

# Gas mobility in rheologically-layered volcanic conduits: the role of decompression rate and crystal content on the ascent dynamics of magmas

*L. Spina<sup>1,2,3\*</sup>, D. Morgavi<sup>2</sup>, A. Costa<sup>4</sup>, B. Scheu<sup>3</sup>, D.B. Dingwell<sup>3</sup>, D. Perugini<sup>2</sup>*

<sup>1</sup> *Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Roma1, Via di Vigna Murata 605, 00143, Rome, Italy.*

<sup>2</sup> *Dipartimento di Fisica e Geologia, Università di Perugia, Piazza dell'Università 1, 06123, Perugia, Italy*

<sup>3</sup> *Department für Geo- und Umweltwissenschaften, Ludwig-Maximilians-Universität München, Theresienstrasse 41, 80333, Munich, Germany.*

<sup>4</sup> *Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Bologna, Via Donato Creti 12, 40128, Bologna, Italy.*

*\*Corresponding author: [laura.spina@ingv.it](mailto:laura.spina@ingv.it)*

*Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Roma1, Via di Vigna Murata 605, 00143, Rome, Italy.*

## **Abstract:**

Unravelling the rheological behaviour of magmas is fundamental for hazard assessment. At shallow depth the combined effects of degassing, vesiculation and crystallization are likely to produce dramatic changes in the rheology, hence modulating flow dynamics and eruptive style. The rheological evolution from a low viscosity crystal-poor, bubble-free, water-rich melt to a highly viscous crystal-rich, vesicular magma containing a water-poor melt often occurs in the conduit. To clarify the viscous flow dynamics of rheologically-layered volcanic conduits, we performed decompression experiments using a magma analogue system characterized by a low-viscous Layer L (10 Pa s) at the bottom and a high-viscous particle-bearing Layer H ( $\geq 1000$  Pa s) at the top. Silicone oils and spherical glass beads are employed as magma and crystal analogues, respectively. Three sets of experiments address the effects of: 1) decompression rate (ca.  $10^{-2}$  and  $10^4$  MPa/s); 2) crystal content in the high viscosity magma (0, 10, 30 and 70 vol.%); and 3) volume ratio of the two rheological layers (0.6 or 0.3). Our results indicate that decompression rate exerts the most dramatic role, yielding changes in time-scale of outgassing up to two orders of magnitude, and affecting the style of decompression response (permeable outgassing or fragmentation). The solid fraction 1) strongly modulates gas mobility, 2) influences the pervasiveness of fragmentation and 3) affects the extent of mingling in the experimental conduit. These results demonstrate that the properties of a shallow, partially-crystallized portion of the magmatic column and its response to varying ascent rate are primary controls on eruptive style.

## **Highlights**

1. First decompression experiments on a rheologically-layered analogue system
2. Decompression history is the most relevant parameter affecting gas mobility
3. At low decompression rate, solid loading affects permeability and mingling efficiency
4. At high decompression rate, solid fraction influences the spatial pervasiveness of fragmentation

42    **Keywords:** Decompression experiments, degassing, fragmentation, analogue magma, viscosity  
43    contrast, mingling

## 1. Introduction

The rheological properties of magmas during their ascent within a feeding system are a crucial determinant of the state and distribution of magmatic fluids. Thus understanding how rheology affects magma flow dynamics and eruptive behaviour is of primary importance. Due to the sensitivity of rheology to crystallinity, liquid composition and vesicularity, rheological contrasts at shallow depth within the feeding systems of volcanoes are likely to be ubiquitous. As magma ascends to the surface, several processes occur which are capable of causing dramatic changes in the physical properties, and in particular in the rheological behaviour and permeability of the magma. Shallow exsolution of H<sub>2</sub>O by magma degassing and resultant magma crystallization are amongst the most common processes inducing rheological stiffening at the top of the magmatic column, thus generating a rheological stratification [e.g. Cashman et al., 2004]. If the gradient of viscosity along the magmatic column is large, the rheological transition can be parameterised as a two-layer flow consisting of a more evolved viscous magma (likely crystal-rich) in the upper region and a less viscous crystal-poor fluid in the lower part [e.g. Barmin et al., 2002].

The resultant heterogeneity in the physical properties of magma within a volcanic conduit has been well-documented in the literature [e.g. Polacci et al., 2009]. Such rheological contrasts are, in turn, anticipated to modulate the style of eruptive activity and hence the range and nature of the volcanic hazards resulting [e.g. Lautze and Houghton, 2005; 2006; Del Bello et al., 2015]. Lava dome eruptions in particular, represent a good example of the control of rheological stratification of the magmatic column in the volcanic conduit on eruptive behaviour, dominated by crystallization kinetics [Melnik and Sparks, 1999; 2005] or thermal effects [Costa and Macedonio, 2002]. Such processes can operate on different time scales [Melnik et al., 2008 and references therein]. Extrusion of lava from active domes is commonly characterized by pulsations of the discharge rate, with large periodic fluctuations often corresponding to cycles observed in ground deformation and/or seismic data from active silicic volcanoes such as Soufriere Hills volcano (Montserrat), Mount Pinatubo (Philippines)

70 and Mt. St. Helens (U.S.A.) [e.g. Melnik et al., 2008 and references therein]. For long-term  
71 fluctuations (years), Barmin et al. [2002] proposed that periodic and non-pulsatory behaviour of  
72 magma discharge rate in extrusive eruptions is a consequence of non-linear dynamics and related to  
73 rheological stiffening of degassing and crystallizing magma, whereby the period of oscillation is  
74 controlled, amongst other factors, by the viscosity ratio between a crystallized region and an  
75 underlying crystal-poor magma. For short-term fluctuations (hours), Costa et al. [2012] modelled  
76 such periodicity in lava extrusion behaviour of Soufriere Hills Volcano (Montserrat) assuming a  
77 stick-slip condition at the shallow conduit wall and a deeper elastic dyke whereby a change in the  
78 periodicity can be due to an increase of overpressure in the system (as observed at Soufriere-Hills  
79 Volcano when the duration of dome cycles decreased from ca. 12 to ca. 7 hours in response to  
80 gravitational unloading of 2 MPa corresponding to ca. 100 m of the dome).

81 Here, we investigate the influence of the state of the physical system (i.e. crystal content,  
82 viscosity ratio, relative proportion of the heterogeneous magmas) at the rheological transition on  
83 eruptive style using decompression experiments. We conducted a series of analogue experiments, to  
84 mimic a scenario whereby two magmas with different viscosities (liquid viscosities: 10 and 1000 Pa  
85 s), are superimposed, one on the other, in a conduit. To account for the contribution to the rheological  
86 contrast resulting from shallow degassing, experiments were performed on a series of samples having  
87 fixed percentages of analogue crystals (0, 10, 30, and 70 vol. %) in the highly viscous magma. The  
88 viscosity ratios obtained at the interface span the range of  $10^2$ - $10^6$ , similar to theoretical estimates for  
89 volcanoes [e.g. Barmin et al., 2002; Capponi et al., 2016]. The experiments were performed at average  
90 decompression rates of  $10^{-2}$  MPa/s and  $10^4$  MPa/s, to constrain the role of overpressure in the system.  
91 Finally, we investigated the effect of the relative proportions of the highly viscous phase versus the  
92 low viscosity magma.

93

## 94 **2. Methods**

95

## 2.1 Experimental apparatus

The experimental setup centres on a cylindrical Plexiglass high-pressure autoclave (height = 20 cm and internal radius  $r = 2.0$  cm; **Figure 1**), sealed at its top by a set of copper diaphragms, and attached to an expansion chamber operating at ambient pressure. The sample is located at the bottom of the autoclave, and saturated under Argon-gas at a pressure of 10 MPa for at least 3 days (6 days for Exp10, in accordance to the increased sample volume; **Table 1**); this time is chosen to allow for a complete saturation of the gas phase [Spina et al., 2016a,b]. Once complete saturation is achieved, the sample is then decompressed either fast or slowly. “Slow” decompression (average decompression rate ca.  $10^{-2}$  MPa/s; Spina et al., 2016a) is performed by opening a dedicated manual valve in the manometer and allowing gas to evacuate the system. Fast decompression is achieved by induced failure of the set of copper diaphragms sealing the autoclave at its top [Spina et al., 2016a,b for further information on the setup]. Fast decompression rate is characterized by an exponential decay, approximated by two linear segments of the decompression curve: an initial decompression rate on the order of  $10^4$  MPa/s (from decompression initiation up to ca. 70% of pressure decay) followed by a slower decompression of  $\approx 10^3$  MPa/s. Like in Spina et al. [2016a], we refer to an average decompression rate of  $10^4$  MPa/s as most representative of the depressurization process. Sample degassing is monitored by tracking the temporal evolution of sample expansion and outgassing periodicity using a high-speed camera with sampling rates of 5000 and 50 fps for the instantaneous and slow decompression experiments, respectively (**Section 2.2** for more details). The pressure state within the system is recorded by a pressure sensor located at the top of the autoclave. Silicone oils with viscosities of ca. 10 and 1000 Pa s and density of  $970 \text{ kg/m}^3$  were chosen as proxies for melts with different rheological properties and juxtaposed vertically, with the high viscosity liquid (hereafter Layer H) placed at the top of the low viscosity liquid (here Layer L; **Figure 1b**). To evaluate the role of solid loading in the uppermost section of the conduit (here Layer H) we added in variable percentages ( $\phi_1$ ; **Table 1**) to the 1000 Pa s silicone oil spherical glass beads, with diameter and aspect ratio (estimated on a population of more than 500 particles) equal to  $83 \pm 11 \mu\text{m}$  and  $1.09 \pm 0.12$ ,

123 respectively, and density of  $790 \text{ kg/m}^3$ . A density of the analogue particles similar to that of the  
124 silicone oil is a fundamental prerequisite to avoid crystal settling. Note that it is our goal here to  
125 address the short-term interaction between two layers at a specific time rather than the long-term  
126 evolution of the system, which would involve crystal settling. The particle-bearing Layer H was  
127 prepared following Cimorelli et al. [2011]: silicone oil and particles were mixed according to their  
128 weight, then centrifuged for at least 1 hour to remove bubbles entrained in the mixture and finally  
129 stirred carefully to establish a homogeneous particle distribution. Next, we first poured Layer L into  
130 the apparatus, and then we slowly superimposed particle-bearing Layer H on the top. The maximum  
131 packing fraction for spheres of equal diameter is about 74 vol.%, whereas random loose packing is  
132 about 60 vol.% [Costa, 2005 and references therein]. Keeping in mind that the particle size  
133 distribution of analogue crystals, in our experiments, is not completely uniform (random packing of  
134 polydisperse spheres is about 87 vol.%; Costa, 2005 and references therein], the value of packing  
135 fraction, estimated for our experiments around 70 vol.% (evaluated according to Cimorelli et al.,  
136 2011), is within this range.

137 The viscosity of Layer H after the addition of particles was estimated with the model of Costa et al.  
138 [2009; **Table 1**] using the best fit parameters obtained by Spina et al. [2016a]. The addition of  
139 particles causes Layer H to appear greyish and more opaque than Layer L. Therefore, it can be readily  
140 detected by image analysis. In order to distinguish the two layers in those experiments without  
141 particles (Exp01, Exp02) we added a small amount ( $0.025 \text{ g/cm}^3$ ) of yellow food colorant (E102,  
142 E172) to the 10 Pa s oil (Layer L).

143 Different sets of experiments were performed to identify the role of 1) solid loading; 2)  
144 decompression rate, and 3) volume ratio, by changing the proportions of the two layers. A complete  
145 description of the experimental conditions is given in **Table 1**.

146

## 147 **2.2. Analytical methods**

148

149 During decompression, the concentration of dissolved volatiles in the saturated samples  
 150 eventually exceeds the equilibrium solubility at the lowered pressure and bubbles nucleate. The  
 151 saturated samples expand under the dominant force of the growing bubble swarm. Then, the mixture  
 152 enters the outgassing regime and periodic oscillations of the sample surface are observed in response  
 153 to the degassing dynamics of the foamed sample. A detailed description of the slow and fast  
 154 decompression of the different systems is given in **Section 3**. To quantify and compare the different  
 155 behaviour of the investigated systems upon decompression, we measured the time-dependent length  
 156 of expansion of the flow front  $F(t)$ , i.e. the total height of the sample at a given time  $t$ , measured at  
 157 the centre of the flow front surface (**Supplementary Table S\_table01**). The measured expansion of  
 158 the sample is a function of bubble nucleation and growth in response to decompression. This  
 159 parameter has been used previously to characterize gas mobility in two phase systems [e.g. Taddeucci  
 160 et al., 2006; Oppenheimer et al., 2015; Spina et al., 2016a;b] and provides information on the average  
 161 velocity  $v$  of the flow front of the expanding mixture at a given time  $t$ , i.e.  $v=(F(t)-F_0)/\Delta t$ , as well as  
 162 the average gas volume ratio  $G(t)=F(t)/[F(t)+F_0]$ , where  $F_0$  is equal to the initial sample length.

163 To quantify the timescale of outgassing in the experiments, we automatically counted the  
 164 number of peaks (i.e. local maxima) and the time elapsed between them in the flow front expansion  
 165 curve, each representing an outgassing episode followed by the abrupt collapse of the sample surface  
 166 (for this purpose we used the Matlab© function *findpeaks.m*). The expansive ascent of the fluid is a  
 167 multi-scale phenomenon, i.e. characterized by the coexistence of timescales related to degassing at  
 168 different scales, from the slow bulk mixture volume change due to exsolution to the small-scale  
 169 surface height variations, related to the foam dynamics, that mark outgassing episodes. Hence, we  
 170 have isolated the latter by picking only local maxima in the flow front expansion curve higher than  
 171 2.5 mm and spaced 15 seconds for slow decompression experiments and 0.5 seconds for fast  
 172 decompression experiments. These parameters have been broadly tested to maximize the detection  
 173 capability. An example of the obtained results is given in **Figure 2**.

### 3. Results

#### 3.1 Visual observation at different decompression rates

The results of slow and fast decompression experiments are shown in **Figures 3** and **4**, respectively. The two sets of experiments are characterized by very different expansion timescales and degassing styles.

In the slow decompression experiments gas exsolution in Layer H is observed both in the experiment performed using only the liquid phase (Exp01) and in particle-bearing runs (Exp03, Ep05, Exp07, Exp09 and Exp10). While the expansion of Layer H is proceeding, the nucleation process initiates in Layer L; the generated bubbles ascend to the Layer L-Layer H interface and interact with the solid network. Bubble nucleation in Layer L occurs first homogeneously within the sample; following which bubbles are also nucleated heterogeneously from the bottom of the autoclave. At a later stage, the system is characterized by vigorous oscillations of the surface, marking cyclic outgassing of the analogue foam (**Figure 3; Supplementary movies S01, S03, S05, S07, S09, S10**), as previously observed in decompressed homogeneous single-layer samples with liquid viscosities above 100 Pa s by Spina et al. [2016a, b]. In all the cases shown in **Figure 3** the absence of rough surfaces in the foam is consistent with the inference that no fragile rupture has occurred along the film between bubbles [e.g. Namiki and Manga, 2008].

In experiments Exp01, Exp03, and Exp05 bubbles from Layer L are easily entrained in the crystal- and/or bubble-rich Layer H. In contrast, in experiments performed at 70 vol.% of crystals (Exp07, Exp09, Exp10; **Supplementary movies S07, S09, S10**) bubbles were observed to be decelerating and gathering at the interface. In all experiments, we noticed the development of a region of mingling between the two layers.

In fast decompression experiments all the particle-bearing samples (Exp04, Exp06, Exp08) experienced fragmentation of Layer H, followed by the ejection of sample fragments. For the same system configuration (i.e. same viscosity contrast), the increased shear-rate induces a fragile-like



201 response of viscous Layer H (**Figures 4, 5; Supplementary movie S04, S06, S08**). In our case, foam  
202 expansion precedes the fragmentation by 2.8, 4.8, 5.4 ms in Exp04, Exp06 and Exp08, respectively  
203 (**Figure 5**), implying that rapid deformation due to bubble expansion is likely to play an active role  
204 in the fragmentation process. Successively, fragmentation occurs; in Exp04 (crystal content equal to  
205 10 vol.%), fragmentation appears along a single surface, that cuts the shallowest portion of the Layer  
206 H (**Figure 4b**). The fragment is ejected and falls back above the non-fragmented part of Layer H. At  
207 higher crystal content (Exp06, 30 vol.%) sample cracking is more pervasive and generate different  
208 fragments that partially fall back on the sample (**Figure 4c**). At the maximum crystal content (Exp08;  
209 70 vol.%) the disruption is completely pervasive, fragmented sample portions are ejected in the low  
210 pressure tank or entrained in the ascending Layer L (**Figure 4d**). In all cases, after fragmentation of  
211 the Layer H, Layer L, partially or completely unloaded, rapidly degasses and expands, permeating  
212 Layer H and reaching the sample surface.

213 Note that whereas for slow decompression experiments it is evident that Layer H and Layer  
214 L are well-coupled, the same is not true for particle-bearing fast decompression experiments, the  
215 partially-fragmented Layer H being permeated and finally overlain by bubble-bearing low viscosity  
216 Layer L. This distinct response (dominated by bubble expansion for Layer L, characterized by  
217 fragmentation for Layer H) prevents efficient coupling and mingling of the two liquids because the  
218 interaction between the two layers is occurring for a very short time during the experiment.

### 219 **3.2 Measurements of sample expansion velocity and outgassing periodicity**

220

221 In **Figure 6** we report the average velocity  $v$  (i.e. ascent speed mm/s) of the flow front of the  
222 expanding mixture measured from initial sample expansion until the beginning of the outgassing  
223 regime (**Section 2.2**).

224 For slow decompression experiments (**Figure 6a**), the average ascent velocity measured  
225 during the first episode of sample expansion ranges between ca. 0.03 and 0.10 mm/s, as a function of  
226 solid loading. A maximum value in the ascent speed is observed at solid loading corresponding to 10  
227 vol.%, followed by a rapid decrease in ascent velocity with particle loading (**Section 4.2**). An increase

228 in the ascent speed of up to 2 orders of magnitude is observed upon increasing the decompression  
229 rate by orders of  $10^6$ . Decompression experiments performed at an average rate of  $10^4$  MPa/s  
230 exhibited average initial velocities of 14.2 to 30.9 mm/s. Additionally, in fast decompression  
231 experiments, the increase in solid loading of Layer H is accompanied by a clear decrease in the ascent  
232 velocity (**Figure 6b**) of the liquid front.

233 In **Figure 7** we show the height of the sample flow front reached during each outgassing  
234 episode (an important indication of the amount of gas in the mixture at that stage; **Figure 7a**) together  
235 with the relative outgassing periodicity (**Figure 7b**), measured as the time interval between  
236 consecutive outgassing peaks (**Section 2.2**), for different experiments with the same system geometry  
237 (Exp09 and Exp10 are reproduced in **Figure 8**). For crystal-bearing experiments at crystal content up  
238 to 30 vol.%, the peak height (corresponding to the maximum sample expansion before outgassing) is  
239 higher than the experiments performed without glass beads. In general, the height of the peaks  
240 decreases as the crystal content increases. The same is observed for the fast decompression runs  
241 (**Figure 7a**); however, maximum peak height here is twice that observed during slow decompression  
242 experiments. The outgassing periodicity is ca. 1-1.7 seconds when pressure is instantaneously  
243 released; a much wider range, spanning in between ca. 15-150/200 seconds, characterizes slow  
244 decompression experiments.

245 The range of peak heights for Exp01 is notably wider than in the other experiments and reflects  
246 longer outgassing periods (**Figure 7a, b**). Exp01 was additionally accompanied by a decrease of the  
247 peak amplitude from average values of 2.6 mm (maxima: 14 mm) in the first half of the experiment  
248 toward average values of 1.28 mm (maxima: 5.37 mm) at the end. This decrease in peak amplitude  
249 marks the progressive mingling of the two layers, accompanied by the disruption of the high-viscous  
250 foam of Layer H as can be seen from **Figure 3a** and **Supplementary movie S01**.

### 251 252 **3.3 Effect of volume ratio of different components** 253

To investigate how the relative volumes of each layer (high viscosity Layer H and low viscosity Layer L) influence the outcomes of experiments, we performed experimental runs featuring the same average decompression rate ( $10^{-2}$  MPa/s) and viscosity contrast ( $2.1 \cdot 10^6$ ), but 1) with different volume ratios of the two layers (Exp09) and 2) doubling the volume of each layer (Exp10). Results are shown in **Figure 3**, **Figure 6a** and **Figure 8**.

No significant difference in the average velocity of the flow front of the various system configurations was detected: in **Figure 6a** all the experiments performed at 70 vol.% of particles fall within the same area. Similarly, outgassing periodicity (**Figure 8b**) does not differ significantly when accounting for different volume ratios of the Layer H and Layer L.

Conversely, a larger ratio of Layer H/Layer L (Exp07, Exp10) has been linked to a higher height of the outgassing peaks (**Figure 8a**), suggesting an efficient role of the solid loading of Layer H in promoting heterogeneous nucleation in slow decompression experiments. As a result, Exp10, featuring the maximum absolute volume of Layer H, shows the highest peak heights. This likely indicates an increasing amount of degassing volume available in this experimental run.

## 4. Discussion

### 4.1 High decompression rate: the fragmentation regime

The effect of decompression rate on gas mobility has been previously investigated both on analogue [e.g. Ichihara et al., 2002; Namiki and Manga, 2006; Kameda et al., 2008, 2013] and natural [e.g. Aldibirov and Dingwell, 2000; Kremers et al., 2010] samples. Two characteristic time-scales have been defined in literature to describe the fragmentation process. The first is the relaxation time, which quantifies the essential condition for reaching the critical stress value for the onset of solid-like brittle failure, defined as:

$$\tau_r = \frac{\eta}{G} \quad (1)$$

280 where  $\eta$  is the zero-shear viscosity and  $G$  represents the unrelaxed rigidity of the material [Barnes et  
 281 al., 1989; Dingwell and Webb, 1989]. On the basis of ultrasound measurements, Ichihara et al. [2004]  
 282 estimated a value of  $G$  for pure silicone oil (i.e. silicone oil without crystals) with a viscosity ca. 1000  
 283 Pa s in the range  $0.22\text{--}3.9 \times 10^6$  Pa. Equation (1) cannot be simply applied to particle-bearing fluids,  
 284 that are rather characterized by complex shear modulus; however, considering the case of the crystal-  
 285 free Exp02 and setting the liquid viscosity equal to 1000 Pas, the relaxation time  $\tau_r$  for Layer H ranges  
 286 from  $2 \times 10^{-4}$  to  $5 \times 10^{-3}$  s. Decompression curves in **Figure 5a** suggest that the unloading process lasted  
 287 ca.  $2\text{--}3 \times 10^{-3}$  s. As a consequence, for Exp 02, there could have been enough time for the viscous Layer  
 288 H to relax; coherently, Exp02 featured no fragmentation. The relaxation time  $\tau_r$  and the  
 289 decompression time  $t_{dec}$  can be combined in a non-dimensional parameter, the Deborah number for  
 290 decompression processes ( $De^*_{dt}$ ), i.e. the ratio between relaxation time  $\tau_r$  and decompression time  $t_{dec}$   
 291 [e.g. Kameda et al., 2013]. This “operational” Deborah number  $De^*_{dt}$  is very convenient for shock-  
 292 tube experiments where decompression time is monitored and should not be directly compared with  
 293 Deborah number computed using strain rate. Within the above-defined range for  $G$ ,  $De^*_{dt}$  for Exp02  
 294 spans the values 0.1–2.5. Kameda et al. [2013] related  $De^*_{dt}$  of ca. 1 to potential delayed, partial and  
 295 intermittent fragmentation, whereas  $De^*_{dt} \ll 1$  marked experiments where no fragmentation was  
 296 observed.

297 Addressing the relaxation time of particle- and bubble- bearing fluids requires to account for  
 298 several decompression rates [e.g. Sumita and Manga, 2008], and to consider the effect of crystals and  
 299 bubbles loading on the physical properties of the system (e.g. increasing viscosity). At particle  
 300 fractions above 40 vol.% solid-like behaviour was observed through oscillatory rheology  
 301 measurements by Namiki and Tanaka [2017], particularly at low interstitial liquid viscosity. In our  
 302 experiments, at the same decompression rate, the addition of crystals induced sample fragmentation;  
 303 the amount of crystals in Layer H is positively related to the pervasiveness of fragmentation process  
 304 (**Figure 4b-d, section 3.1**). In fact, our decompression experiments spanned from the almost  
 305 monotone single detachment observed in Exp04 (10 vol.%) to the quasi-complete disruption observed

306 in Exp08 (70 vol.%). Cordonnier et al., [2012] observed that the critical applied stress scales with  
 307 connected liquid fraction ( $1-\phi^*$ ), being  $\phi^*$  equal to the ratio between crystal fraction  $\phi$  and maximum  
 308 packing fraction  $\phi_{\max}$ . In other words, the critical stress of the liquid has to be corrected for stress  
 309 localization due to the geometrical effect of particles. The relative decrease of relaxation timescale to  
 310 trigger fragmentation with increasing solid fraction is likely to be responsible for the observed  
 311 increase in fragmentation efficiency at high solid loading. It follows that the rapid decompression of  
 312 microlite-dense shallow regions of the conduit, which originate as a consequence of crystallization  
 313 driven by decompression and cooling within the shallow conduit, might result in increased  
 314 fragmentation efficiency. Earlier and deeper fragmentation associated to the presence of crystals was  
 315 postulated also by Mourtada-Bonnefoi and Mader [2004]. In fact, Mourtada-Bonnefoi and Mader  
 316 [2004] performed instantaneous decompression experiments on bubble- or particle-bearing GRA  
 317 solutions, and found that pre-existing bubbles and heterogeneous nucleation due to the presence of  
 318 particles enhance flow dynamics significantly. Additionally, they investigated the spatial effect of  
 319 crystal distribution by using either sinking (i.e. silicon carbide, glass, mustard seeds) or floating  
 320 particles (i.e. coriander seeds) as analogue for crystals. While sinking particles show little effect on  
 321 flow dynamics, floating particles provided a high concentration of surface bubbles that lead to an  
 322 earlier fragmentation pulse.

323 However, foam acceleration prior to fragmentation has been observed in all the particle-  
 324 bearing experiments, i.e. the experiments exhibit “liquid-like” fragmentation behaviour, as expected  
 325 for low viscosity liquids [Kameda et al., 2013]. Hence, we accounted also for the second characteristic  
 326 time-scale for bubble bearing magmas, that controls viscous expansion of the bubbles [e.g. Thomas  
 327 et al., 1994, Ichihara et al., 2008]:

$$\tau_v = (1 - \psi) \frac{4\eta}{3p_0} \quad (2)$$

329 with  $p_0$  and  $\psi$  are the initial pressure and porosity, respectively. For a liquid viscosity of Layer H of  
 330 1000 Pa s, an initial pressure of 10 MPa, and investigating a range of initial porosity of 0.01-0.9 the

characteristic timescale  $\tau_v$  is on the order of  $10^{-5}$ - $10^{-4}$ s, shorter than decompression time  $\tau_{dec}$ , slightly above the range estimated for relaxation time scale  $\tau_r$ . The condition  $\tau_v < \tau_{dec}$  is in agreement with the evidence of no fragmentation in Exp02. For particle-bearing experiments, the concurrent increase in mixture viscosity is likely to increase the timescale for viscous expansion, especially for Exp08 (70 vol.%), making bubble viscous expansion a less likely primary mechanism for fragmentation.

Kameda et al. [2017] highlighted the importance of inhomogeneous bubble distribution within the sample inducing stress concentration, causing fragmentation along the surface defined by alignment of bubbles. In our experiments, in principle, we do not have relevant heterogeneities in bubble distribution, as the samples were carefully centrifuged to eliminate entrained bubbles during sample preparation, although in particle-rich experiments (e.g. Exp07, Exp08, Exp09 and Exp10) the very high viscosity of Layer H might prevent the removal of pre-existing bubbles even under centrifugal force. However, heterogeneous bubble nucleation might be responsible for inhomogeneity in bubbles distribution. Additionally, the latter changes as a function of solid loading, shifting toward connected tortuous bubble network as the percentage of solids increases [Polacci et al., 2008; Spina et al., 2016a]; the increasing gas channelling in the solid network might contribute to the pervasive disruption of Layer H observed at larger solid content (e.g. Exp08).

Lane et al. (2008) investigated the post-fragmentation dynamics of rapidly decompressed gum rosin and diethyl mixtures, and reported slow expansive ascent of the samples, accompanied by the occurrence of pressure cycles. According to the authors, bubbles pressure increases due to the confining yield strength and low permeability of the degassed flow head cap of the sample, until bubbles coalesce and collapse, forming a preferential pathway for gas escape. Pressure decrease acts then to self-seal the foam by closing the pathway, leading to the onset of a new cycle. Although there are some remarkable differences in that 1) they do not observe any physical motion of the sample surface coincident with the cycles (likely, as they claim, due to the concurrence of impulsive events at the surface); 2) the samples are characterized by a degassed plug, we believe the source of outgassing periodicity in our experiments to be similar. In fact, as suggested by Spina et al. [2016b]

for decompression experiments performed in the diluted regime for single-layer systems, the repeated foam collapse (as observed here for Layer L in fast decompression experiments and for the Layer H/Layer L system during slow decompression) is likely to be generated by the repetition of local foam collapse events (i.e. opening of gas escape pathways), facilitated by contextual bubble growth and foam shearing.

## 4.2 Slow decompression experiments: the permeability regime

Slow decompression experiments do not show fragmentation, in accordance with their very low Deborah number for decompression ( $De_{dt} < 2 \cdot 10^{-5}$ ). As detailed below, degassing occurs by permeable outgassing; the extent and style of degassing depend on the crystal loading. This is in accordance with Spina et al. [2016a], that performed slow decompression experiments on single-layer homogeneous particle-free samples with different liquid viscosities (1, 10, 100, and 1000 Pa s) and particle-bearing samples with liquid viscosity of 100 Pa s. Additionally, the authors found no significant sample expansion nor foam build-up and outgassing in particle-free samples with liquid viscosity of 10 Pa s, whereas significant sample expansion, and sample flow front height increase up to 40-50 vol.% was observed in 1000 Pa s runs [Spina et al. 2016a]. In our experimental observations of two-layer systems, clear sample expansion and foam build-up and outgassing was noticed (**Section 3.1**). This evidence suggests a primary control of the shallower portion of the sample (Layer H) – although volumetrically inferior – in the flow dynamics at the surface.

The addition of solid particles to Layer H increases gas volume ratio within the system due to heterogeneous bubble nucleation, as seen from the higher flow front expansion peaks of Exp03 and Exp05 compared to Exp01 (**Figure 7a**). Further increase in the solid loading causes the level of the mixture to descend. Similarly, the average velocity of the flow front increases due to the addition of solid particles, and decays exponentially with increasing solid content (**Figure 6a**). This is likely to occur in response to the competition between the enhanced heterogeneous bubble nucleation, driving the gas volume increase, and the compelling force exerted by the solid network with increasing solid

fraction, hindering bubble expansion, as previously noticed in homogeneous samples [Spina et al., 2016a]. Additionally, the decrease in the bulk gas content due to the diminished liquid fraction available for saturation at 70 vol.% contributed to suppression of the expansive response of the samples. These observations are consistent with those from experiments performed at high temperature on vesiculating crystal-bearing (50-80 vol.%) hydrous dacite by Pistone et al. [2017]. These authors found that at crystal contents below 60 vol.%, bubble coalescence generates permeability within shear bands; at crystal content of 60-70 vol.% bubbles might become trapped within the solid network and gas expansion is limited, whereas above 80 vol.% large overpressurization triggers brittle fragmentation of the sample. Similarly, we observed limited gas expansion, linked to bubble trapping/coalescence leading to an overall decrease of the gas volume ratio in experiments Exp07, Exp09, and Exp10; on the contrary, lower crystal contents (Exp03 and Exp05) are characterized by larger gas/volume ratio. The positive effect of crystal confinement in favouring gas migration along channels and gas fingering (i.e. bubbles forced into more viscous fluids tend to form finger-like branches) has been proved by numerical and experimental studies [Oppenheimer et al., 2015; Parmigiani et al. 2017]. Parmigiani et al., [2017] found that the favourable crystallinity window for crystal confinement and limited capillary resistance allowing optimal volatile migration is in the range 40-70 vol.%. Above this threshold, an increase of pore pressure by compaction, development of bubble overpressure and capillary fracturing are to be expected [Holtzman et al., 2012; Parmigiani et al., 2017]. Therefore, both our study and previous investigations point to the evidence that the extent of crystallization (hence the growth rate) is tuning the eruptive style by determining optimal windows for gas confinement/migration. Therefore, variation in the crystal growth rate at the shallow portion of the magmatic column likely results in relevant variation of gas permeability. Enhanced gas loss, for instance, may either reduce the potential for explosive eruptions or further promote the formation of dense plugs in the conduit, increasing the probability of Vulcanian eruptions [Lindoo et al., 2017].



409 Despite a slight tendency of outgassing periodicity to gather in a narrow range at larger crystal  
410 volume fraction, no marked effects of crystal content in Layer H on outgassing timescales has been  
411 detected in our experiments, suggesting that crystal content is not the primary factor in controlling  
412 the frequency of outgassing.

413 Slowly decompressed crystal-rich samples (Exp07, Exp09, and Exp10, 70 vol.%) experienced  
414 effective mingling between Layer H and Layer L. Indeed, before being entrained in the crystal-rich  
415 layer, bubbles slow down at the interface, and are occasionally trapped for few seconds  
416 (**Supplementary movie S07, S09, S10**). When the buoyant bubbles are finally entrained in Layer H,  
417 displacement, due to bubble ascent, causes some filaments from Layer H to be dripped down, likely  
418 in response to the onset of Rayleigh-Taylor instability [Rayleigh, 1900; Taylor, 1950]. Rayleigh-  
419 Taylor instability is an almost ubiquitous process in nature that originates when a density gradient  
420 due to gravitational force opposite to the vertical pressure gradient occurs. Descending plumes are  
421 additionally folded by vortex generated by floating bubbles and stretched by the eddies, due to local  
422 Kelvin-Helmholtz instabilities [Kelvin, 1871; Helmholtz, 1868] associated to the relative movement  
423 between the upward-moving light gas bubbles from Layer L and the heavier crystal-rich Layer H. By  
424 the end of the experiment, several plumes of Layer H had already reached the bottom of the sample  
425 and crystals originally absent from Layer L are now distributed throughout the experimental system.  
426 Notably, also the Saffman-Taylor instability [Saffman and Taylor, 1958], i.e. the formation of  
427 morphological patterns at the unstable interface among two fluids due to the displacement of the more  
428 viscous fluid (here Layer H) by the less viscous one (here Layer L), can be invoked to explain the  
429 observed plumes. Saffman-Taylor instability has been observed in natural samples and experiments  
430 with a notable morphological variability and complexity that depends on the viscosity ratio between  
431 the two fluids [e.g. Perugini and Poli, 2005].

432 Conversely, in fast decompression experiments, the fragile response of Layer H determined a  
433 rheological decoupling that prevented effective mingling between the two phases (**Figure 4 and**  
434 **Supplementary movie S04, S06, S08**). As clearly visible in Exp08, after the pervasive fragmentation

435 of Layer H, a relevant number of fragments from Layer H are entrained in the ascending Layer L as  
436 isolated crystal-rich blobs (**Supplementary movie S08**).

437 The crystal-rich experiments Exp07, Exp09, and Exp10 show analogies with lava dome periodicity.  
438 In fact, while experiments performed with lower crystal content are dominated by the dynamics of  
439 the foam, at higher crystal content the behaviour of the Layer H is more likely attributable to stick-  
440 slip behaviour, as previously noticed for homogeneous mixture with more than 40 vol.% in Spina et  
441 al. [2016a]. A stick–slip mechanism was invoked also for cyclic behaviour observed during extrusive  
442 flow from silicic volcanoes [e.g. Costa et al., 2012]. Moreover, experiments with very high crystal  
443 content highlighted the relevance of bubble motion in promoting mingling between the two layers.  
444 The relevance of bubble advection in magma mixing for systems with viscosity contrast up to 4000  
445 has also been demonstrated from experimental studies performed on basaltic-rhyolite systems  
446 [Wiesmaier et al., 2015]. Furthermore, Del Bello et al. [2015] experimentally observed that gas  
447 passing through stratified magma might promote mingling. At shallow level, exsolved volatiles are  
448 expected to interact with a partially solid plug, likely promoting mingling among the least evolved  
449 and the most crystallized layer. Here we suggest that very high decompression rates, leading to  
450 mechanical decoupling between the top layer and the crystal-free fresh magma as observed in Exp08,  
451 are more likely to produce enclave morphologies in the volcanic products.

452

## 453 **5. Implication for natural systems**

454

455 The presence of a rheological stratification within the volcanic conduit has been broadly  
456 documented in literature [e.g. Denlinger and Hoblitt, 1999; Lautze and Houghton, 2006; Lensky et  
457 al., 2008]. The intense processes capable of severely influencing the physical properties of the magma,  
458 such as gas-exsolution and crystallization, occur routinely at shallow depth. The extent of  
459 crystallization (i.e. crystal nucleation and growth rates) depends on several factors such as the degree  
460 of undercooling and the viscosity of the nucleating medium [e.g. Gibb, 1974; Vetere et al., 2013].

461 We investigated a wide range of viscosity contrasts ( $10^2$ - $10^6$ ) to encompass different degrees of  
462 crystallization of the shallow magmatic column. An increase in the absolute viscosity (liquid +  
463 crystals) of up to two orders of magnitude has been measured by Kolzenburg et al. [2016] upon  
464 cooling and crystallization of a primitive basaltic melt (from 1250 to 1150 °C), at a cooling rate of 3  
465 K/min, similar to that expected in shallow dykes. Capponi et al. [2016] provided a viscosity contrast  
466 for Stromboli volcano between the degassed and crystallized plug and the underlying fresh magma  
467 in the range 33-3300. Lava dome eruptions are characterized by a viscosity contrast between the  
468 upper conduit and the underlying magma on the order of  $10$ - $10^2$  [Barmin et al. 2002]. Similarly, in  
469 our study, experiments Exp01 to Exp06 have viscosity contrast between Layer H and Layer L on the  
470 order of  $10^2$ . Higher viscosity contrasts (up to more than  $10^6$ ) have been reported in literature in  
471 various volcanological settings such as lava flows and explosive eruptions [e.g. Freund and Tait,  
472 1986; Vetere et al., 2015; Morgavi et al., 2016].

473 The experiments performed in this work mimic the following decompression history of a  
474 volcanic system, characterized by two main stages: (i) time zero, representing the condition where  
475 two magmas with different physical properties are superimposed within the conduit; such vertical  
476 heterogeneity can result from a sharp gradient in the physical properties due to crystallization upon  
477 cooling of the shallowest section of the magmatic column or the progressive ascent of different  
478 batches of magmas in time; (ii) time  $t_i$  outlines the eruptive phase: the system is decompressed or  
479 driven to surface by conduit replenishment; as a consequence, bubble nucleation and/or fragmentation  
480 occurs.

481 In volcanic systems, decompression-induced nucleation might occur both as a consequence  
482 of an external trigger, as for instance unroofing due to flank collapse [e.g. Belousov et al., 2007;  
483 Manconi et al., 2009], or resulting from the unloading of the magmatic column by static  
484 decompression [e.g. Poland et al., 2009] or by lithostatic release following the removal of temporal  
485 barrier [e.g. Johnson, 1998; Johnson and Lees, 2000; Massaro et al., 2018] or finally, be driven by  
486 magma ascent following conduit replenishment. According to Aldibirov and Dingwell [2000],

487 different decompression rates from less than one second up to tens of seconds are expected for cap  
488 rock cracking and gravitational driven landslides, respectively. In our fast decompression  
489 experiments, we reproduced an average decompression rate of  $10^4$  MPa/s, with a total pressure drop  
490 of 10 MPa. Remarkably, the hydrostatic pressure drop due to the 18 May 1980 landslide at Mt. St.  
491 Helens has been estimated to be up to 20 MPa. According to Namiki and Manga [2006]  
492 decompression rates above  $2 \cdot 10^4$  Pa/s are typical of explosive eruption in volcanoes with higher  
493 decompression rates ( $>10^6$  Pa/s) being linked to fragmentation processes. More generally speaking,  
494 decompression rates in natural systems spans several log units [Cassidy et al., 2018], hence magmatic  
495 systems might be characterized by different decompression timescales. According to Deborah  
496 numbers for decompression  $De^*_{dt}$  estimated for crystal-free analogue magmas (**Sections 4.1 and 4.2**),  
497 the decompression rates explored in this work help constraining the behaviour of layered systems in  
498 the “no fragmentation” regime of Layer H (slow decompression experiments:  $De^*_{dt} \ll 1$ ) and brittle-  
499 like to brittle regime of Layer H (fast decompression experiments:  $De^*_{dt} \approx 1$  and  $De^*_{dt} \gg 1$ ) [Kameda  
500 et al., 2013]. Lower bulk strain rates are required to induce brittle transition in crystal bearing magmas  
501 [Wadsworth et al. 2017].

502 Notably, in volcanic systems, subsurface crystallization is generally accompanied by gas  
503 exsolution, culminating in gas depletion, if permeable degassing is effective. Alternatively,  
504 overpressure is likely to develop when outgassing is not allowed. For instance, with the exception of  
505 the very outermost degassed region, the shallowest region of the conduit feeding lava domes bears  
506 volatiles and crystals: rapid crystallization leads to intense diffusion of volatiles and growth of  
507 bubbles [e.g. Melnik and Sparks, 2005; Costa et al., 2013]. The high bubble volume fraction observed  
508 in our experiments in Layer H is coherent with such a process, although experiments exhibit no gas  
509 depletion or enrichment in the liquid component of Layer H before decompression.

510

## 511 **6. Conclusions**

512           Magmas with the same pre-eruptive composition and volatile content may generate different  
513 eruptive styles depending on the degassing modes during conduit ascent. To constrain the role of  
514 different parameters, i.e. physical properties and decompression rate, on the degassing styles of non-  
515 homogeneous system we performed decompression experiments on analogue samples. Our  
516 experimental data illustrate that decompression rate (hence ascent rate) plays the most influential role  
517 on eruptive style, controlling outgassing periodicity and eruptive history. In fact, changing different  
518 experimental parameters of similar order of magnitudes for example viscosity ratio and  
519 decompression rate the second one shows the higher effect on the eruptive style. Decompression rate,  
520 modified over 6 orders of magnitude, showed the most dramatic effect on eruptive style compared to  
521 other experimental variables such as viscosity contrast at the interface, that was modified in a similar  
522 range (4 orders of magnitude). Thus, for a given volcanic system, very different scenarios can be  
523 envisaged when accounting for possible external triggers (such as un-roofing due to flank collapse)  
524 that can lead to a significant variation of the decompression rate. In fast decompression experiments,  
525 the solid fraction affects the spatial pervasiveness of fragmentation of Layer H, ranging from a  
526 complete absence of fragmentation (0 vol.%) to complete disruption (70 vol.%) at the same  
527 decompression rate, directly influencing the fragmentation efficiency of the upper layer. In slow  
528 decompression experiments, crystal content tunes gas volume ratio and determines the optimal  
529 window for gas confinement/migration: this evidence suggests that rate of crystal growth at the  
530 shallow portion of the magmatic column severely affects permeability; and that the control exerted  
531 by crystallinity on gas loss has direct implications for hazard assessment.

532

### 533           *Acknowledgments*

534           *The authors wish to thank the Editor, T. Mather, and A. Namiki and an anonymous reviewer*  
535 *for their constructive comments that helped to improve the clarity of the manuscript. L.S., D.M., D.P.*  
536 *wish to thank the ERC Consolidator “CHRONOS” project (Grant No. 612776). D.M. and L.S. are*  
537 *grateful to the project MORGABASE for partial support. A.C. acknowledges the support of a visit to*

538 *LMU via the Research Group “Magma to Tephra: Ash in the Earth System” (DBD). L.S. and D.M.*  
539 *thanks B. Morgavi for constant support during the writing of the manuscript. The authors also thank*  
540 *E. Carocci for support during preliminary investigations.*

*Figure captions*

**Figure 1:** a) Sketch of the experimental apparatus: the sample is inserted in the High Pressure Autoclave. A manometer, connected to an Argon tank, provides gas for sample saturation. Decompression is performed by blowing the set of diaphragms at the top of the High Pressure Autoclave (average decompression rate  $10^4$  MPa/s) or by opening a dedicated valve located in the manometer and evacuating the system (average decompression rate  $10^{-2}$  MPa). An oscilloscope allows for the collection of data from the pressure sensor; b) details on the high pressure autoclave and sample configuration.

**Figure 2:** Flow front expansion curves (blue solid line, left side y-axis) and decompression curves (red solid line, right side y-axis) of (a) Exp06 (average decompression rate  $10^4$  MPa/s) and (b) Exp05 (average decompression rate  $10^{-2}$  MPa/s). Flow front expansion curves were measured by picking the position of the flow front at its centre using high-speed-videos (sampling rate of 5000 and 50 fps for a) and b), respectively). Red circles represent detections of local maxima corresponding to outgassing episodes. Green dotted lines mark the first expansion phase of the sample.

**Figure 3:** Snapshots of slow decompression experiments, with increasing solid content in Layer H from the top to the bottom of the picture corresponding to (a) 0 vol.% (Exp01); (b) 10 vol.% (Exp03); (c) 30 vol.% (Exp05); (d-f) 70 vol.% (Exp07, Exp09, Exp10). In (a-d) the volume ratio between Layer H and Layer L is equal to 0.6, whereas in (e) it corresponds to 0.3 and in (f) Exp10 features the same volume ratio as in a-d, but the double absolute volume for each component. Snapshots I to X were taken each 200 s from video trigger (ca. 70s from decompression, at a pressure of 6 MPa). Spatial scale is provided by the diameter of the experimental conduit equal to 2 cm. The yellow line in frames IV and V in (b) marks the top of the sample holder; please note the vertical shift in the relative position between the camera and the autoclave in frames V-X compared to frames I-IV. In (e) frames IX and X are missing, due to a slightly shorter duration of the movie compared to the other experiments.

567 **Figure 4:** Snapshots of fast decompression experiments, with increasing solid content in  
568 Layer H from the top to the bottom of the picture corresponding to (a) 0 vol% (Exp02); (b) 10 vol%  
569 (Exp04); (c) 30 vol% (Exp06); (d-f) 70 vol% (Exp08). Snapshots were taken before decompression  
570 (I), at time zero (II) and at 0.2 (III), 0.4 (IV), 2 (V), 4 (VI), 6 (VII), 8 (VIII) and 10 seconds (IX) after  
571 decompression. Spatial scale is provided by the diameter of the experimental conduit.

572 **Figure 5:** Details of fragmentation process in particle-bearing experiments. a) Pressure curves  
573 for fast decompression experiments without (Exp02) and with (Exp04-Exp06-Exp08) particles. b, c,  
574 d) Snapshots of Exp04 (10vol.%), Exp06 (30vol.%) and Exp08 (70vol.%), respectively. Spatial scale  
575 is provided by the diameter of the experimental conduit, which is equal to 2 cm.

576 **Figure 6:** Initial average velocity of the flow front expanding mixture measured from high-  
577 speed-videos for slow (a) and fast (b) decompression experiments at different viscosity ratios (i.e.  
578 different solid loading in Layer H). The error in the measurement is of the same order of the symbol  
579 size.

580 **Figure 7:** (a) Height of detected peaks in the flow front expansion curves during outgassing  
581 episodes plotted against viscosity ratio; (b) elapsed times between two consecutive outgassing  
582 episodes, defined as local maxima in the flow front expansion curve (**Section 2.2**). In (a) and (b) slow  
583 decompression experiments are marked by full dots, whereas fast decompressions are indicated by  
584 crosses. Pink and azure chart areas mark experiments performed with pure silicone oils and crystal-  
585 bearing samples, respectively.

586 **Figure 8:** (a) Height of detected peaks in the flow front expansion curves during outgassing  
587 episodes and (b) elapsed time between two consecutive outgassing episodes versus volume ratio of  
588 Layer H to Layer L of the investigated system configuration.

589

590 *Table captions*

591 **Table 1:** Summary of experimental conditions.

592



*Supplementary files captions*

**Supplementary movie S01:** Accelerated movie (32x) of Exp01, featuring 10 Pa s silicone oil bearing a small amount (0.025 g/cm<sup>3</sup>) of yellow food colorant (E102, E172) in Layer L, 1000 Pa s silicone oil in Layer H. Average decompression rate is equivalent to ca. 10<sup>-2</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S02:** Decelerated movie (0.19x) of Exp02 featuring 10 Pa s silicone oil bearing a small amount (0.025 g/cm<sup>3</sup>) of yellow food colorant (E102, E172) in Layer L, 1000 Pa s silicone oil in Layer H. Average decompression rate is equivalent to ca. 10<sup>4</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S03:** Accelerated movie (32x) of Exp03, featuring 10 Pa s silicone oil in Layer L, 1000 Pa s silicone oil bearing 10 vol.% of crystals in Layer H. Average decompression rate is equivalent to ca. 10<sup>-2</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S04:** Decelerated movie (0.19x) of Exp04 featuring 10 Pa s silicone oil in Layer L, 1000 Pa s silicone oil bearing 10 vol.% of crystals in Layer H. Average decompression rate is equivalent to ca. 10<sup>4</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S05:** Accelerated movie (32x) of Exp05, featuring 10 Pa s silicone oil in Layer L, 1000 Pa s silicone oil bearing 30 vol.% of crystals in Layer H. Average decompression rate is equivalent to ca. 10<sup>-2</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S06:** Decelerated movie (0.19x) of Exp06 featuring 10 Pa s silicone oil in Layer L, 1000 Pa s silicone oil bearing 30 vol.% of crystals in Layer H. Average decompression rate is equivalent to ca. 10<sup>4</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

**Supplementary movie S07:** Accelerated movie (32x) of Exp07, featuring 10 Pa s silicone oil in Layer L, 1000 Pa s silicone oil bearing 70 vol.% of crystals in Layer H. Average decompression rate is equivalent to ca. 10<sup>-2</sup> MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

619           **Supplementary movie S08:** Decelerated movie (0.19x) of Exp08 featuring 10 Pa s silicone  
620 oil in Layer L, 1000 Pa s silicone oil bearing 70 vol.% of crystals in Layer H. Average decompression  
621 rate is equivalent to ca.  $10^4$  MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6.

622           **Supplementary movie S09:** Accelerated movie (32x) of Exp09, featuring 10 Pa s silicone oil  
623 in Layer L, 1000 Pa s silicone oil bearing 70 vol.% of crystals in Layer H. Average decompression  
624 rate is equivalent to ca.  $10^{-2}$  MPa/s. The volume ratio between Layer H and Layer L is equal to 0.3.

625           **Supplementary movie S10:** Accelerated movie (32x) of Exp10, featuring 10 Pa s silicone oil  
626 in Layer L, 1000 Pa s silicone oil bearing 70 vol.% of crystals in Layer H. Average decompression  
627 rate is equivalent to ca.  $10^{-2}$  MPa/s. The volume ratio between Layer H and Layer L is equal to 0.6;  
628 the absolute volume of each phase is doubled than Exp01-08.

629

630

631

632

633

634

635

636

## 637           **References**

638           Aldibirov, M., & Dingwell, D. B. (2000). Three fragmentation mechanisms for highly viscous  
639 magma under rapid decompression. *Journal of Volcanology and Geothermal Research*, 100(1-4),  
640 413-421.

641           Barnes, H.A., Hutton, J.F., Walters, K., (1989). *An Introduction to Rheology*. Elsevier

642           Barmin, A., Melnik, O., & Sparks, R. S. J. (2002). Periodic behavior in lava dome eruptions.  
643 *Earth and Planetary Science Letters*, 199(1-2), 173-184.

644 Belousov, A., Voight, B., & Belousova, M. (2007). Directed blasts and blast-generated  
645 pyroclastic density currents: a comparison of the Bezymianny 1956, Mount St Helens 1980, and  
646 Soufrière Hills, Montserrat 1997 eruptions and deposits. *Bulletin of Volcanology*, 69(7), 701.

647 Capponi, A., James, M. R., & Lane, S. J. (2016). Gas slug ascent in a stratified magma:  
648 Implications of flow organisation and instability for Strombolian eruption dynamics. *Earth and*  
649 *Planetary Science Letters*, 435, 159-170.

650 Cashman, K. V., Sparks, R. S. J., & Hawkesworth, C. J. (2004). Volatile controls on magma  
651 ascent and eruption. *The State of the Planet: Frontiers and Challenges in Geophysics*, 150, 109-124.

652 Cassidy, M., Manga, M., Cashman, K., & Bachmann, O. (2018). Controls on explosive-  
653 effusive volcanic eruption styles. *Nature communications*, 9(1), 2839.

654 Cimorelli, C., Costa, A., Mueller, S., & Mader, H. M. (2011). Rheology of magmas with  
655 bimodal crystal size and shape distributions: Insights from analog experiments. *Geochemistry,*  
656 *Geophysics, Geosystems*, 12(7).

657 Cordonnier, B., Caricchi, L., Pistone, M., Castro, J., Hess, K.-U., Gottschaller, S., Manga, M.,  
658 Dingwell, D.B., Burlini, L., (2012). The viscous-brittle transition of crystal-bearing silicic melt: direct  
659 observation of magma rupture and healing. *Geology* 40, 611–614.

660 Costa A. (2005) Viscosity of high crystal content melts: dependence on solid fraction,  
661 *Geophys. Res. Lett.*, 32, L22308, doi: 10.1029/2005GL024303

662 Costa, A., & Macedonio, G. (2002) Nonlinear phenomena in fluids with temperature-  
663 dependent viscosity: an hysteresis model for magma flow in conduits, *Geophysical Research Letter*,  
664 29, 1402, doi:10.1029/2001GL014493

665 Costa, A., Caricchi, L., & Bagdassarov, N. (2009). A model for the rheology of particle -  
666 bearing suspensions and partially molten rocks. *Geochemistry, Geophysics, Geosystems*, 10(3).

667 Costa, A., Wadge, G., & Melnik, O. (2012). Cyclic extrusion of a lava dome based on a stick-  
668 slip mechanism. *Earth and Planetary Science Letters*, 337, 39-46.

669 Costa, A., Wadge, G., Stewart, R., & Odbert, H. (2013). Coupled subdaily and multiweek  
670 cycles during the lava dome eruption of Soufrière Hills Volcano, Montserrat. *Journal of Geophysical*  
671 *Research: Solid Earth*, 118(5), 1895-1903.

672 Del Bello, E., Lane, S. J., James, M. R., Llewellyn, E. W., Taddeucci, J., Scarlato, P., &  
673 Capponi, A. (2015). Viscous plugging can enhance and modulate explosivity of strombolian  
674 eruptions. *Earth and Planetary Science Letters*, 423, 210-218.

675 Denlinger, R. P., & Hoblitt, R. P. (1999). Cyclic eruptive behavior of silicic volcanoes.  
676 *Geology*, 27(5), 459-462.

677 Dingwell, D. B., & Webb, S. L. (1989). Structural relaxation in silicate melts and non-  
678 Newtonian melt rheology in geologic processes. *Physics and Chemistry of Minerals*, 16(5), 508-516.

679 Freundt, A., & Tait, S. R. (1986). The entrainment of high-viscosity magma into low-viscosity  
680 magma in eruption conduits. *Bulletin of Volcanology*, 48(6), 325-339.

681 Gibb F. G. (1974) Supercooling and the crystallization of plagioclase from a basaltic magma.  
682 *Mineral. Mag.* 39(306), 641–653

683 Helmholtz, Hermann von- (1868). "Über discontinuierliche Flüssigkeits-Bewegungen [On the  
684 discontinuous movements of fluids]". *Monatsberichte der Königlichen Preussische Akademie der*  
685 *Wissenschaften zu Berlin*. 23: 215–228

686 Holtzman, R., M. L. Szulczewski, & R. Juanes (2012), Capillary fracturing in granular media,  
687 *Phys. Rev. Lett.*, 108(26), 264504.

688 Ichihara, M., Rittel, D., & Sturtevant, B. (2002). Fragmentation of a porous viscoelastic  
689 material: implications to magma fragmentation. *Journal of Geophysical Research: Solid Earth*,  
690 107(B10).

691 Ichihara, M., Ohkunitani, H., Ida, Y., & Kameda, M. (2004). Dynamics of bubble oscillation  
692 and wave propagation in viscoelastic liquids. *Journal of Volcanology and Geothermal Research*,  
693 129(1-3), 37-60.

694 Ichihara, M. (2008). Dynamics of a spherical viscoelastic shell: implications to a criterion for  
695 fragmentation/expansion of bubbly magma. *Earth and Planetary Science Letters* 265, 18–32.

696 Johnson, J. B., Lees, J. M., & Gordeev, E. I. (1998). Degassing explosions at Karymsky  
697 volcano, Kamchatka. *Geophysical Research Letters*, 25(21), 3999-4002.

698 Johnson, J. B., & Lees, J. M. (2000). Plugs and chugs—seismic and acoustic observations of  
699 degassing explosions at Karymsky, Russia and Sangay, Ecuador. *Journal of Volcanology and*  
700 *Geothermal Research*, 101(1-2), 67-82.

701 Kameda, M., Kuribara, H., & Ichihara, M. (2008). Dominant time scale for brittle  
702 fragmentation of vesicular magma by decompression. *Geophysical Research Letters*, 35(14).

703 Kameda, M., Ichihara, M., Shimanuki, S., Okabe, W., & Shida, T. (2013). Delayed brittle-  
704 like fragmentation of vesicular magma analogue by decompression. *Journal of Volcanology and*  
705 *Geothermal Research*, 258, 113-125.

706 Kameda, M., Ichihara, M., Maruyama, S., Kurokawa, N., Aoki, Y., Okumura, S., & Uesugi,  
707 K. (2017). Advancement of magma fragmentation by inhomogeneous bubble distribution. *Scientific*  
708 *reports*, 7(1), 16755.

709 Kelvin, L. (William Thomson) (1871). "Hydrokinetic solutions and observations".  
710 *Philosophical Magazine*. 42: 362–377.

711 Kolzenburg, S., Giordano, D., Cimorelli, C., & Dingwell, D. B. (2016). In situ thermal  
712 characterization of cooling/crystallizing lavas during rheology measurements and implications for  
713 lava flow emplacement. *Geochimica et Cosmochimica Acta*, 195, 244-258.

714 Kremers, S., Scheu, B., Cordonnier, B., Spieler, O., & Dingwell, D. B. (2010). Influence of  
715 decompression rate on fragmentation processes: An experimental study. *Journal of Volcanology and*  
716 *Geothermal Research*, 193(3-4), 182-188.

717 Lane, S.J., Phillips, J.C. & Ryan, G.A. (2008). Dome-building eruptions: insights from  
718 analogue experiments. *Geological Society, London, special Publications* 307.1 (2008): 207-237.

719           Lautze, N.C. & Houghton, B.F., (2005). Physical mingling of magma and complex eruption  
720 dynamics in the shallow conduit at Stromboli volcano, Italy. *Geology*, 33 (5), 425.  
721 <http://dx.doi.org/10.1130/G21325.1>.

722           Lautze, N.C. & Houghton, B.F., (2006). Linking variable explosion style and magma textures  
723 during 2002 at Stromboli volcano, Italy. *Bulletin of Volcanology*, 69 (4), 445–460.  
724 <http://dx.doi.org/10.1007/s00445-006-0086-1>.

725           Lensky, N.G., Sparks, R.S.J., Navon, O., & Lyakhovsky, V. (2008). Cyclic activity at  
726 Soufriere Hills Volcano, Montserrat: Degassing-induced pressurization and stick-slip extrusion. In:  
727 Lane, S.J., Gilbert, J.S. (Eds.), *Fluid Motions in Volcanic Conduits: A Source of Seismic and*  
728 *Acoustic Signals*, vol. 307. Geological Society Special Publication, pp. 169–188.

729           Lindoo, A., Larsen, J. F., Cashman, K. V., & Oppenheimer, J. (2017). Crystal controls on  
730 permeability development and degassing in basaltic andesite magma. *Geology*, 45(9), 831-834.

731           Manconi, A., Longpré, M. A., Walter, T. R., Troll, V. R., & Hansteen, T. H. (2009). The  
732 effects of flank collapses on volcano plumbing systems. *Geology*, 37(12), 1099-1102.

733           Massaro S., Sulpizio R., Costa A., Capra L., & Lucchi F. (2018) On eruptive style transition  
734 at calc-alkaline volcanoes: the example of the 1913 eruption of Fuego de Colima volcano (Mexico),  
735 *Bulletin of Volcanology*, 80 (7), doi:10.1007/s00445-018-1235-z

736           Melnik, O., & Sparks, R.S.J., (1999). Nonlinear dynamics of lava dome extrusion. *Nature* 402,  
737 37–41.

738           Melnik, O., & Sparks, R. S. J., (2005). Controls on conduit magma flow dynamics during lava  
739 dome building eruptions. *J. Geophys. Res.* 110 (B022). doi:10.1029/2004JB003183.

740           Melnik, O., & Sparks, R. S. J., Costa, A., & Barmin, A. (2008). Volcanic eruptions: cyclicity  
741 during lava dome growth. In *Meyers: Encyclopedia of Complexity and Systems Science*. Springer.

742           Morgavi, D., Arzilli, F., Pritchard, C., Perugini, D., Mancini, L., Larson, P., & Dingwell, D.  
743 B. (2016). The Grizzly Lake complex (Yellowstone Volcano, USA): Mixing between basalt and  
744 rhyolite unraveled by microanalysis and X-ray microtomography. *Lithos*, 260, 457-474.

745 Mourtada-Bonnefoi, C.C., & Mader H.M. (2004). Experimental observations of the effect of  
746 crystals and pre-existing bubbles on the dynamics and fragmentation of vesiculating flows. *Journal*  
747 *of Volcanology and Geothermal Research* 129.1-3 (2004): 83-97.

748 Namiki, A., & Manga, M. (2006). Influence of decompression rate on the expansion velocity  
749 and expansion style of bubbly fluids. *Journal of Geophysical Research: Solid Earth*, 111(B11).

750 Namiki, A., & Manga, M. (2008). Transition between fragmentation and permeable  
751 outgassing of low viscosity magmas. *Journal of Volcanology and Geothermal Research*, 169(1-2),  
752 48-60.

753 Namiki, A., & Tanaka, Y. (2017). Oscillatory rheology measurements of particle - and  
754 bubble - bearing fluids: Solid - like behavior of a crystal - rich basaltic magma. *Geophysical*  
755 *Research Letters*, 44(17), 8804-8813.

756 Oppenheimer, J., Rust, A. C., Cashman, K. V., & Sandnes, B. (2015). Gas migration regimes  
757 and outgassing in particle-rich suspensions. *Flow and Transformations in Porous Media*, 96.

758 Parmigiani, A., Degruyter, W., Leclaire, S., Huber, C., & Bachmann, O. (2017). The  
759 mechanics of shallow magma reservoir outgassing. *Geochemistry, Geophysics, Geosystems*.

760 Perugini, D., & Poli, G. (2005). Viscous fingering during replenishment of felsic magma  
761 chambers by continuous inputs of mafic magmas: field evidence and fluid-mechanics experiments.  
762 *Geology*, 33(1), 5-8.

763 Pistone, M., Whittington, A. G., Andrews, B., & Cottrell, E. (2017). Crystal-rich lava dome  
764 extrusion during vesiculation: An experimental study. *Journal of Volcanology and Geothermal*  
765 *Research*.

766 Polacci, M., Baker, D. R., Bai, L., & Mancini, L. (2008). Large vesicles record pathways of  
767 degassing at basaltic volcanoes. *Bulletin of Volcanology*, 70(9), 1023-1029.

768 Polacci, M., Baker, D.R., Mancini, L., Favretto, S., & Hill, R.J., (2009). Vesiculation in  
769 magmas from Stromboli and implications for normal Strombolian activity and paroxysmal explosions

770 in basaltic systems. *Journal of Geophysical Research*, 114, B01206.  
 771 <http://dx.doi.org/10.1029/2008JB005672>.

772 Poland, M. P., Sutton, A. J., & Gerlach, T. M. (2009). Magma degassing triggered by static  
 773 decompression at Kīlauea Volcano, Hawai ‘i. *Geophysical Research Letters*, 36(16).

774 Rayleigh, J. W. S. (1900), *Scientific Papers* Cambridge University Press, Cambridge,  
 775 England,), Vol. II, p. 200.

776 Saffman, P. G., & Taylor, G. I. (1958). The penetration of a fluid into a porous medium or  
 777 Hele-Shaw cell containing a more viscous liquid. *Proc. R. Soc. Lond. A*, 245(1242), 312-329.

778 Spina, L., Cimorelli, C., Scheu, B., Di Genova, D., & Dingwell, D. B. (2016a). On the slow  
 779 decompressive response of volatile-and crystal-bearing magmas: An analogue experimental  
 780 investigation. *Earth and Planetary Science Letters*, 433, 44-53.

781 Spina, L., Scheu, B., Cimorelli, C., Arciniega - Ceballos, A., & Dingwell, D. B. (2016b).  
 782 Time scales of foam stability in shallow conduits: Insights from analogue experiments. *Geochemistry,*  
 783 *Geophysics, Geosystems*, 17(10), 4179-4194.

784 Sumita, I., & M. Manga (2008), Suspension rheology under oscillatory shear and its  
 785 geophysical implications, *Earth Planet. Sci. Lett.*, 269, 467–476.

786 Thomas, N., Jaupart, C. & Vergnolle, S. (1994). On the vesicularity of pumice. *Journal of*  
 787 *Geophysical Research* B 99, 15,633–15,644

788 Taddeucci, J., Spieler, O., Ichihara, M., Dingwell, D. B., & Scarlato, P. (2006). Flow and  
 789 fracturing of viscoelastic media under diffusion-driven bubble growth: An analogue experiment for  
 790 eruptive volcanic conduits. *Earth and Planetary Science Letters*, 243(3), 771-785.

791 Taylor, G I. (1950), *Proc. R. Sot. London Ser*, AZO1, 192.

792 Vetere F., Iezzi G., Behrens H., Cavallo A., Misiti V., Dietrich M., Knipping J., Ventura G.  
 793 & Mollo S. (2013) Intrinsic solidification behaviour of basaltic to rhyolitic melts: a cooling rate  
 794 experimental study. *Chemical Geology* 354, 233–242.



795            Vetere, F., Petrelli, M., Morgavi, D., & Perugini, D. (2015). Dynamics and time evolution of  
796    a shallow plumbing system: The 1739 and 1888–90 eruptions, Vulcano Island, Italy. *Journal of*  
797    *Volcanology and Geothermal Research*, 306, 74–82.

798            Wadsworth, F. B., Witcher, T., Vasseur, J., Dingwell, D. B., & Scheu, B. (2017). When does  
799    magma break?. In *Volcanic Unrest* (pp. 171-184). Springer, Cham.

800            Wiesmaier, S., Morgavi, D., Renggli, C. J., Perugini, D., De Campos, C. P., Hess, K. U., &  
801    Dingwell, D. B. (2015). Magma mixing enhanced by bubble segregation. *Solid Earth*, 6(3), 1007.

