothermal gas equilibria. *Geochim. Cosmochim. Acta* 44,
692 2021–2032.
- 693 24. Goldblatt, C., Claire, M.W., Lenton, T.M., Matthews, A.J., Watson, A.J., Zahnle, K.J.,
694 2009. Early Earth. *Nat. Geosci.* 2, 1–6. doi:10.1038/ngeo692
- 695 25. Grewal, D.S., Dasgupta, R. and Farnell, A., 2020. The speciation of carbon, nitrogen,

- 696 and water in magma oceans and its effect on volatile partitioning between major
697 reservoirs of the Solar System rocky bodies. *Geochim. Cosmochim. Acta* 280, 281-
698 301.
- 699 26. Halldórsson, S.A., Hilton, D.R., Barry, P.H., Füri, E., Grönvold, K., 2016. Recycling
700 of crustal material by the Iceland mantle plume: New evidence from nitrogen
701 elemental and isotope systematics of subglacial basalts. *Geochim. Cosmochim. Acta*
702 176, 206–226. doi:10.1016/j.gca.2015.12.02
- 703 27. Hirschmann, M.M., 2016. Constraints on the early delivery and fractionation of
704 Earth’s major volatiles from C/H, C/N, and C/S ratios. *American Mineralogist* 101,
705 540–553.
- 706 28. Hirschmann, M.M., 2018. Comparative deep Earth volatile cycles: The case for C
707 recycling from exosphere/mantle fractionation of major (H₂O, C, N) volatiles and
708 from H₂O/Ce, CO₂/Ba, and CO₂/Nb exosphere ratios. *Earth Planet. Sci. Lett.* 502,
709 262–273. doi:10.1016/j.epsl.2018.08.023
- 710 29. Hoernle, K., Zhang, Y.-S., Graham, D., 1995. Seismic and geochemical evidence for
711 large-scale mantle upwelling beneath the eastern Atlantic and western and central
712 Europe. *Nature* 374, 34–39. doi:10.1038/374034a0
- 713 30. Jambon, A., Weber, H., Begemann, F., 1985. Helium and argon from an Atlantic
714 MORB glass : concentration, distribution and isotopic composition. *Earth Planet. Sci.*
715 *Lett.* 73, 255–268.
- 716 31. Javoy, M., Pineau, F., 1991. The volatiles record of a “popping” rock from the Mid-
717 Atlantic Ridge at 14°N: chemical and isotopic composition of gas trapped in the
718 vesicles. *Earth Planet. Sci. Lett.* 107, 598–611. doi:10.1016/0012-821X(91)90104-P
- 719 32. Johnson, B., Goldblatt, C., 2015. The nitrogen budget of Earth. *Earth-Science Rev.*
720 148, 150–173. doi:10.1016/j.earscirev.2015.05.006

- 721 33. Labidi, J., Barry, P.H., Bekaert, D. V., Broadley, M.W., Marty, B., Giunta, T., Warr,
722 O., Sherwood Lollar, B., Fischer, T.P., Avice, G., Caracausi, A., Ballentine, C.J.,
723 Halldórsson, S.A., Stefánsson, A., Kurz, M.D., Kohl, I.E., Young, E.D., 2020.
724 Hydrothermal $^{15}\text{N}^{15}\text{N}$ abundances constrain the origins of mantle nitrogen. *Nature* 580,
725 367–371. doi:10.1038/s41586-020-2173-4
- 726 34. Le Voyer, M., Hauri, E.H., Cottrell, E., Kelley, K.A., Salters, V.J.M., Langmuir, C.H.,
727 Hilton, D.R., Barry, P.H., Füre, E., 2019. Carbon fluxes and primary magma CO_2
728 contents along the global mid-ocean ridge system. *Geochem., Geophys. Geosyst.* 20,
729 1387–1424. doi:10.1029/2018GC007630
- 730 35. Li, K., Li, L., Pearson, D. G., & Stachel, T., 2019. Diamond isotope compositions
731 indicate altered igneous oceanic crust dominates deep carbon recycling. *Earth Planet.*
732 *Sci. Lett.*, 516, 190-201.
- 733 36. Li, J., Chen, B., Mookherjee, M., Morard, G., 2020. Carbon versus other light
734 elements in earth's core, in: *Deep Carbon*. pp. 40–65.
- 735 37. Libourel, G., Marty, B., Humbert, F., 2003. Nitrogen solubility in basaltic melt. Part I.
736 Effect of oxygen fugacity. *Geochim. Cosmochim. Acta* 67, 4123–4135.
- 737 38. Liu, J., Dorfman, S.M., Lv, M., Li, J., Zhu, F., Kono, Y., 2019. Loss of immiscible
738 nitrogen from metallic melt explains Earth's missing nitrogen. *Geochem. Persp. Lett.*
739 11, 18–22.
- 740 39. Kelemen, P. B., & Manning, C. E., 2015. Reevaluating carbon fluxes in subduction
741 zones, what goes down, mostly comes up. *Proceed. Nat. Acad. Sci.*, 112(30), E3997-
742 E4006.
- 743 40. Mackenzie, F.T. and Lerman, A., 2006. *Carbon in the Geobiosphere: Earth's outer*
744 *shell*. Springer, 341 pp. ISBN 978-1-4020-4238-6
- 745 41. Marty, B., 2012. *The origins and concentrations of water, carbon, nitrogen and noble*

- 746 gases on Earth. *Earth Planet. Sci. Lett.* 313–314, 56–66.
747 doi:10.1016/j.epsl.2011.10.040
- 748 42. Marty, B., Dauphas, N., 2003. The nitrogen record of crust-mantle interaction and
749 mantle convection from Archean to present. *Earth Planet. Sci. Lett.* 206, 397–410.
- 750 43. Marty, B., Tolstikhin, I.N., 1998. CO₂ fluxes from mid-ocean ridges, arcs and plumes.
751 *Chem. Geol.* 145, 233–248.
- 752 44. Marty, B., Zimmermann, L., 1999. Volatiles (He, C, N, Ar) in mid-ocean ridge
753 basalts: Assessment of shallow-level fractionation and characterization of source
754 composition. *Geochim. Cosmochim. Acta* 63, 3619–3633.
- 755 45. Mikhail, S., Howell, D., 2016. A petrological assessment of diamond as a recorder of
756 the mantle nitrogen cycle. *Amer. Mineral.* 101, 780–787.
- 757 46. Moreira, M., Kunz, J., Allègre, C.J., 1998. Rare gas systematics in Popping Rock :
758 Isotopic and elemental compositions in the upper mantle. *Science* 279, 1178–1181.
- 759 47. Mukhopadhyay, S., 2012. Early differentiation and volatile accretion recorded in
760 deep-mantle neon and xenon. *Nature* 486, 101–104. doi:10.1038/nature11141
- 761 48. Mukhopadhyay, S., Parai, R., 2019. Noble Gases: A Record of Earth’s Evolution and
762 Mantle Dynamics. *Annu. Rev. Earth Planet. Sci.* 47, 389–419. doi:10.1146/annurev-
763 earth-053018-060238
- 764 49. Nakai, S., Wakita, H., Nuccio, M.P., Italiano, F., 1997. MORB-type neon in an
765 enriched mantle beneath Etna, Sicily. *Earth Planet. Sci. Lett.* 153, 57–66.
766 doi:10.1016/S0012-821X(97)00161-1
- 767 50. Ozima, M., Podosek, F.A., 2002. Noble gas geochemistry. Cambridge University
768 Press, Cambridge.
- 769 51. Palme, H., O’Neill, H., 2013. Cosmochemical Estimates of Mantle Composition, 2nd
770 ed, *Treatise on Geochemistry: Second Edition*. Elsevier Ltd. doi:10.1016/B978-0-08-

- 771 095975-7.00201-1
- 772 52. Plank, T. and Manning, C.E., 2019. Subducting carbon. *Nature*, 574, 343-352
- 773 53. Remusat, L. (2015). Organics in primitive meteorites. *EMU Notes in Mineralogy*, 15,
774 1-33.
- 775 54. Rosenthal, A., Hauri, E.H., Hirschmann, M.M., 2015. Experimental determination of
776 C, F, and H partitioning between mantle minerals and carbonated basalt, CO₂/Ba and
777 CO₂/Nb systematics of partial melting, and the CO₂ contents of basaltic source regions.
778 *Earth Planet. Sci. Lett.* 412, 77–87. doi:10.1016/j.epsl.2014.11.044
- 779 55. Roskosz, M., Bouhifd, M.A., Jephcoat, A.P., Marty, B., Mysen, B.O. (2013) Nitrogen
780 solubility in molten metal and silicate at high pressure and temperature. *Geochimica et*
781 *Cosmochimica Acta* 121, 15–28.
- 782 56. Sleep, N.H., Zahnle, K., Neuhoff, P.S., 2001. Initiation of clement surface conditions
783 on the earliest Earth. *Proc. Natl. Acad. Sci. U. S. A.* 98, 3666–3672.
- 784 57. Staudacher, T., 1987. Upper mantle origin for Harding County well gases. *Nature* 325,
785 605–607.
- 786 58. Tucker, J.M. and Mukhopadhyay, S., 2014. Evidence for multiple magma ocean
787 outgassing and atmospheric loss episodes from mantle noble gases. *Earth Planet. Sci.*
788 *Lett.*, 393, 254-265.
- 789 59. Tucker, J.M., Mukhopadhyay, S., Gonnermann, H.M., 2018. Reconstructing mantle
790 carbon and noble gas contents from degassed mid-ocean ridge basalts. *Earth Planet.*
791 *Sci. Lett.* 496, 108–119. doi:10.1016/j.epsl.2018.05.024
- 792 60. Werner, C., Brantley, S., 2003. CO₂ emissions from the Yellowstone volcanic system.
793 *Geochem., Geophys. Geosyst.* 4. doi:10.1029/2002GC000473
- 794 61. Zartman, R.E., Wasserburg, G.J., Reynolds, J.H., 1961. Helium, argon and carbon in
795 some natural gases. *J. Geophys. Res.* 66–1, 277–306.

Type	Country	Location	Sample name	Excess ¹²⁹ Xe	Neon isotopes	CO ₂ %	δ ¹³ C ‰, PDB	N ₂ %	δ ¹⁵ N ‰, ATM	He ppmv	Ar ppmv	³ He/ ⁴ He, Ra	⁴⁰ Ar/ ³⁶ Ar	Refs.
<i>CO₂-rich gases</i>														
CO ₂ well gas	USA	Bueyeros NM	Mitchell n°4	√	DMM	99.9	-3.9	0.12	n.d.	45	28	3.2	22500	1,2
CO ₂ well gas	USA	Bueyeros NM	Mitchell n°12	√	DMM	99.8	-4.1	0.15	n.d.	47	29	3.2	34000	1,2
Bubbling water	Germany	Eifel	Victoriaquelle	√	DMM	99.77	-2.1	0.18	-1.7	32	37	4.4	4040	3,4,5
Bubbling water	Germany	Eifel	Schwefelquelle	√	DMM	99.68	-2.0	0.24	-1.8	31	62	4.5	8287	3,4,5
CO ₂ well gas	Czech Rep.	Eger graben	Bublak	n.d.	n.d.	99.8	-2.0	0.20	-3.2	19	36	5.9	551	3
CO ₂ well gas	Italy	Etna area	Naftia	√	DMM	97.6	-1.0	0.53	n.d.	64	78	6.6	1445	6, 7
Volcanic gas	Tanzania	Oldoinyo Lengai	OLD-2	n.d.	DMM	99.34	-2.4	0.66	-4.1	27	90	6.9	947	8
Geothermal gas	USA	Yellowstone, inside	Mud Volcano 2A	n.d.	PM	98.50	-2.6	0.26	-8.6	15	40	14.4	414	5,7,9
Geothermal gas	USA	Yellowstone, inside	Mud Volcano 2B	n.d.	PM	98.10	-2.7	0.31	-1.9	18	77	13.8	450	5,7,9
Geothermal gas	USA	Yellowstone, border	Turbid Lake 3A	n.d.	PM	96.34	-2.9	0.23	-1.9	34	67	5.27	433	5,7,9
Geothermal gas	USA	Yellowstone, border	Turbid Lake 3B	n.d.	PM	96.55	-2.6	0.22	-0.1	38	62	5.19	433	5,7,9
Geothermal gas	USA	Yellowstone, outside	Brimstone Basin 4A	√	PM	98.15	-2.7	0.16	0.2	72	52	2.77	1536	5,7,9
Geothermal gas	USA	Yellowstone, outside	Brimstone Basin 4B	√.	n.d.	98.05	-2.8	0.16	-1.4	68	52	2.85	1417	5,7,9
<i>Basaltic glasses</i>														
MORB glass	Mid-Atlantic	14°N	2IID43	√	DMM	95.0	-3.7	0.12	-3.1	33	24	8.1	24500	10,11
Subglacial glass	Iceland	64°N	DICE	√	PM	99.9	-3.6	0.04	0.2	14	6.2	17.9	7047	12,13,14,15
<i>Atmosphere</i>						0.04		78.08	0	5.24	9340	1	298.6	16

Table 1: List of selected samples, chemical compositions and geodynamic settings. For the Yellowstone samples, *inside*, *border* and *outside* refer to the location of the sampled sites with respect to the limit of the caldera. DMM and PM refer to the depleted MORB mantle and the plume mantle, respectively, as defined by Ne isotopes measured in these samples. The $^3\text{He}/^4\text{He}$ ratios are expressed as multiples of the atmospheric value of $R_a = 1.39 \times 10^{-6}$. “n.d”: not determined. ppmv: part per million in volume. References: 1: Zartman et al. (1961); 2: Staudacher (1987); 3: Bräuer et al. (2013); 4: Bekaert et al. (2019); 5: Labidi et al. (2020) for N isotope data; 6: Nakai et al. (1997); for He, Ne and Xe isotope data; 7: Chemical analysis carried out at INGV Palermo. 8: Fischer et al. (2009); 9: Broadley et al (2020) for He, Ne and Xe isotope data; 10: Javoy and Pineau (1991); 11: Ne and He data from Moreira et al. (1998), the $^{40}\text{Ar}/^{36}\text{Ar}$ ratio is the highest value reported by these authors; 12: He, Ne and Xe isotope data from Mukhopadhyay (2012), the $^{40}\text{Ar}/^{36}\text{Ar}$ ratio is the highest value reported by this author; 13: Carbon isotopes for sample MD-1 from the same site (Barry et al., 2014); 14: Chemical analysis (CO_2 , He isotopes) carried out at CRPG; 15: Marty and Dauphas (2003) for N isotopes; 16: Ozima and Podosek (2002).

Location	Name	Type	C/ N Meas.	N/ Ar Meas.	He/ Ar Meas.	N/ Ar Corr. Atm.	C/ N Corr. Atm.	⁴ He/ ⁴⁰ Ar* Corr. Atm.	C/N Corr for atm. cont & degas. fract.	
									1.8	3.0
<i>CO₂-rich gases</i>										
Bueyeros NM	Mitchell n°4	DMM	416	86	1.59	86	417	1.60	459	688
Bueyeros NM	Mitchell n°12	DMM	333	103	1.60	103	333	1.60	366	549
Eifel	Victoriaquelle	DMM	277	97	0.87	101	285	0.94	479	718
Eifel	Schwefelquelle	DMM	208	77	0.50	79	211	0.52	565	847
Eger graben	Bublak	n.d.	249	100	0.53	108	559	1.15	798	1198
Etnean province	Naftia	DMM	92	136	0.82	161	98	1.03	152	228
Oldoinyo Lengai	OLD-2	DMM	76	145	0.30	192	83	0.43	256	384
Yellowstone, inside	Mud Volcano 2A	PM	189	130	0.39	355	241	1.35	304	456
Yellowstone, inside	Mud Volcano 2B	PM	158	81	0.23	158	232	0.67	511	767
Yellowstone, border	Turbid Lake 3A	PM	209	69	0.50	131	343	1.57	382	573
Yellowstone, border	Turbid Lake 3B	PM	219	71	0.61	139	352	1.92	334	502
Yellowstone, outside	Brimstone Basin 4A	PM	307	62	1.38	66	344	1.67	366	548
Yellowstone, outside	Brimstone Basin 4B	PM	306	62	1.31	67	348	1.61	379	569
<i>Basalt glasses</i>										
MAR 14°N	2IID43	DMM	396	100	1.36	101	396	1.37	492	738
Iceland 60°N	DICE	PM	1130	142	2.26	146	1141	2.26	952	1428
CO₂-rich gases, mean									412	617
All MORB									535 ± 224	
N-MORB									273 ± 106	
T-MORB									433 ± 392	
E-MORB									1839 ± 641	

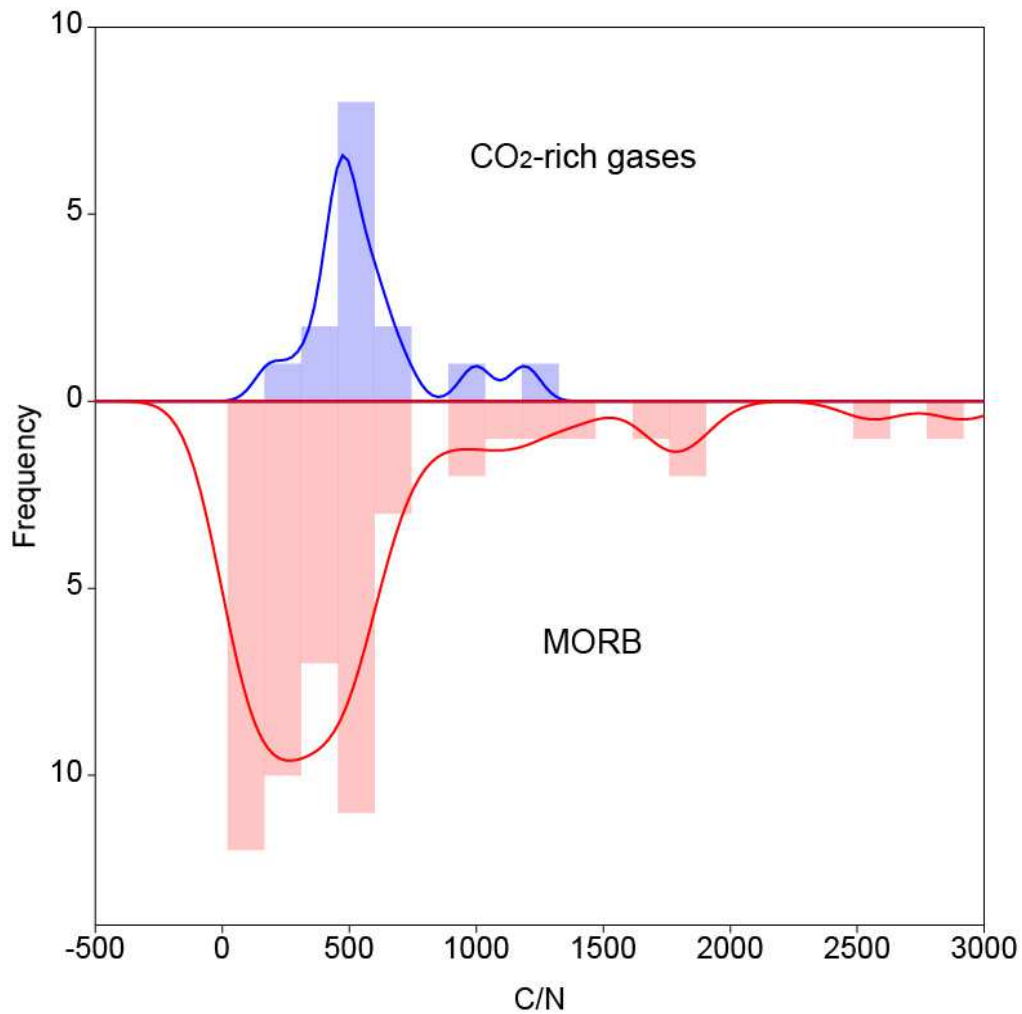
Table 2: Abundance ratios of CO₂-rich gases (same notation as in *Table 1*). The ratios are corrected for atmospheric contamination using *Eqns. 2-5* (first three numerical columns). The 4th column gives the *C/N* ratio where the N content has been corrected for atmospheric contamination. The 5th column gives the ${}^4\text{He}/{}^{40}\text{Ar}^*$ ratio where ${}^4\text{He}$ equals to the He content and ${}^{40}\text{Ar}^*$ is approximated by the total Ar content corrected for air contamination. n.d.: not determined. Note that the ${}^4\text{He}/{}^{40}\text{Ar}^*$ ratios are within, or close to, the mantle production/accumulation range of 1.8-3.0. The exact value of this ratio is not precisely known since it depends on the parent (K+U)/Th abundance ratio and on the residence time of noble gases in the respective mantle sources. The last two columns represent the *C/N* molar ratios corrected for atmospheric contamination and degassing fractionation. Correction for the latter has been made using *Eqn. 1* for the two end-member values of the mantle range. The mean *C/N* ratio of the CO₂-rich gases analyzed here excluding glass data (412-617) of with glass data (453-680) is consistent with the mean MORB (535 ± 224 ; Marty and Zimmermann, 1999).

Location	Type	C/	N/	³⁶ Ar/	⁴ He/	C/ ³ He		N/ ³ He	
		³ He	³ He	N [§]	⁴⁰ Ar*	Corr. degas. fract.		Corr. degas. fract.	
		Meas.	Corr. Atm	Corr. Atm	Corr. Atm	10 ⁹		10 ⁶	
		10 ⁹	10 ⁶	10 ⁻⁷		1.8	3.0	1.8	3.0
<i>CO₂-rich gases</i>									
Harding Bueyeros NM	CM	5.12	12.3	2.91	1.60	4.99	4.50	10.9	6.54
Harding Bueyeros NM	CM	4.88	14.7	2.43	1.60	4.76	4.29	13.0	7.81
Eifel	CM	5.01	18.1	2.46	0.94	4.38	3.94	9.15	5.49
Eifel	CM	5.06	24.3	3.18	0.52	3.91	3.52	6.93	4.16
Eger graben	n.d.	8.34	21.2	2.31	1.15	7.60	6.84	9.52	5.71
Etna area	CM	1.66	18.0	1.56	1.03	1.48	1.33	9.71	5.83
Oldoinyo Lengai	CM	3.82	50.3	1.30	0.43	2.85	2.56	11.1	6.67
Yellowstone	PM	3.29	17.4	2.82	1.35	3.10	2.79	10.2	6.12
Yellowstone	PM	2.86	18.1	6.32	0.67	2.33	2.09	4.55	2.73
Yellowstone	PM	1.42	6.8	7.61	1.57	1.38	1.24	3.62	2.17
Yellowstone	PM	1.26	5.7	7.21	1.92	1.28	1.15	3.82	2.29
Yellowstone	PM	3.87	12.6	15.1	1.67	3.80	3.42	10.4	6.25
Yellowstone	PM	4.39	14.3	15.0	1.61	4.29	3.86	11.3	6.78
<i>Mean</i>				5.40		3.55	3.20	8.79	5.27
<i>Median</i>				2.90		3.80	3.42	9.71	5.83
<i>Basalt glasses</i>									
MAR 14°N	CM	2.60	6.57	2.48	1.37	2.46	2.21	5.00	3.00
N 64°	PM	3.03	2.53	6.84	2.26	3.18	2.86	3.15	1.89

Table 3: C^3He , N^3He and $N^{36}Ar$ ratios of analyzed gases (same notation as in Table 2). [§]: $^{36}Ar/N$ ratios corrected for atmospheric contamination assuming $^{40}Ar/^{36}Ar$ ratios of 40,000 for the convective MORB mantle (Bekaert et al., 2019) and 10,000 for the plume mantle (Mukhopadhyay, 2012), respectively

	Surface	Mantle	BSE
N, ppmw	1.56±0.06	1.10±0.55	
C, ppmw	27±5	A2: 479 ^{Q1: 323} _{Q3: 646} A3: 337 ^{Q1: 211} _{Q3: 504}	
³⁶Ar, 10⁻⁶ ppmw	50	1.29 ^{Q1: 0.82} _{Q3: 1.89}	
C/N mol/mol	21±6	A2: 474 ^{Q1: 399} _{Q3: 618} A3: 352 ^{Q1: 167} _{Q3: 606}	A2: 219 ^{Q1: 174} _{Q3: 259} A3: 158 ^{Q1: 112} _{Q3: 215}
³⁶Ar/N, 10⁻⁷ mol/mol	124±5	2.90 ^{Q1: 2.35} _{Q3: 7.40}	74.3 ^{Q1: 65.7} _{Q3: 85.2}

Table 4: C, N and ³⁶Ar concentrations (inventories divided by the mass of the silicate Earth of 4 x 10²⁷ g) and elemental ratios. For the surface of the Earth (atmosphere and crust), C and N data are from Hirschmann (2018) and refs. therein and the ³⁶Ar concentration corresponds to the atmospheric inventory (Ozima and Podosek, 2002), the crustal reservoir being neglected. For the mantle, the N concentration is from Hirschmann (2018) and the C and ³⁶Ar concentrations are computed from the mantle C/N and ³⁶Ar/N ratios estimated here (see text and Supplementary Material). A2 and A3 refer to *Approach 1* and *Approach 2*, respectively, as described in the main text and in the Supplementary Material section. A2 is median and interquartile range (between brackets, Q1 = 1st interquartile and Q3 = 3rd interquartile, respectively) of gas data only and A3 is median and interquartile range of gas+MORB data.



797

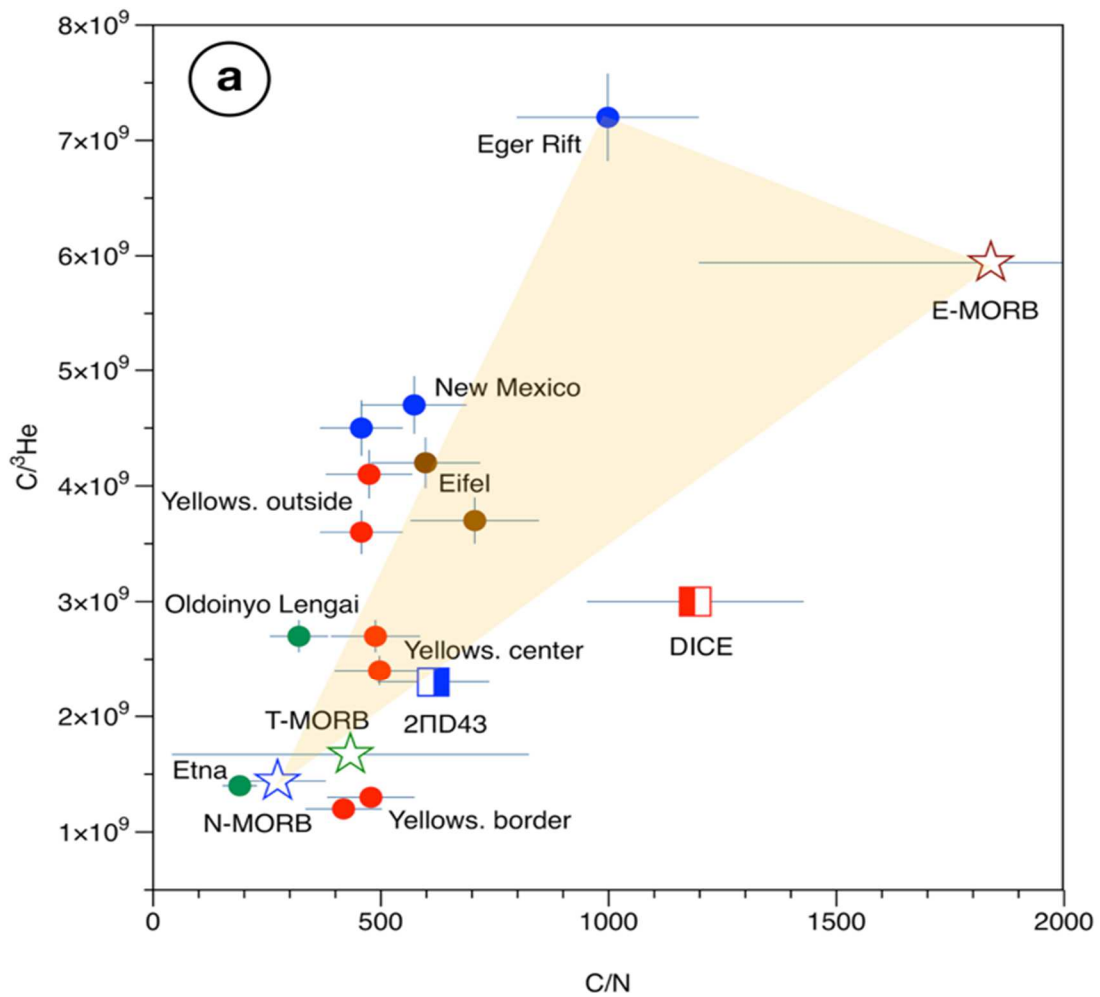
798

799 **Figure 1:** Kernel density distribution (made with Past4[®] software) of *C/N* ratios in CO₂-rich
 800 gases (this work, including gas-rich glasses) and MORB samples (Marty and Zimmermann,
 801 1999). For the CO₂-rich gases, each data corresponds to the mean of the corrected ratios using
 802 ⁴He/⁴⁰Ar* of 1.8 or 3, assigning an uncertainty equal to the difference between the two
 803 calculated values (typically 20 %).

804

805

806



807

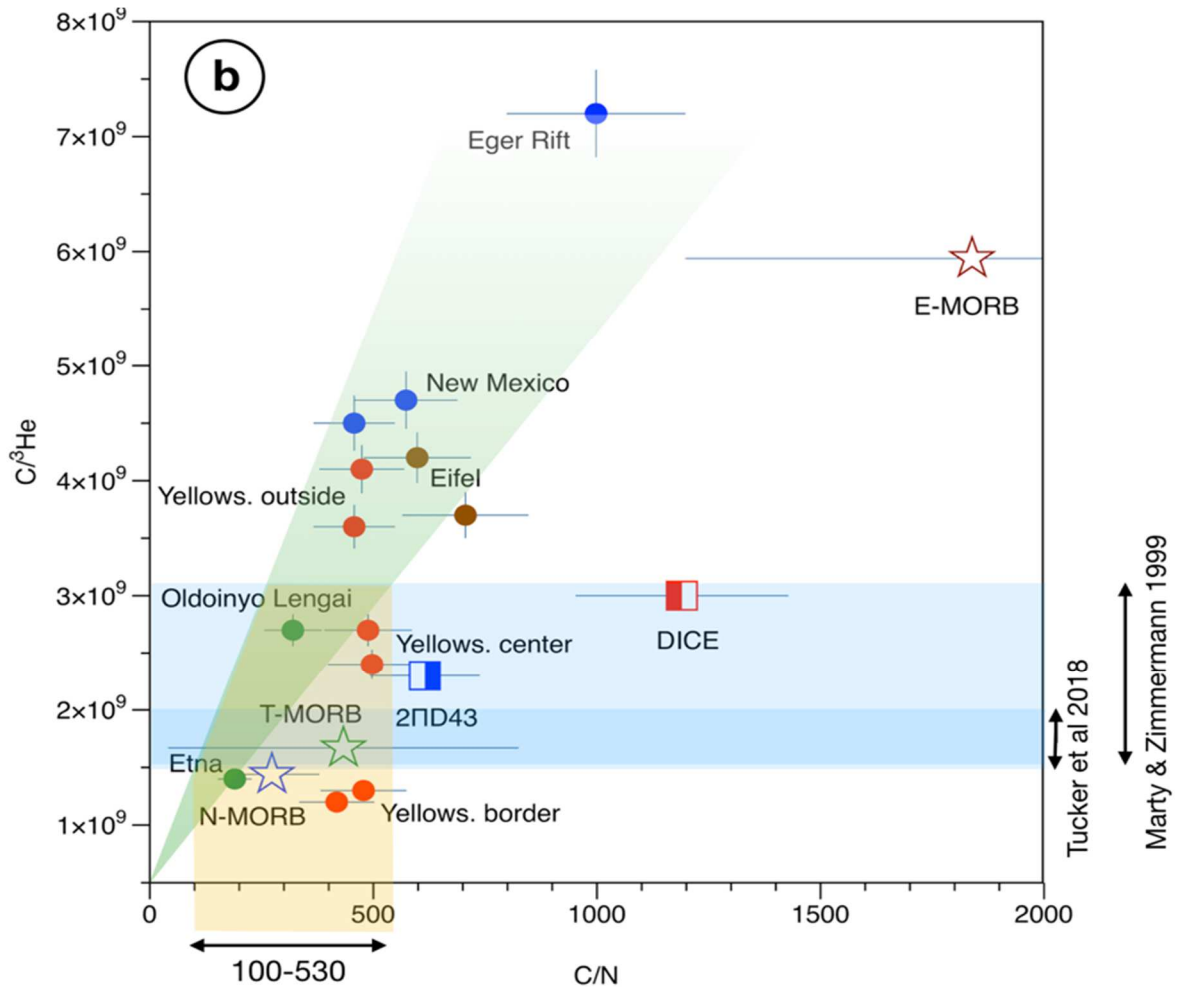
808

809

810 **Figure 2a:** C^3He versus C/N for CO_2 -rich gases (data from *Tables 2 & 3*). Red dots:
811 Yellowstone gases; Green dots: volcanic gases; Brown dots: Eifel gases; Blue dots: New
812 Mexico and Eger Rift gases. Variations can be understood as resulting from a 3-end-member
813 mixing between N-MORB, E-MORB and a crust-influenced component (the latter
814 exemplified by the Eger Rift Bublak gas.

815

816



817

818

819

820

821

822 **Figure 2b:** same format as *Fig. 2a*. Attempt of correction for contribution of crustal C and N.

823 The green area represents C^3He and C/N ratios trending towards the origin (no carbon). N-

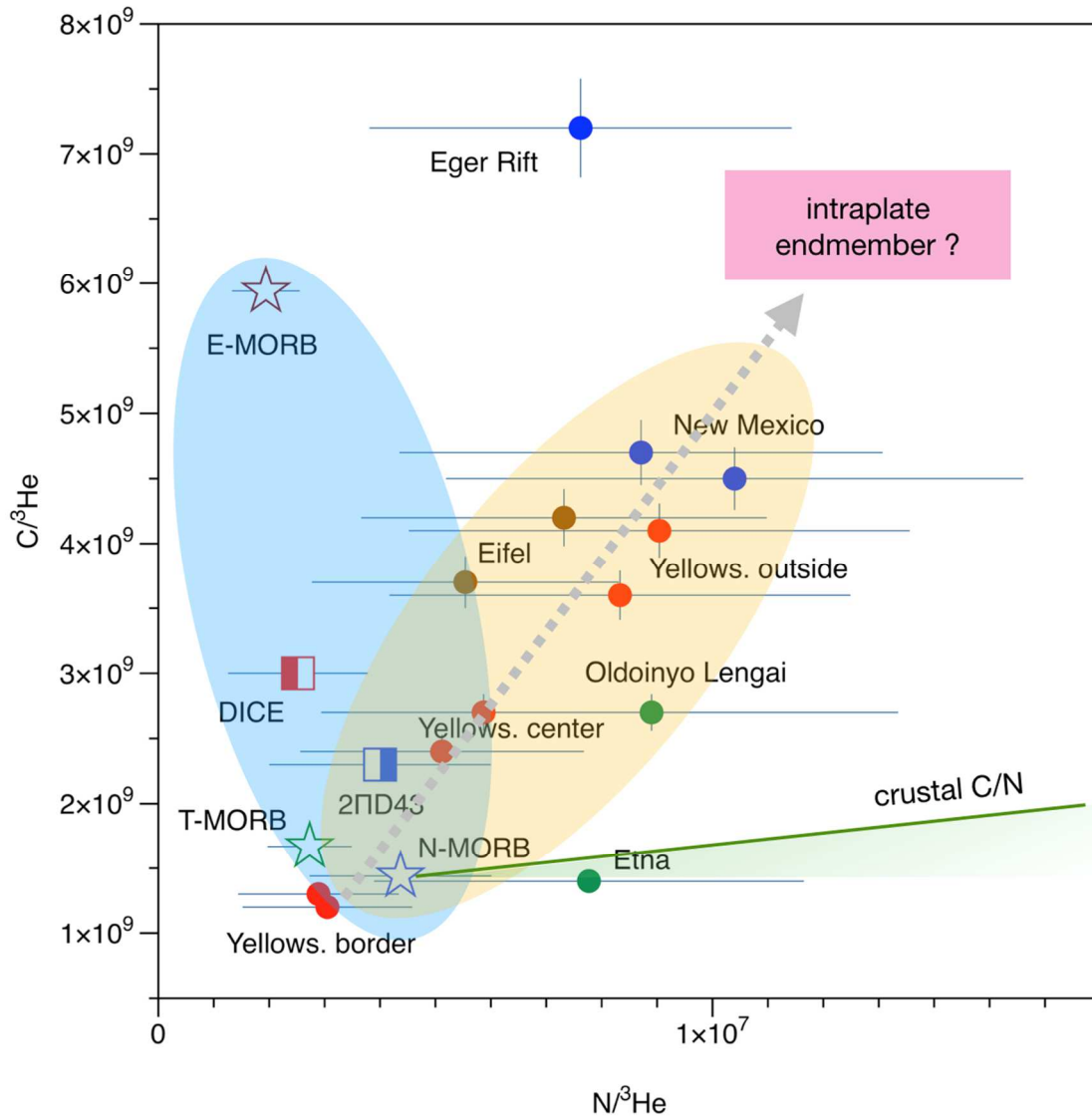
824 MORB and E-MORB averages are from Marty and Zimmermann (1999). The blue areas

825 represent the ranges of C^3He ratios for the MORB mantle according to Tucker et al. (2018)

826 and Marty and Zimmermann (1999), respectively. The mantle source C/N ratio is obtained by

827 intersecting the green area with the blue area.

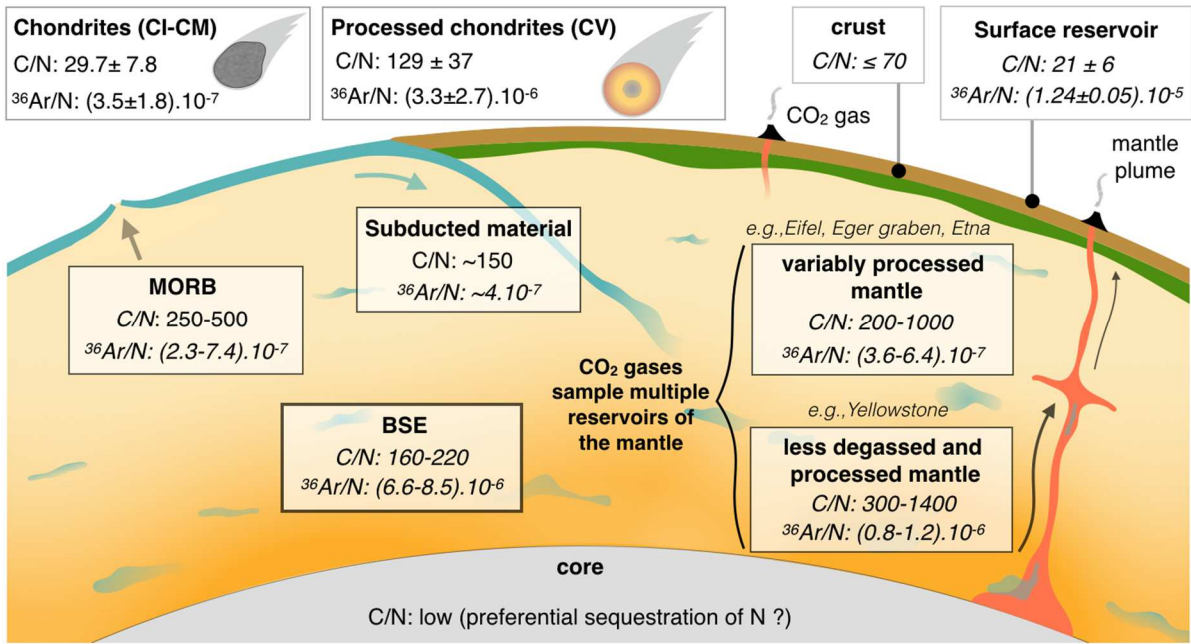
828



829

830

831 **Figure 3:** C^3He ratios as a function of the N^3He ratios for CO_2 -rich gases. Same symbols as
 832 in Fig. 2. The MORB ellipse (blue) encircles N-MORB and E-MORB values (Marty and
 833 Zimmermann, 1999). The yellow ellipse encompasses CO_2 -rich gases, with an approximate
 834 slope corresponding to a C/N ratio of ~ 450 , which is higher than the crustal ratio (≤ 70 , see
 835 Section 4.2). Uncertainties on the N^3He ratios are represented by the difference between
 836 corrected values using the two possible mantle $^4He/^{40}Ar^*$ ratios (1.8 and 3.0).

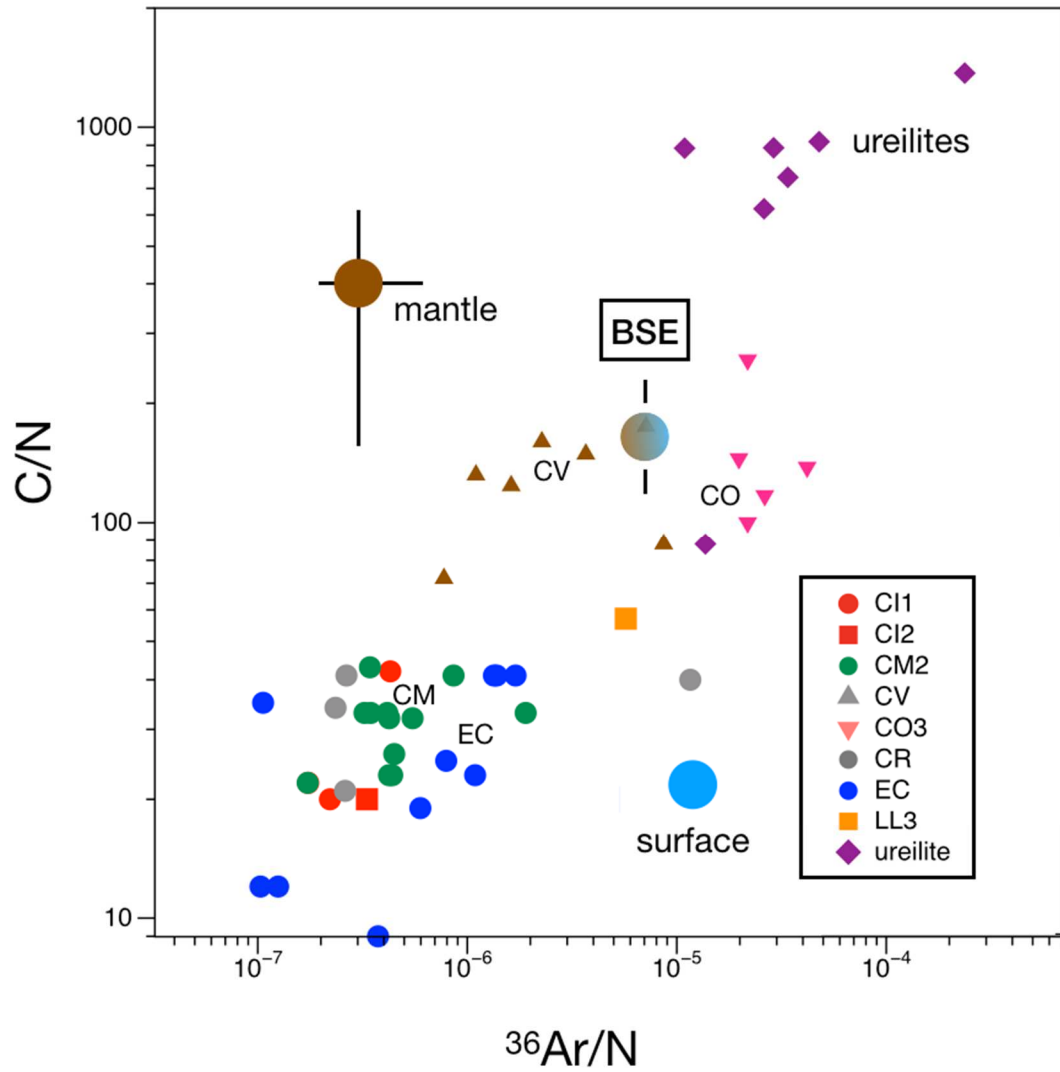


837

838 **Figure 4:** Graphic summary of the estimates presented in this work (data from *Tables 3 and*

839 *4*). Abundance ratios are molar.

840



841

842

843 **Figure 5:** C/N versus $^{36}\text{Ar}/N$ for meteorites and terrestrial reservoirs (note the logarithmic
 844 scale). See *Table 4* for terrestrial estimates, *Supplementary Table S1* for meteoritic data, and
 845 *Supplementary Fig. S5* for discussion of meteoritic data.

846

847

848

849 Alexander, C.M.O., Bowden, R., Fogel, M.L., Howard, K.T., Herd, C.D.K., Nittler, L.R.,
850 2012. The provenances of asteroids, and their contributions to the volatile inventories of
851 the terrestrial planets. *Science* 337, 721–3. doi:10.1126/science.1223474

852 Alexander, C.M.O., Howard, K.T., Bowden, R., Fogel, M.L., 2013. The classification of CM
853 and CR chondrites using bulk H, C and N abundances and isotopic compositions.
854 *Geochim. Cosmochim. Acta* 123, 244–260. doi:10.1016/j.gca.2013.05.019

855 Bischoff, A., Palme, H., Ash, R.D., Clayton, R.N., Schultz, L., Herpers, U., Stöffler, D.,
856 Grady, M.M., Pillinger, C.T., Spettel, B., Weber, H., Grund, T., Endreß, M., Weber, D.,
857 1993. Paired Renazzo-type (CR) carbonaceous chondrites from the Sahara. *Geochim.*
858 *Cosmochim. Acta* 57, 1587–1603. doi:10.1016/0016-7037(93)90014-N

859 Bogard, D.D., Clark, R.S., Keith, J.E., Reynolds, M.A., 1971. Noble gases and radionuclides
860 in Lost City and other recently fallen meteorites. *J. Geophys. Res.* 76, 4076–4083.
861 doi:10.1029/jb076i017p04076

862 Crabb, J., Anders, E., 1981. Noble gases in E-chondrites. *Geochim.* 45, 2443–2464.

863 Downes, H., Abernethy, F.A.J., Smith, C.L., Ross, A.J., Verchovsky, A.B., Grady, M.M.,
864 Jenniskens, P., Shaddad, M.H., 2015. Isotopic composition of carbon and nitrogen in
865 ureilitic fragments of the Almahata Sitta meteorite. *Meteorit. Planet. Sci.* 50, 255–272.
866 doi:10.1111/maps.12413

867 Eugster, O., Lorenzetti, S., Krähenbühl, U., Marti, K., 2007. Comparison of cosmic-ray
868 exposure ages and trapped noble gases in chondrule and matrix samples of ordinary,
869 enstatite, and carbonaceous chondrites. *Meteorit. Planet. Sci.* 42, 1351–1371.
870 doi:10.1111/j.1945-5100.2007.tb00579.x

871 Göbel, R., Ott, U., Begemann, F., 1978. On trapped noble gases in ureilites. *J. Geophys. Res.*
872 83, 855. doi:10.1029/JB083iB02p00855

873 Grady, M.M., Verchovsky, A.B., Franchi, I.A., Wright, I.P., Pillinger, C.T., 2002. Light
874 element geochemistry of the Tagish Lake CI2 chondrite : Comparison with CI1 and CM2
875 meteorites. *Meteorit. Planet. Sci.* 37, 713–735.

876 Grady, M.M., Wright, I.P., Carr, L.P., Pillinger, C.T., 1986. Compositional differences in
877 enstatite chondrites based on carbon and nitrogen stable isotope measurements. *Geochim.*
878 *Cosmochim. Acta* 50, 2799–2813.

879 Grady, M.M., Wright, I.P., Swart, P.K., Pillinger, C.T., 1985. The carbon and nitrogen
880 isotopic composition of ureilites : implications for their genesis 49.

881 Kerridge, F., 1985. Carbon, hydrogen and nitrogen in carbonaceous chondrites: abundances
882 and isotopic compositions in bulk samples. *Geochim. Cosmochim. Acta* 49, 1707–1714.

883 Mazor, E., Heymann, D., Anders, E., 1970. Noble gases in carbonaceous chondrites. *Geochim.*
884 *Cosmochim. Acta* 34, 781–824.

885 Murty, S.V.S., Mahajan, R.R., Jenniskens, P., Shaddad, M.H., Eldien, B., 2010. Noble gases
886 and nitrogen in the Almahata Sitta ureilite. *Meteorit. Planet. Sci.* 45, 1751–1764.
887 doi:10.1111/j.1945-5100.2010.01095.x

888 Nakamura, T., Noguchi, T., Zolensky, M.E., Tanaka, M., 2003. Mineralogy and noble-gas
889 signatures of the carbonate-rich lithology of the Tagish Lake carbonaceous chondrite:
890 evidence for an accretionary breccia. *Earth Planet. Sci. Lett.* 207, 83–101.
891 doi:10.1016/S0012-821X(02)01127-5

892 Okazaki, R., Nagao, K., 2017. Title Primordial and cosmogenic noble gases in the Sutter's
893 Mill CM chondrite. *Meteorit. Planet. Sci.* 52, 669–689. doi:doi.org/10.1111/maps.12819

894 Ott U., Loehr H.P. and Begemann F. (1985) Trapped noble gases in 5 more ureilites and the
895 possible role of Q. *Lunar Planet. Sci.* **16**, 639-640.

896 Patzer, A., Schultz, L., 2001. Noble gases in enstatite chondrites I: Exposure ages, pairing,
897 and weathering effects. *Meteorit. Planet. Sci.* 36, 947–961. doi:10.1111/j.1945-

898 5100.2001.tb01932.x

899 Pearson, V.K., Sephton, M.A., Franchi, I.A., Gibson, J.M., Gilmour, I., 2006. Carbon and

900 nitrogen in carbonaceous chondrites: Elemental abundances and stable isotopic

901 compositions. *Meteorit. Planet. Sci.* 41, 1899–191